Supporting Information

Controlling the Antimicrobial Action of Surface Modified Magnesium Hydroxide Nanoparticles

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1. A schematic of the synthesis method of Mg(OH)₂NPs

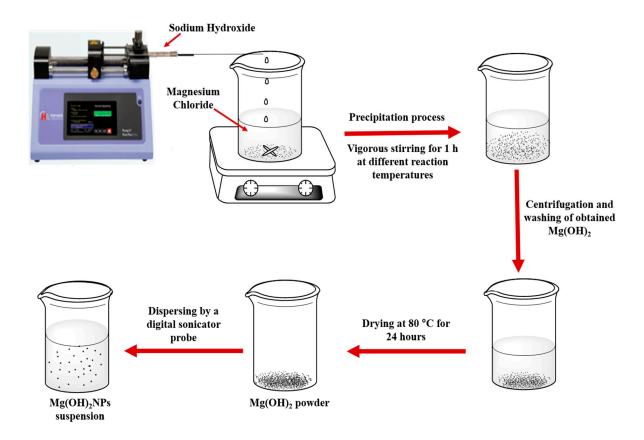
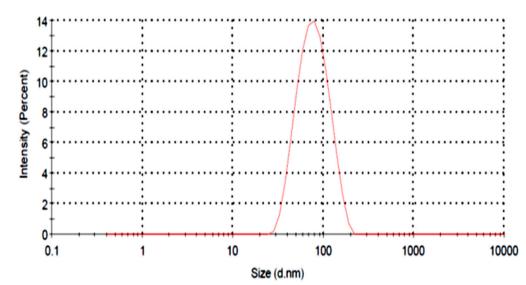


Figure S1. A schematic overview summarizing the synthesis method of Mg(OH)₂NPs.



2. The particles size and zeta potential of Mg(OH)₂NPs

Figure S2. Particles size of Mg(OH)₂NPs made from a magnesium chloride at 75°C.

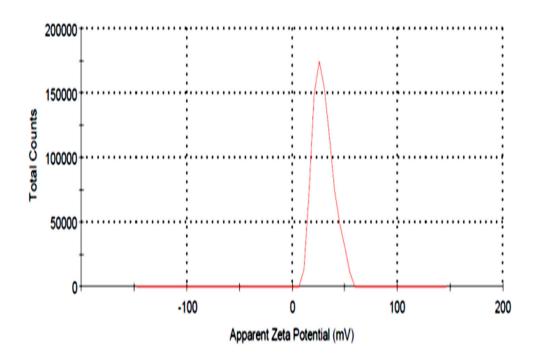


Figure S3. The zeta potential of Mg(OH)₂NPs made from a magnesium chloride at 75°C.

3. Preparation and Characterization of Mg(OH)₂NPs

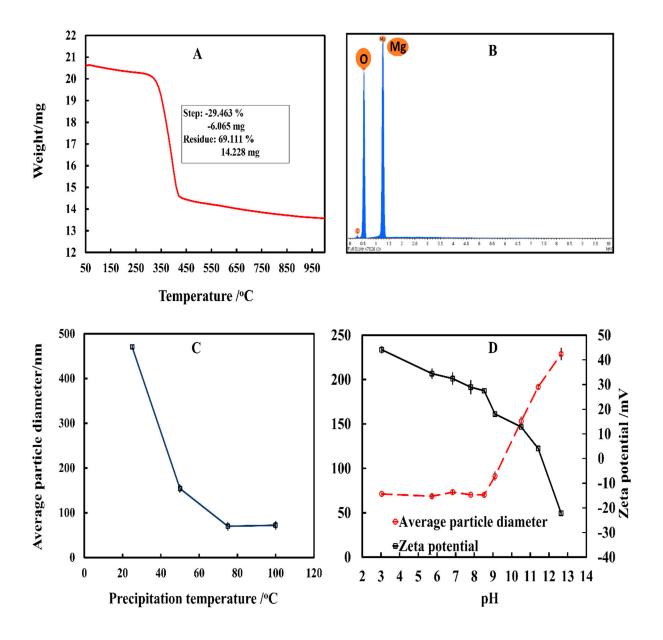
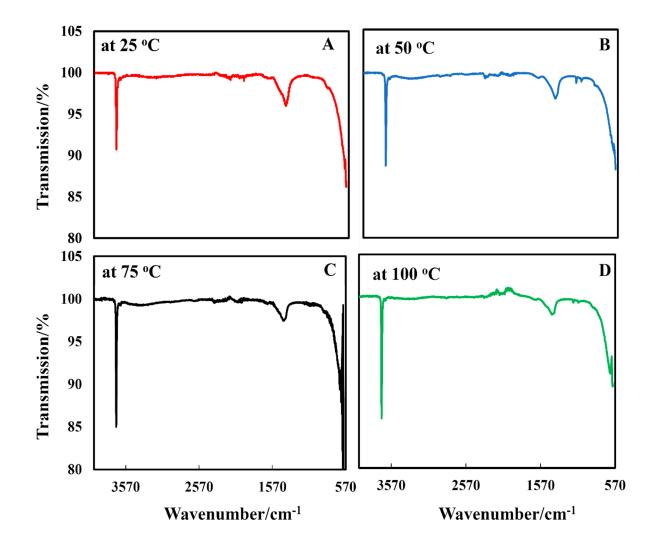


