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PAPER

Fabrication of viable cyborg cells with cyclodextrin functionality

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5 We describe two alternative methods for surface functionalisation of *Saccharomyces cerevisiae* cells with cyclodextrin molecules without affecting the cell viability. The first strategy involved using epichlorohydrin as a cross-linking agent which binds covalently the cyclodextrin to the glycoproteins on the cell wall. The second strategy of interfacing of the cells with CD involved polyelectrolyte mediated deposition of cyclodextrin sulphate on the cell surface. We used the formation of host-guest inclusion
10 complex of a dye with the grafted cyclodextrins to estimate the average number of CD molecules grafted per cell which can reach up to hundreds of millions of CD molecules. This indicates more than one monolayer of CDs on the cell surface within the surface layer surrounding the yeast cell membrane. Fluorescein diacetate was used to check the viability of the cells after functionalisation. Living cells functionalised with CDs may find many potential applications as they can be loaded with drugs,
15 immunosuppressants and other molecules forming inclusion complexes with their cyclodextrin interface. Therefore, we foresee such cells being used as novel selective biosorbents in polluted waters, whole cell biosensors, drug delivery, cell therapy and cell implants applications.

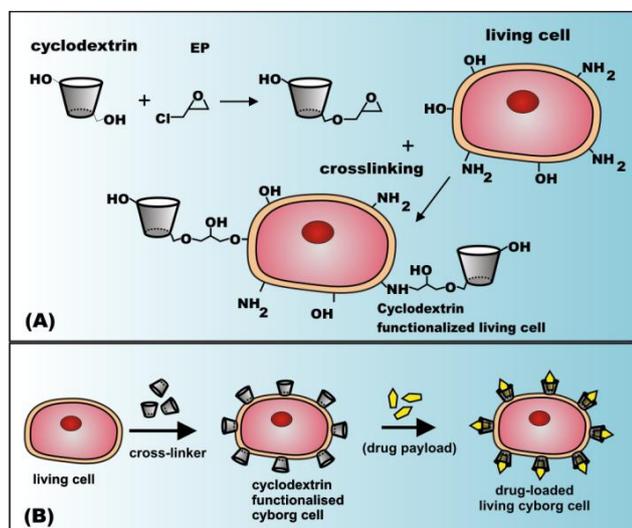
Introduction

Interfacing living cells with nanoparticles and polymers is an
20 emerging new field¹⁻⁴ which brings together microbiology, surface chemistry, and physical chemistry of colloids. The modified cells can perform novel functions due to their different interface with the external media. Such “cyborg cells”^{1a} have been created by using a variety of living cells species, ranging from microbial cells
25 (bacteria,^{1b} algae² and fungi^{3,4}), multi-cellular organisms⁵ (nematode) to mammalian cells for a wide range of possible new applications. Many native cells bear negative surface charge due to dissociation of surface carboxylic groups originating from carbohydrates or proteins. This has been utilised in a range of cell
30 functionalisation techniques based on the Layer-by-Layer (LbL) deposition of oppositely multi-charged species such as polyelectrolytes,⁶ magnetic particles,^{7,8} colloidal particles,⁹ metal nanoparticles^{6,10} and oligonucleotides. Cyborg cells produced using the LbL method have been used in whole-cell biosensors,
35 toxicity micro-screening devices¹¹ and cell-based therapies¹² while non-viable cells have been used in areas such as biosorbents^{13,14} biocatalysts and microelectronics.⁴ For example, magnetically-functionalised yeast cells were shown to effectively absorb heavy metal ions as well as industrial dyes.¹⁵ While cyborg cells
40 interfaced with a variety of nanoparticles and polymers¹⁻¹⁰ have already been fabricated, the surface functionalisation of living cells

with cyclodextrins (CDs) has not yet been reported. Cyclodextrins are derived from starch by enzymatic reactions¹⁶⁻¹⁹ and are used as molecular containers in a range of pharmaceutical formulations²⁰,
45 ²¹, drug delivery applications²²⁻²⁵, food^{26, 27}, flavours^{28, 29} and consumer personal care products³⁰. Cells functionalised with cyclodextrins may find many exciting new applications as living drug carriers where the cyclodextrins grafted on the cell surface can carry additional payload supplementing the cells own
50 functions. For example, cyclodextrin functionalised cyborg cells can be loaded with immunosuppressants and used as cell implants. In addition, such cell can be used as whole cell biosensors and biosorbents. However, grafting cyclodextrins to living cells can be a challenge as it requires aggressive reaction conditions which may
55 adversely affect the cells viability.