Figure S4. (A) Thermal gravimetric analysis pattern of Mg(OH)₂NPs powder. (B) The EDX spectra of the uncoated Mg(OH)₂NPs. (C) The impact of reaction temperature on the size of the produced Mg(OH)₂NPs. (D) Variations in particle size and zeta potential of Mg(OH)₂NPs suspensions with pH.



4. Fourier Transform Infrared Spectroscopy (FTIR) of Mg(OH)₂NPs

Figure S5. FTIR spectra of the as prepared Mg(OH)₂NPs at different reaction temperatures; (A) 25 °C, (B) 50 °C,(C) 75 °C and (D) 100 °C.

5. EDX diagram of S. cerevisiae cells with Mg(OH)₂NPs

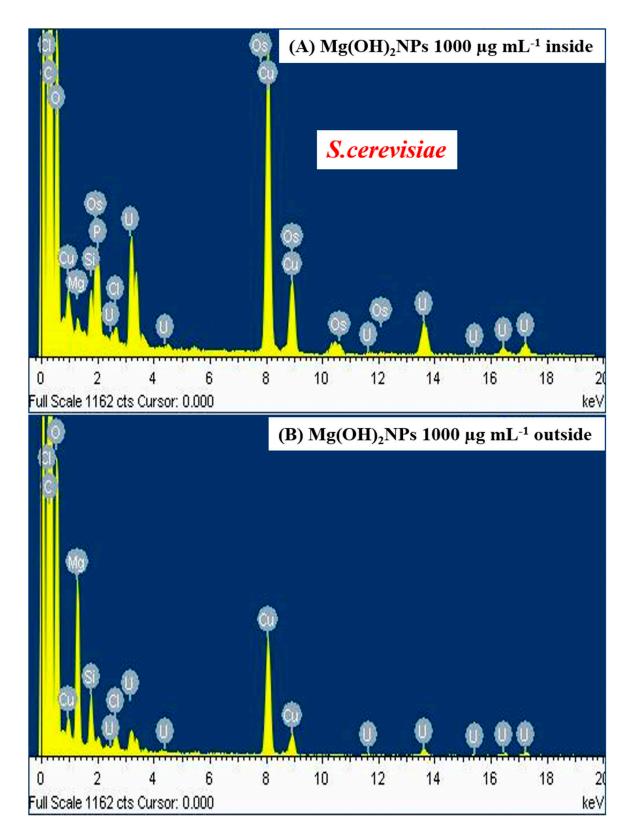
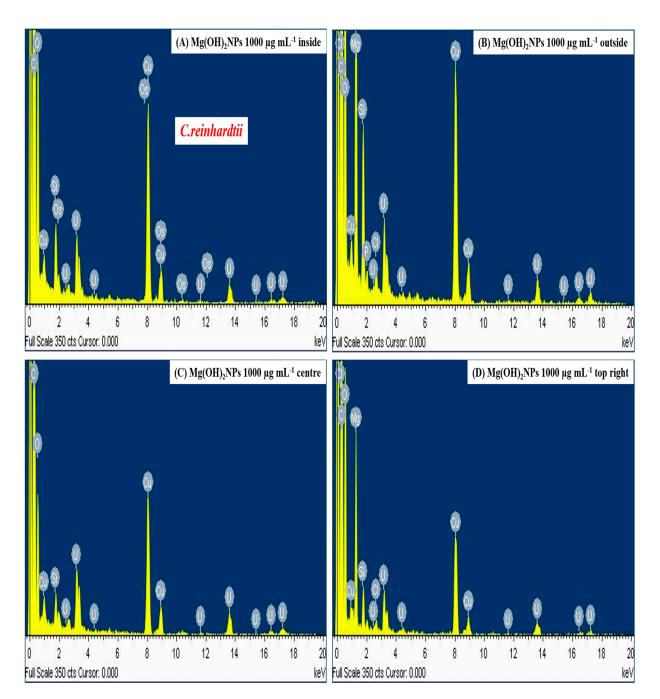


Figure S6. EDX diagram of *S.cerevisiae* cells at 1000 μ g mL⁻¹: (A) the inside membrane of *S.cerevisiae* and (B) the outside membrane of *S.cerevisiae*. The data indicate the existence of Mg(OH)₂NPs on the outside part of the cell membrane.



6. EDX chart of the *C. reinhardtii* with Mg(OH)₂NPs

Figure S7. EDX chart of the *C. reinhardtii* with Mg(OH)₂NPs at 1000 μ g mL⁻¹: (A) inside membrane of *C. reinhardtii* and (B) outside membrane of *C. reinhardtii*; (C) centre cell and (D) top right. This demonstrates the lack of internalised Mg(OH)₂NPs in *C. reinhardtii* even at NPs concentration 1000 μ g mL⁻¹.

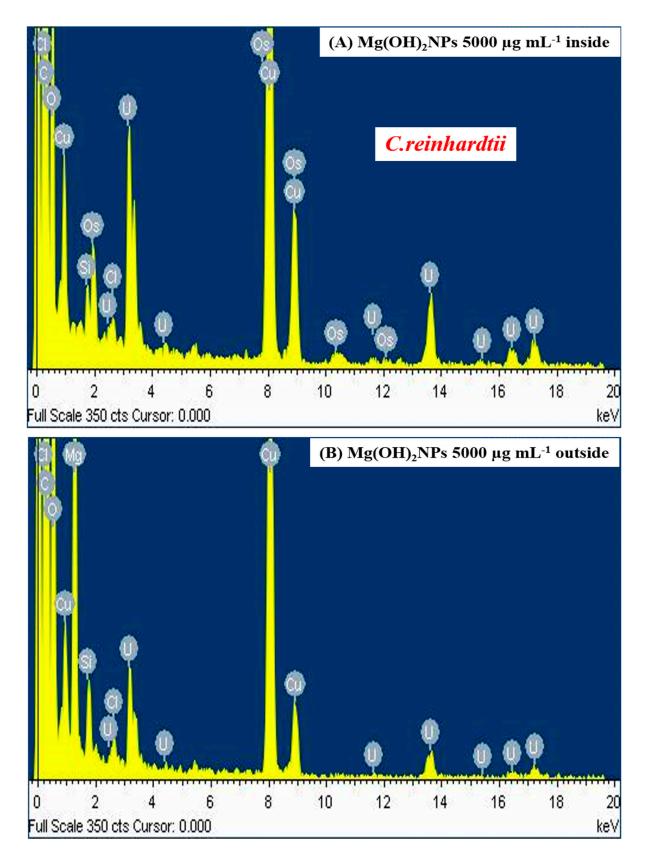
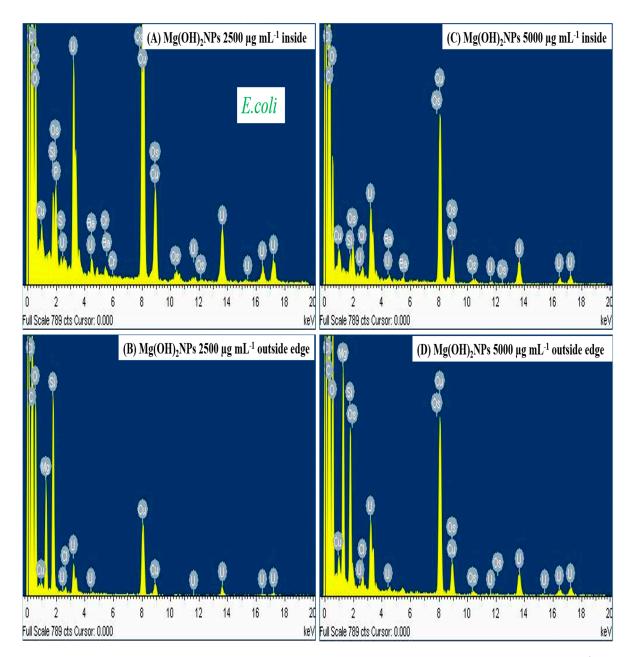