Here we report for the first time a successful functionalisation of living yeast cells with three types of cyclodextrin (α -CD, β -CD and γ -CD). We developed two alternative strategies for interfacing cells with cyclodextrin. The first one involves using
60 epichlorohydrin (EP) as a cross-linking agent as shown schematically in Scheme 1. This method does not significantly impact the cells viability and involves the following stages: (i) The cells are incubated in aqueous solution of cyclodextrin. (ii) Epichlorohydrin is added to the cell dispersion which grafts the
65 cyclodextrin molecules to the cells surface. (iii) The cyclodextrin-functionalised cells are filtrated from the solution. The advantage of using EP as a cross-linker for CD is that the reaction can be

carried out at room temperature at low EP concentration which preserves the cells viability. We have applied this method to functionalise yeast cells with cyclodextrins as a proof of principle.



Scheme 1 (A) Cross-linking reactions of living cells with cyclodextrins (CDs) and epichlorohydrin (EP). (B) Fabrication steps of living CD-functionalised cyborg cells loaded with drugs or immunosuppressant.

In addition, we employed a second strategy which involves the layer-by-layer deposition of cationic polyelectrolyte and β -CD sulphate on yeast cells and assessed the effect of the number coats on the cell viability. With discuss both methods in details below.

Materials and Methods

Materials

Alpha cyclodextrin (α -CD), beta cyclodextrin (β -CD), gamma cyclodextrin (γ -CD), β -cyclodextrin sulphate sodium salt, Epichlorohydrin (EP), Fluorescein Diacetate (FdA³¹), polyallylamine hydrochloride (PAH, MW 58 kDa) all of 98 % purity were purchased from Sigma (UK) and were used without further purification. Methyl Orange (MO, grade Reag. Ph. Eur.) was sourced from Sigma (UK). Deionised water was used in all experiments, obtained by using a Milli-Q® water system. Baker's dry yeast was purchased from TESCO (UK).

Functionalisation of yeast cells with cyclodextrins using epichlorohydrin as a cross-linker

The yeast cells were pre-washed several times with Milli-Q water prior to their functionalisation. The cross-linking reaction was carried out by using a fixed amount of yeast cells (~0.5 g) while the concentration of cyclodextrin (α -CD, β -CD or γ -CD, from Sigma) in the reaction mixture was varied from 0 to 0.4 % m/v. For each particular sample, a fixed amount of CD was dissolved in 5 mL milli-Q water followed by addition of the cells and homogenisation with a magnetic stirrer. The CDs were grafted onto the cells surface by adding 2 μ L of EP to each sample including the control one (without CD). After incubation (with stirring) in the reaction mixture at room temperature for 24 hrs the cell samples were centrifuged (3000 rpm for 5 minutes) and washed first with milli-Q water and several times with 2 mM phosphate buffer solution at pH 7. The functionalisation of yeast

cells with cyclodextrins using epichlorohydrin as a cross-linker was carried out under buffered condition using sodium hydrogen phosphate buffer solution at pH 7. Hence pH does not change in the course of the reaction. Epichlorohydrin is potentially toxic for the cells at moderate and high concentrations³⁴ therefore a very small amount was used to reduce its cytotoxic effects on the cells so that the functionalized cells retained their viability after functionalisation. This protocol was performed for yeast incubated with both α -CD, β -CD and γ -CD. To demonstrate the success of this procedure in grafting cyclodextrin to the surface of the yeast cells, we used the ability of the cyclodextrins to form host-guest non-covalent inclusion complexes. A sample of 75 mg of each batch of CD- functionalised yeast cells was incubated in a stock solution of 60 μ M Methyl Orange in 2 mM phosphate buffer at pH 7 for 60 minutes under agitation. Subsequently, the cells were centrifuged and the supernatant was collected through 0.2 μ m Whatman® Anotop® syringe filter. The MO concentration in the filtrates was measured spectroscopically by using standard calibration curves from a series of MO solutions at the same pH 7 phosphate buffer concentration.

UV-Vis spectroscopy

UV-vis spectroscopy was used to quantitatively analyse the formation of host-guest complex between MO and the cyclodextrins attached covalently to the yeast cell surfaces. A series of standard aqueous solutions of MO in 2 mM sodium phosphate buffer at pH 7 were prepared in 25 mL volumetric flasks and their absorbance spectra from 600 nm to 200 nm were measured by double beam Perkin Elmer UV-vis spectrometer Bio Lambda 40 using UV Winlab software. From these spectra, calibration curves were produced by plotting the absorbance at λ_{\max} = 460 nm against the MO concentration of a given standard aqueous solution. 2 mM of aqueous solution of sodium phosphate buffer (pH 7) was used as a base line of the absorbance in the UV-vis instrument. The absorbance spectra of 60 μ M MO filtrates that were used to incubate α -CD, β -CD or γ -CD functionalised yeast cells were also obtained by scanning the absorbance vs. the wavelength to find out how the absorbance at λ_{\max} was affected by the formation of MO-CD inclusion complex on the cells surface. The amount of MO that formed inclusion complexes with α -CD, β -CD or γ -CD molecules was calculated using the calibration curve of the absorbance (at 460 nm) of aqueous standard solutions of MO. From the MO concentration in the filtrate of solution exposed to the same mass of CD-functionalised cells and a control experiment with non-functionalised yeast cells, the number of cyclodextrin per yeast cell was estimated using the equation

$$n_c = \frac{\pi(C_r - C_\beta)V_s N_A \rho_c d_c^3}{6m}, \quad (1)$$

where n_c is the number of β -CD per cell, C_r is the MO residual concentration (in M) after filtering out the yeast cells (for the β -CD non treated) and C_β is the equivalent concentration of MO after filtrating the β -CD-treated cells. The concentration difference in Eq.(1) reflects the amount of MO retained in the MO-CD inclusion complexes on the cell surface. Here V_s is the volume of the MO stock solution (60 μ M) incubating the sample of cells of mass m , N_A is Avogadro's number, ρ_c is cell density and d_c is the average cell diameter.

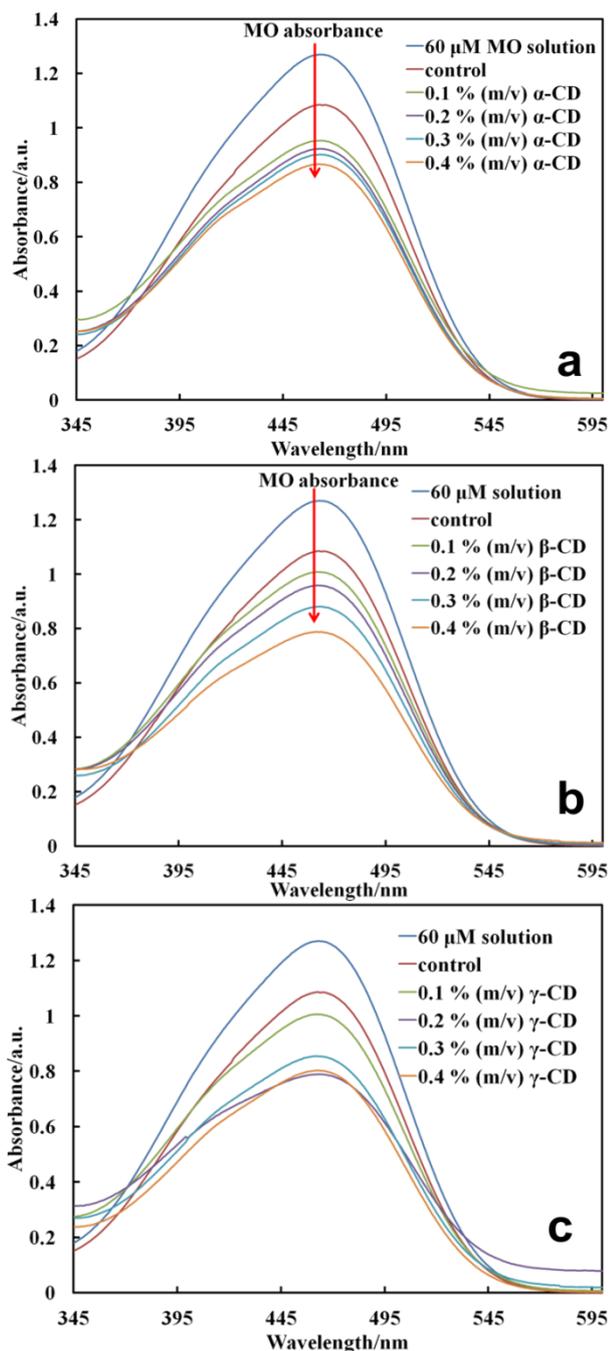


Fig. 1 Absorbance of the filtrate of MO solution after incubation with cell functionalised with CDs at different concentration in the reaction mixture at 5 mM EP. Results are presented with cells functionalised with (A) α -CD, (B) β -CD, (C) γ -CD. The graphs also give the absorbance of the filtrate from the control experiment with cells which were not treated with CD and the stock solution of MO. The red arrows show the decrease in MO absorbance as CD concentration is increased.

In this estimate, we assume that the cells are spherical and that each grafted CD molecule on the cells surface captures one MO molecule from the stock solution.

Viability test of the yeast cells

The viability of the cells after the functionalisation was assessed by using fluorescence microscopy. This was done by incubating

small amount of washed functionalised yeast cells in Fluorescein Diacetate (FdA) solution for 10 minutes. The samples were covered with aluminium foil to reduce photo-bleaching and after 10 minutes the samples were washed multiple times with milli-Q water and analysed with Olympus BX-51 fluorescence microscope using mercury vapour lamp as a light source and Olympus Burner U-RFL-T-200 for mercury vapour excitation. The yeast cells were irradiated with blue light (435 nm) using FITC filters sets. The viability test compares the number of fluorescing cells and non-fluorescing cells in a given sample on the microscope glass slide. Thus counting of the fluorescing and the non-fluorescing cells was done by overlaying images from fluorescence microscopy and brightfield optical microscopy of the same field of view of the sample on the microscope glass slide and then counting the fluorescing cells out of that total number of cells. In Figure 6 we present the viability percentage calculated for the control and the CD-functionalised cells in the overlaid images.