Figure S8. EDX chart of the *C. reinhardtii* with $Mg(OH)_2NPs$ at 5000 µg mL⁻¹: (A) inside the membrane of the *C. reinhardtii* and (B) outer the membrane of the *C. reinhardtii*.



7. EDX chart of the *E. coli* with Mg(OH)₂NPs

Figure S9. EDX diagram of *E. coli* cells incubated with Mg(OH)₂NPs at 2500 μ g mL⁻¹ and 5000 μ g mL⁻¹: (A) *E. coli* inside wall and (B) *E. coli* outer wall areas. (C) *E. coli* inside wall and (D) *E. coli* outer wall areas. The data indicate the existence of Mg(OH)₂NPs on the external part of the cell membrane.

8. Schematic representation of the bacterial cell wall

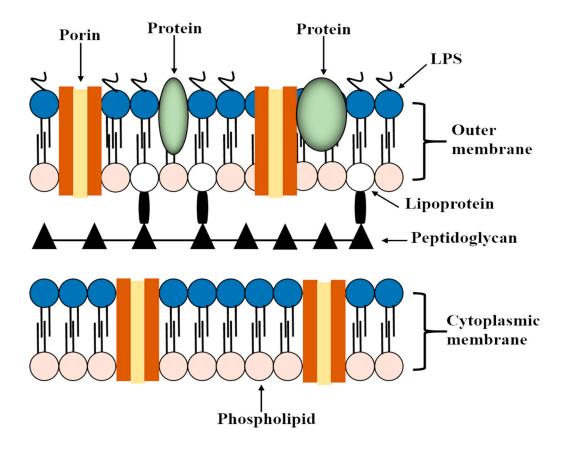
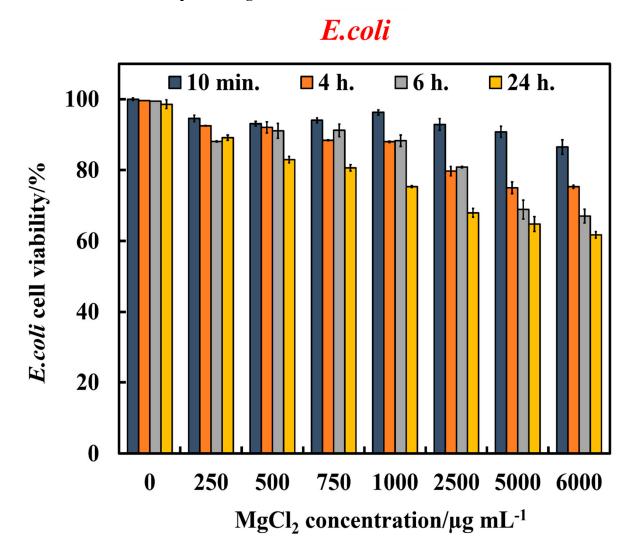


Figure S10. Schematic overview of the bacterial cell wall [37].



9. The antibacterial impact of MgCl₂ on *E. coli*

Figure S11. Antibacterial impact of various concentration of MgCl₂ towards *E. coli* for various exposure times. The experiment was achieved via incubated of *E. coli* with MgCl₂ for one day.