Influence of the epichlorohydrin on the number of cyclodextrin grafted on the yeast cells

This was investigated using β -CD and carried out by keeping the mass of yeast cells and mass of cyclodextrins fixed but varying the amount of the EP in the reaction mixtures. The influence of varying EP concentration on the uptake of methyl orange by the non-functionalised cells was done by running control experiments where the amount of yeast cells was the same but there was no cyclodextrin in the reaction mixture. EP may cause cell-cell binding which would manifest itself as a cell aggregation. However, we did not observe significant cell aggregation by studying the cells under optical microscope which suggested that the reaction was mostly confined to CD-cell surface binding. We attribute this to the moderately low cell concentration in the reaction mixture and the stirring of the solution during the cross-linking reaction.

Functionalisation of yeast cells using Layer-by-Layer sequential deposition of PAH and β -cyclodextrin sulphate

Yeast cells were first subjected to a cycle of washing with milli-Q water/centrifugation (4 times) to remove any impurities. The functionalisation was carried out by sequential deposition of PAH and β -CD sulphate. In this procedure, as sample of 0.4 g of yeast cells dispersed in 3 mL of milli-Q water was first delivered by a SP100i syringe pump (Flow rate 6 mL/h) to a stirred (250 rpm) 3 mL of 1 mg/mL aqueous solution of PAH. The excess PAH after the deposition step was rinsed 4 times with milli-Q water through a cycle of centrifugation (3000 rpm for 5 min) and decanting the supernatant. Then, the PAH pre-coated yeast cells were re-dispersed in 3 mL milli-Q water and this then was delivered by a SP100i syringe pump (Flow rate 6 mL/h) to a stirred (250 rpm) 3 mL of 1 mg/mL aqueous solution of β -cyclodextrin sulphate. The excess β -cyclodextrin sulphate after adsorption step was rinsed 4 times with milli-Q-water through a cycle of centrifugation (3000 rpm for 5 min) and decanting the supernatant) leaving a second layer of negatively charged β -CD sulphate on the yeast surface. The above procedure was repeated up to until 4 layers of PAH- β -CD sulphate were produced. Some variations in the concentrations of both PAH and β -CD sulphate were introduced (2 and 3 mg/mL) while keeping all other parameters constant. The aqueous phase was modified by using 1M NaCl.

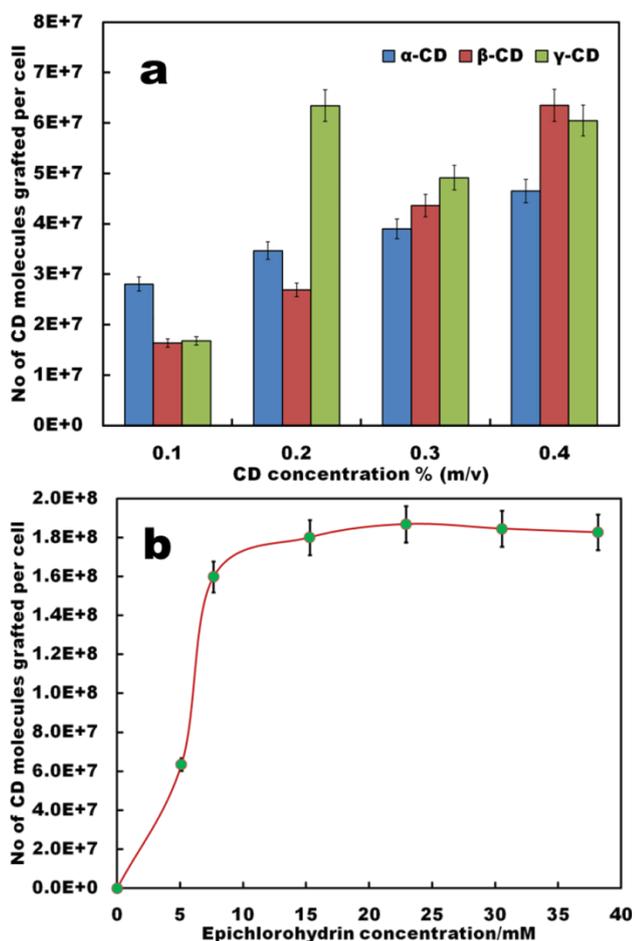


Fig. 2 (a) Number of CD molecules grafted per cell as a function of the concentration of CD (α , β and γ) in the reaction mixture at fixed EP concentration (5 mM). (b) Effect of the EP concentration on the number of CDs grafted per cell at fixed CD concentration (0.4% m/v β -CD).

Determination of the yeast cells average cell diameter

The yeast cells size distribution was measured by laser diffraction using Malvern Mastersizer 2000 coupled to Hydro 2000sm dispersion unit. The instrument covered a particle size range of 20 nm to 2000 μ m. The background was set by running the measurement in milli-Q water without the sample. The samples were then added manually to a dispersion tank filled with milli-Q-water and stirred at 330 rpm. The diluted samples were then pumped into the measurement cell of the Mastersizer. The cell size distribution was calculated from Fraunhofer's model using water as a dispersant with refractive index of 1.33. The average of ten runs was taken as a representation of measured particle size distribution.