10. Comparison of the antimicrobial activity of uncoated and polyelectrolyte-coated Mg(OH)₂NPs on *S.cerevisiae*, *C. reinhardtii* and *E. coli*

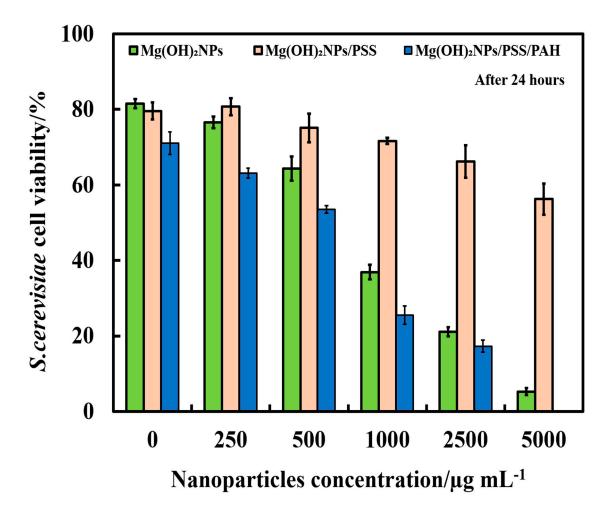


Figure S12. *S.cerevisiae* cell viability after incubation as a function of nanoparticle concentration for up to 24 hours with uncoated and polyelectrolyte-coated Mg(OH)₂NPs.

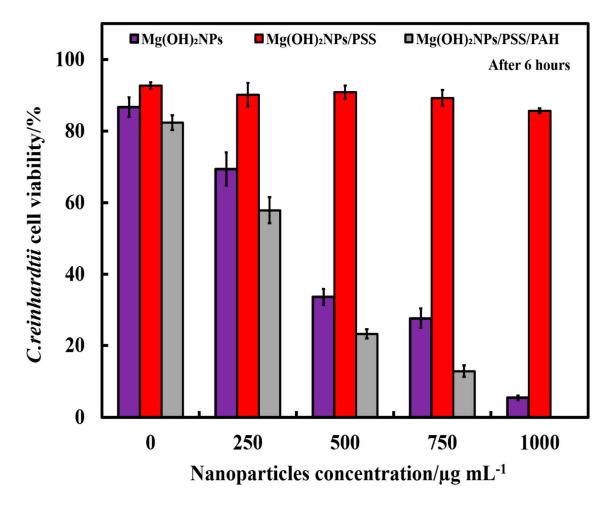


Figure S13. The antialgal activity of uncoated and polyelectrolyte-coated Mg(OH)₂NPs.

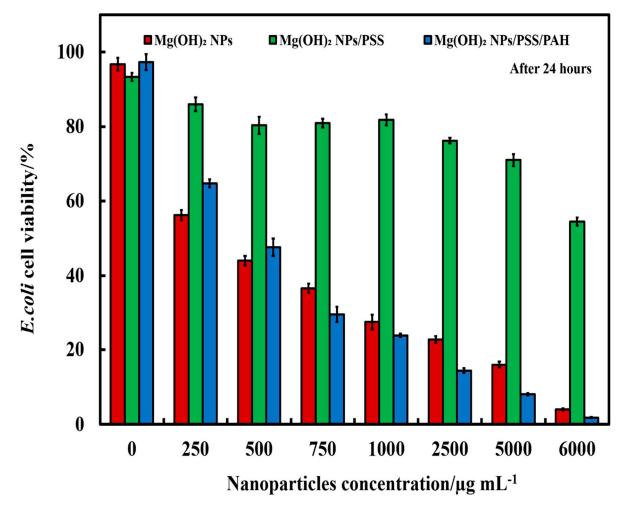
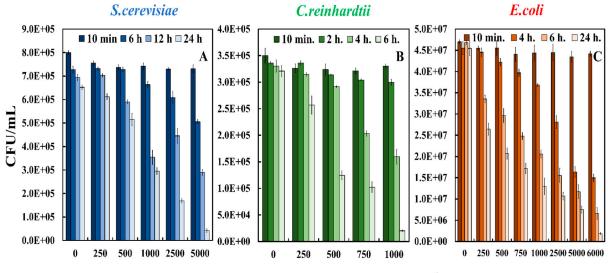


Figure S14. Relationship between the antibacterial efficiency of uncoated and polyelectrolytecoated Mg(OH)₂NPs on the viability of *E. coli*. *E. coli* was incubated for one day to 0, 250, 500,750, 1000, 2500, 5000 and 6000 μ g mL⁻¹ of different types of nanoparticles.