Zeta potential measurements

The functionalisation of yeast cells by consecutive deposition of PAH and β -CD sulphate was monitored by measuring the cells zeta potential. After each deposition, cells were washed by a cycle of centrifugation thrice and subsequently dispersed in milli-Q-water to dilute them. The zeta potential of the dispersed cells was measured using Malvern Zeta 3000HS with a flow-through sample cell. The average of 10 measurements was taken to represent the measured potential.

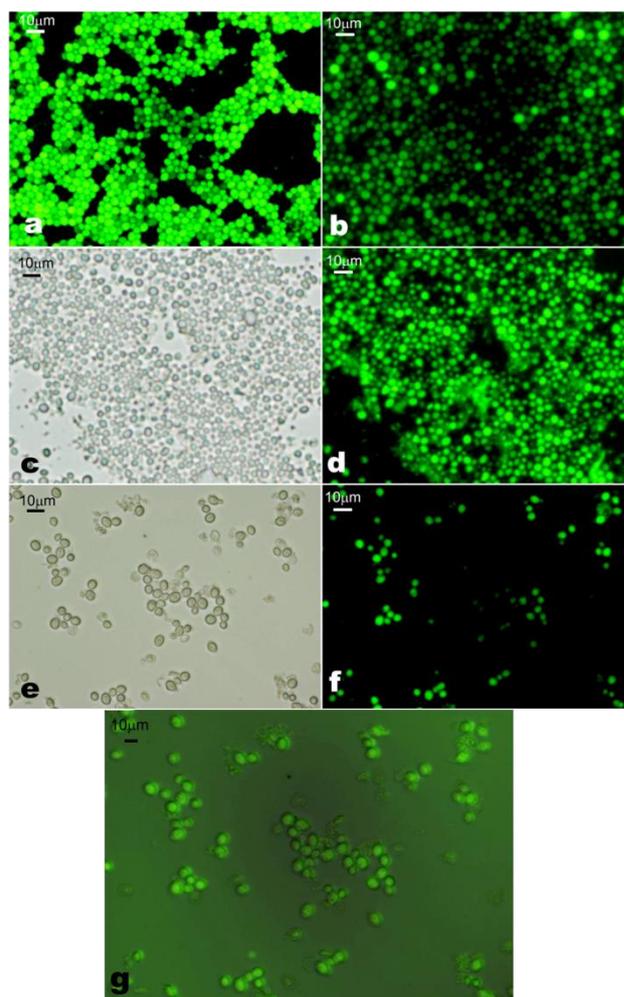


Fig. 3 Cell viability test with fluorescein diacetate showing the cell fluorescence after functionalisation with β -CDs. (a) and (b) are the control sample (non-treated) in bright field and fluorescence microscopy with a FITC filter set. (c) and (d) are yeast cells functionalised with 0.1% m/v β -CD in bright field and fluorescence microscopy, respectively while (e) and (f) show the yeast cells functionalised with 0.4% m/v β -CD. The concentration of the EP cross-linker used in both experiments was 0.6 mM. (g) Overlay of image E and F showing that above 80% of the cells retain their viability.

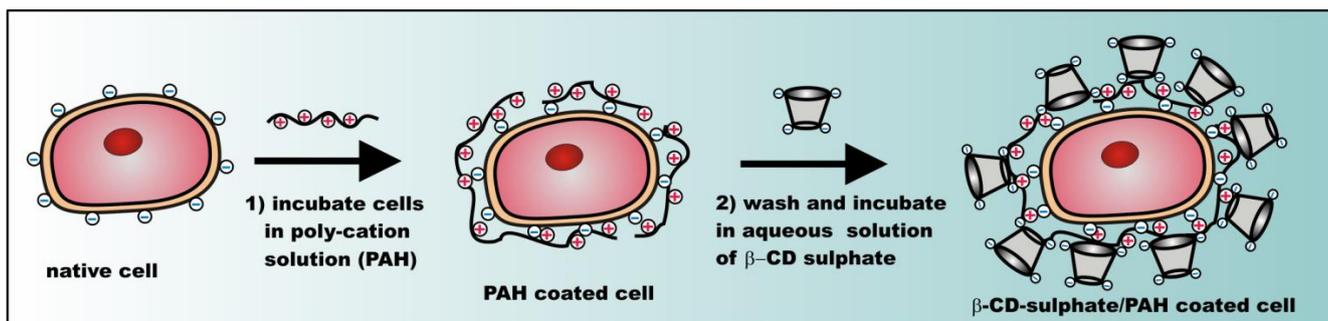
Results and Discussion

In this study, we have established two methods of functionalising living yeast cells with cyclodextrins molecules which allow active components to be encapsulated on the cell surface. The protocols involved in both cases are simple, preserve the cell viability and do not involve expensive equipment or chemicals.

Functionalisation of living yeast cells with CDs using epichlorohydrin as cross-linker

Scheme 1 shows only two of the possible reaction of cross-linking of CDs to the surface of cells. However, the opening of the epoxy ring of EP can also produce other side reactions of conjugation of CD in complex polymers.³²

Figure 1 shows the absorbance of the MO solution filtrates after incubation with a fixed amount of cells functionalised at constant amount of EP but varying concentration of various CDs in the reaction mixture.



Scheme 2 Schematic illustrating the sequential deposition of cationic polyelectrolyte (PAH) and β -CD sulphate to functionalise the cell surface with CD.

One sees that the functionalisation of the cell with CD consistently lowered the peak absorbance of MO with the increase of the CD concentration in the reaction mixture. Note that the control sample (cells treated with EP without CDs) also lowered the peak absorbance of MO but to lower extent compared to cell samples cross-linked with CDs. This is related to the non-specific adsorption of MO on the cell surface layer. Since the binding constant of this type of dyes in the inclusion CD complex is very large¹⁷ and the MO concentration in the stock solution is high we can assume that all CD molecules on the cell surface host a MO molecule upon incubation. The binding constant of MO- α -CD is 7300 mol⁻¹dm³ while MO- β -CD has a value of 2700 mol⁻¹dm³ for 1:1 complexes according to studies done by Hamai and Handa.³³ From the material balance of the specifically adsorbed MO on the CDs grafted on the cell surface we estimated the number of CD molecules per cell using the Eq.(1). The average diameter of the yeast cells was $6 \pm 1 \mu\text{m}$ from the Mastersizer measurement. Note that the absorbance of MO in the filtrates varies for a series of cell samples functionalised with

CD (α -CD, β -CD and γ -CD) at different concentrations in the reaction mixture and fixed concentration of EP. As expected, the depletion of the MO from the stock solution depends on the type of CD used for the cells functionalisation. This was not surprising because CDs have different cavity volumes which determined whether the incoming dye guest molecule formed a perfect fit, loose fit or it was too big to fit in the CD cavity. As seen in Figure 1, comparing the cells functionalised with the same highest amount of CDs ($\sim 0.4 \text{ g/mL}$) and incubated in MO at the same concentration, β -CD functionalised cells showed the lowest maximum absorbance. We observed a consistent increase in the number of CDs per cell with the increase of the CD concentration upon grafting (Figure 2a). This was consistent with our expectations since we were relying on the absorbance of the dyes i.e. the dye which “disappeared” from the solution must be trapped in the CD cavity hence as more and more dye molecules were trapped by CDs grafted on the cell few were left in the filtrate therefore giving low absorbance maximum. However, we found that the extent at which the maximum absorbance was lowered depended on the type of CD used. In Figure 2a, the number of CD molecules grafted per cell is not linearly dependent on CD concentration, especially at the γ -CD concentration of 0.2 % (m/v). There are two possible factors which may contribute to this non-linearity of the attached γ -CD per cell and the γ -CD concentration:

(i) The number of CD molecules grafted per cell was estimated using host-guest interactions between CD hydrophobic cavity and the methyl orange. These interactions are formed through non-covalent bonds and depend on the perfect fit between the host and the guest molecules. Therefore there is a different dynamic equilibrium between the complex and free molecules, which could explain this observed non-linearity especially for γ -CD with the largest cavity which forms the least stable complex with methyl orange. (ii) The reactivity of the 6'-OH groups of the γ -CD can be different than those of α -CD and β -CD due to the higher conformational flexibility of the γ -CD molecule.

We studied the effect of the EP cross-linker concentration on the average number of CD molecules per cell by using a fixed amount (0.5 g) of yeast cells and β -CD at $\sim 0.4 \text{ g/mL}$ while varying the EP concentration in the reaction mixture. Figure 2b shows that for β -CD functionalised cells, the increasing of the EP concentration lead to a sharp increase in the average number of CD molecules grafted per cell and a plateau was reached above 12 mM EP which was an indication of saturation of the binding sites for EP on the cell surface layer. However, it has been reported that the presence of EP in the solution may have cytotoxic effect on the exposed cells even at low and moderate EP concentration for a range of mammalian cells.³⁴

We tested the viability of the CD-functionalised yeast cells by using Fluorescein Diacetate (FdA³¹). FdA is a non-ionic dye which diffuses through the cell membrane and can undergo hydrolysis by non-specific esterase to produce fluorescein molecule which accumulates inside the cells which can be detected by fluorescence microscopy.¹⁸ Upon treatment with FdA the yeast cells turn fluorescent if their cell membranes are intact and the enzymes in the cell interior are active. Our FdA test results show that most of the yeast cells exhibit green fluorescence which indicates that the cells retained their viability after functionalisation with CD in the presence of EP (Figure 3).