11. Colony forming units (CFUs) of the bare Mg(OH)₂NPs and PAH-modified Mg(OH)₂NPs against cells

Figure S15, Figure S16, Figure S17 and Figure S18 shows the CFUs assay of *S. cerevisiae, C. reinhardtii* and *E. coli* where the control samples of untreated bacteria were compared with the ones treated with bare Mg(OH)₂NPs, Mg(OH)₂NPs/PSS and Mg(OH)₂NPs/PSS/PAH. The stock cultures of *C. reinhardtii* which were used for testing with typical concentration of 4×10^5 cells per mL and 9×10^5 cells per mL for *S. cerevisiae* determined by automatic cell counter (Cellometer Auto X4) and the *E. coli* bacterial culture stock was approximately 5×10^7 cells per mL.



Mg(OH)₂NPs concentration/µg mL⁻¹

Figure S15. Colony forming unit (CFU) count of bare Mg(OH)₂NPs on (A) *S.cerevisiae* (B) *C. reinhardtii* and (C) *E. coli* at various particle concentrations. The cells were incubated with the Mg(OH)₂NPs at different periods of time shown.

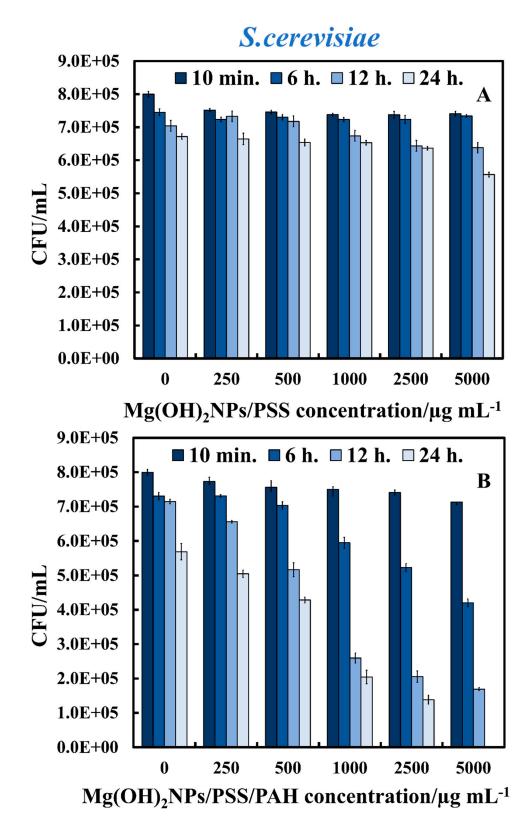


Figure S16. Colony forming unit (CFU) count of *S.cerevisiae* as a function of nanoparticle concentration after incubation for up to 24 hours with (A) Mg(OH)₂NPs/PSS and (B) Mg(OH)₂NPs/PSS/PAH.

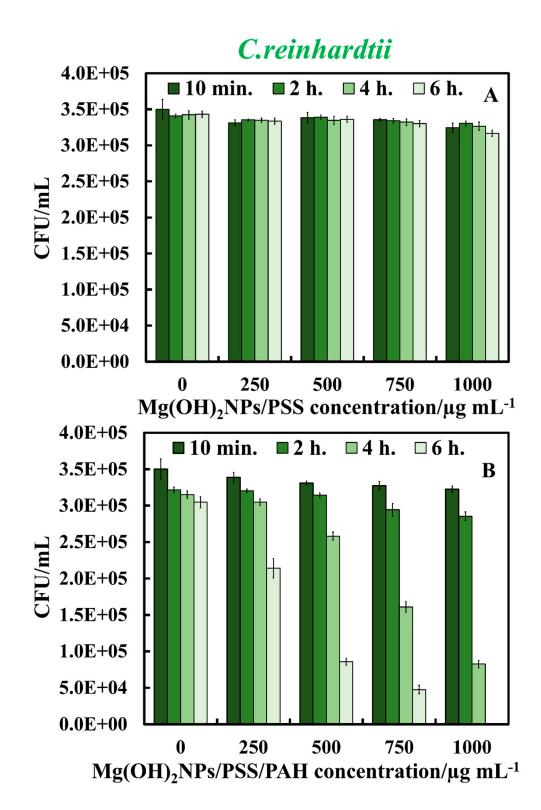


Figure S17. Colony forming unit (CFU) count of *C. reinhardtii* as a function of nanoparticle concentration after incubation for up to 6 h with (A) Mg(OH)₂NPs/PSS and (B) Mg(OH)₂NPs/PSS/PAH.

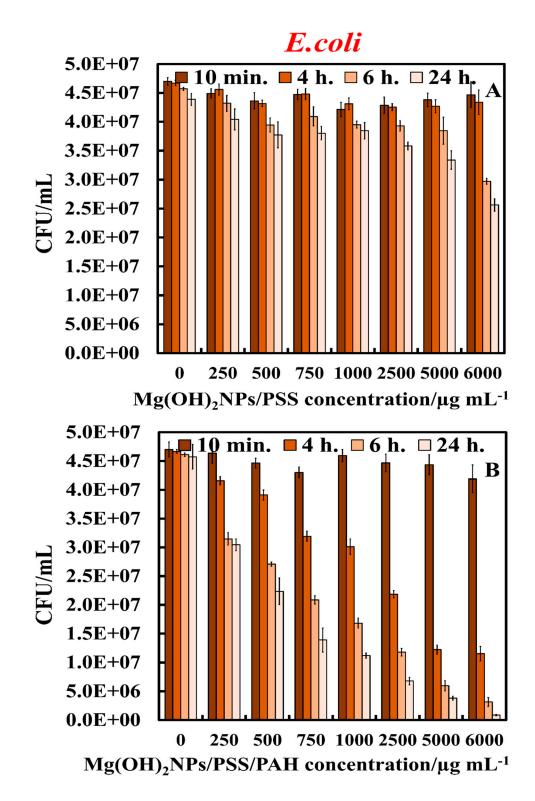
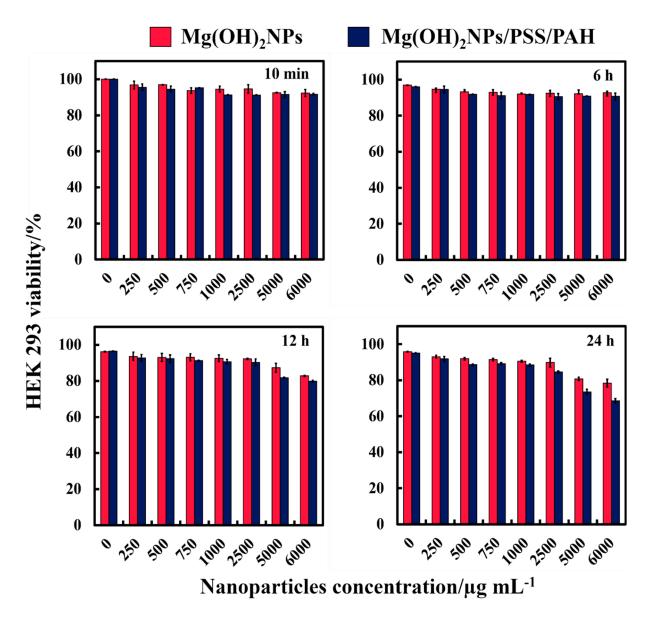


Figure S18. Colony forming unit (CFU) count of *E. coli* after treatment with (A) Mg(OH)₂NPs/PSS and (B) Mg(OH)₂NPs/PSS/PAH for various incubation times as a function of the NPs concentration.



12. HEK 293 cell viability in the presence of bare Mg(OH)₂NPs and PSS/PAH-coated Mg(OH)₂NPs

Figure S19. Comparison of the cell viability of human embryonic kidney cells (HEK 293 cell line) upon incubation as a function of nanoparticle concentration for up to 24 h with bare Mg(OH)₂NPs and Mg(OH)₂NPs/PSS/PAH.

13. Antimicrobial assay of SiO₂NPs on E. coli

Figure S20 shows the antimicrobial assay of SiO₂NPs on *E. coli* for up to 24 hours of exposure at different SiO₂NPs concentrations. We used SiO₂NPs as a negative control for comparing the toxicity of the Mg(OH)₂NPs. One can see a very small effect on the presence of bare SiO₂NPs on the *E. coli* viability. Note that even the control sample of *E. coli* have lost some of their viability over this period of time due to depletion of the media.