Functionalisation of yeast cells by LbL deposition of PAH and β -cyclodextrin sulphate

In addition to grafting CD molecules to the yeast cells using EP as a cross-linking agent, we exploited the availability of the negative surface charges present on the native yeast cells due to the ionization of polysaccharides making the cell wall. We used the negative charges of the yeast cells to anchor cationic polyelectrolyte (PAH) and deposit anionic β -CD sulphated sodium salt by LbL deposition. Scheme 2 shows the schematic for the LbL

functionalisation of yeast cells with β -CD sulphate. After deposition of each PAH/ β -CD sulphate layer on the yeast cells, the dispersion was washed by a cycle of re-dispersing in milli-Q water and centrifugation at 3000 rpm for 5 minutes (four times) to remove the excess polyelectrolyte. The presence of each deposited thin layer was assessed by measuring the zeta potential of the functionalised cells after washing and re-dispersing in milli-Q water. We also varied the concentrations of the polyelectrolytes from 1 mg/mL to 3 mg/mL. As can be seen from Figure 4, the success of the sequential build-up of the coatings from PAH and β -cyclodextrin sulphate was reflected in oscillation of the zeta potential of the coated cells. When PAH was the outer coat, the zeta potential increased (positive) compared to the non-coated yeast cells (negative). The PAH does not get internalised by the yeast cells due to the relatively thick (about 200 nm) cell wall consisting of glycoproteins and carbohydrates which surrounds the lipid membrane. The PAH coated cells were stable since the cells were washed by centrifugation and when observed under optical microscope, the cells still maintained their integrity and the zeta potential measurement showed that the PAH was on their surface as the potential was positive ($\zeta \approx +40$ mV) while the non-coated cells are negatively charged ($\zeta \approx -20$ mV). Note that when the β -CD sulphate was the outer layer, the measured zeta potential was more negative than the non-coated yeast cells. The reversal of zeta potential in the LbL deposition has been reported in many articles to confirm the deposition of oppositely charged polyelectrolytes (e.g. see Ref.2) For example, Hillberg and Tabrizian have shown the charge alternation with successive deposition of chitosan (polycation) and alginate (polyanion) to encapsulate *E.coli*.¹² This result therefore was a confirmation of the functionalisation of yeast cells with modified β -CD. We observed that as the number of layers was increased, the zeta potential became slightly more negative as well as become more positive with increasing the layering. This shows that the cell's surface becomes more charged with the increase the number of deposited layers and this trend was observed for all concentrations up to 4 layers. As indicated above epichlorohydrin is cytotoxic³⁴ therefore the functionalisation of cells using epichlorohydrin as a cross-linker may be an issue for very sensitive cell cultures and in vivo applications. This is why we also used the layer-by-layer method as alternative method for this situation. In addition the PAH in this procedure may also be replaced with more biocompatible polycations such as chitosan or branched polyethyleneimine (PEI 1.8k) for CD-functionalisation of more sensitive cell cultures and for safer in vivo applications of the functionalised cells.

The best results of LbL deposition were achieved at 2 mg/mL polyelectrolyte and 2 mg/mL β -CD sulphate concentrations. The thickness, strength and morphology of polyelectrolyte layers can be further tailored by altering the pH, the ionic strength and the poly-ion materials.³⁵ We investigated the effect of the ionic strength by depositing the polyelectrolytes from solution of 1 mM NaCl. As can be seen in Figure 4, the sequential deposition of PAH and β -CD sulphate in the presence of 1 mM NaCl reduced very slightly the magnitude of the zeta potential which is attributed to the electrostatic screening.

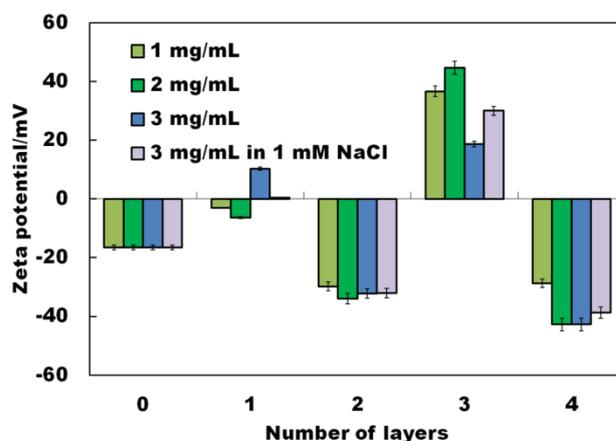


Fig. 4 Zeta potential change with the sequential deposition of PAH (polycation) and β -CD sulphate on the surface of viable yeast cells.