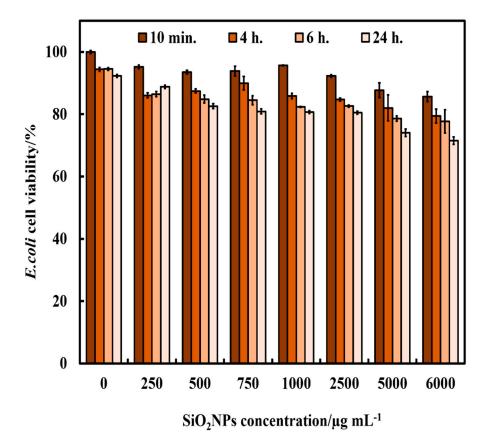


Figure S20. Antibacterial activity of SiO₂NPs at various concentrations on *E. coli*. The *E. coli* cells were incubated with the SiO₂NPs at 10 min, 4 h, 6 h and 24 h of exposure before being washed and tested for their cell viability.

14. Anti-yeast, anti-algal and antibacterial activity of free PAH in solution

There is no free PAH in the PSS/PAH-coated Mg(OH)₂NPs suspension since we wash the PSS/PAH-coated Mg(OH)₂NPs multiple times by centrifugation and discard the supernatant. To check possible effects from leaching polyelectrolytes we did antimicrobial testing of free PAH on S. cerevisiae, C. reinhardtii and E. coli. Figure S21 shows the antimicrobial assay of the free PAH on S. cerevisiae, C. reinhardtii and E. coli for up to 6 hours for C. reinhardtii and 24 hours for S. cerevisiae and E. coli of exposure. Both runs were done at the varying overall PAH concentration and different incubation times. One can see a very small effect on the presence of free PAH on the cells viability. One can conclude that the free PAH does not measurably impact the cell viability up to 5000 μ g mL⁻¹ for *S. cerevisiae*, 1000 μ g mL⁻¹ for *C*. *reinhardtii* and 6000 µg mL⁻¹ for *E. coli*. Note that in our Mg(OH)₂NPs/PSS/PAH nanoparticles there is not ant free PAH and free PSS as the particles have undergo multiple washing/centrifugation cycles after their surface functionalization. However, at these concentrations of the PAH- coated on Mg(OH)₂NPs, the effect of the Mg(OH)₂NPs on S. cerevisiae, C. reinhardtii and E. coli is very significant - see Figure 4, Figure 9, Figure 10 and Figure 11 in the main paper, respectively. Therefore, one may conclude that the PSS/PAHcoated Mg(OH)₂NPs show excellent anti-yeast, anti-algal and antibacterial with these cells which is not related to free PAH.

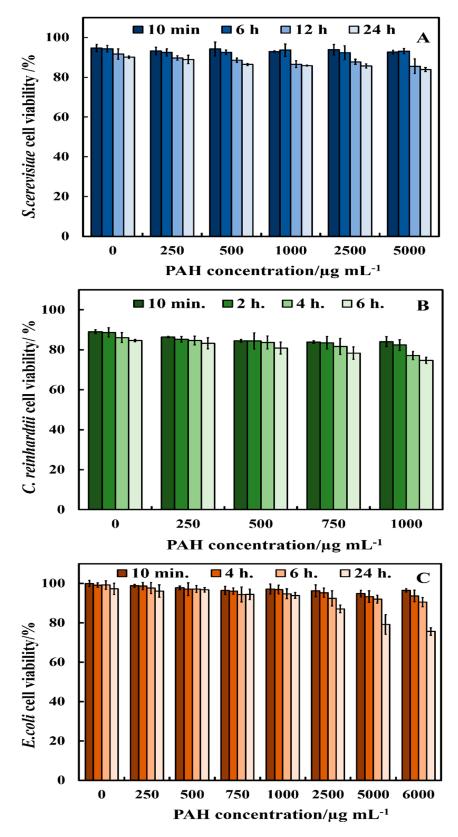


Figure S21. The anti-yeast, anti-algal and antibacterial activity of free PAH at various concentrations on (A) *S. cerevisiae*, (B) *C. reinhardtii* (C) *E. coli*. The cells were incubated with the free PAH at different times of exposure before being washed and tested for their cell viability.