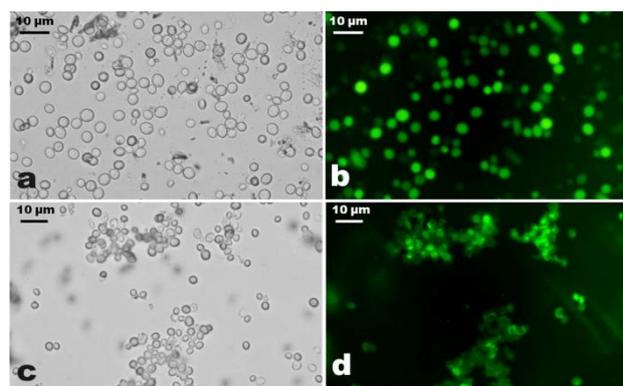


Fig. 5 Micrographs showing the viability of yeast cells testes with FdA. Transmitted light for bare yeast cells (a) and (b) fluorescence micrographs for bare yeast cells while (c) and (d) show transmitted and fluorescence micrographs after the deposition of four consecutive layers of PAH and β -CD sulphate.

The calculation of the number of CDs deposited on the cell surface by layer-by-layer assembly is challenging due to the presence of PAH between the CD layers as it also adsorbs methyl orange molecules electrostatically in addition to the ones bond to the CDs. However, we have shown using zeta potential measurement that the β -CD sulphate molecules were deposited on the surface of the cells as shown by the cell surface charge reversal (see Figure 4).

Cell viability of CD-functionalised cells by the LbL method

We monitored the viability of the cells after each deposition using FdA. Samples taken after each layer deposition were incubated in a solution by adding a drop of FdA solution (5 mg/mL) dissolved in acetone. We observed that with the increase of the number of PAH/ β -CD sulphate layers, the fluorescence intensity of the cells decreased. After the fourth layer, the viability as tested with FdA visibly worsened and hence we only applied this LbL procedure up to four layers. One possible explanation for this result could be that the build up of CD molecules on the cell surface depletes the FdA which can lead to lower penetration in the cell interior. We also observed some partial aggregation in CD-functionalised cells compared to the native ones. This was dependent on the way of introducing the cell into the polyelectrolyte solution and could be subject of further

optimisation to avoid formation of large aggregates. Figure 5 shows the micrographs of the bare yeast cells and after the deposition of the fourth layer.

In practical applications, it could be a good idea to check the stability of the CD/PAH/cells in the presence of a serum. We will make such tests in future publications of results with CD-functionalised cells for particular applications. However, even the present results indicate that the CD/PAH-coated cells are stable upon further coating with oppositely charged polyelectrolytes as presented in Figure 4. Thus, further treatment with PAH and β -CD sulphate does not incur cell aggregation provided that the polyelectrolyte is in excess with respect to the cells. However, adding of small amount of polyelectrolyte of opposite charge to the cell suspension causes cell aggregation rather than charge reversal.

Conclusions

In summary, we produced a new type of living cyborg cells functionalised with three different types of CDs using epichlorohydrin as a cross-linking agent without affecting the cells viability. We used the formation of host-guest inclusion complex of Methyl Orange dye with the grafted CD to estimate the average number of CD molecules cross-linked per cell which can reach up to tens of millions. In addition, we employed the Layer-by-Layer deposition to functionalise living yeast cells with cationic polyelectrolyte and β -CD sulphate. Living cells functionalised with CDs may find many potential applications as they can be loaded with drugs, immunosuppressants and other molecules forming inclusion complexes with their cyclodextrin interface. Therefore, we foresee such cells being used as novel selective biosorbents in polluted waters, whole cell biosensors, drug delivery, cell therapy and cell implants applications. The cyclodextrin functionalized cells are not limited to having drugs on their surfaces. Such cells can also be used as biosorbents¹⁴ in polluted waters where they use cyclodextrins on their surfaces to extract the pollutants molecules of the right size from water depending on the cyclodextrin used. All other potential applications of these cells are based on utilizing the CDs on the cell surface as encapsulating sites.

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Notes and references

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1. R. F. Fakhrullin, A. I. Zamaleeva, R. T. Minullina, S. A. Konnova and V. N. Paunov, *Chem. Soc. Rev.*, 2012, **41**, 4189-4206.
2. R. F. Fakhrullin, L. V. Shlykova, A. I. Zamaleeva, D. K. Nurgaliev, Y. N. Osin, J. García-Alonso and V. N. Paunov, *Macromolecular Bioscience*, 2010, **10**, 1257-1264.
3. A. I. Zamaleeva, A.V. Porfireva, G. A. Evtugyn and R.F. Fakhrullin, *Langmuir*, 2010, **26**, 2671-2679.
4. R. Kempaiah, S. Salgado, W. L. Chung and V. Maheshwari, *Chem. Commun.*, 2011, **47**, 11480.

5. R. T. Minullina, Y. N. Osin, D. G. Ishmuchametova and R. F. Fakhrullin, *Langmuir*, 2011, **27**, 7708-7713.
6. R. F. Fakhrullin, A. I. Zamaleeva, M. V. Morozov, D. I. Tazetdinova, F. K. Alimova, A. K. Hilmutdinov, R. I. Zhdanov, M. Kahraman and M. Culha, *Langmuir*, 2009, **25**, 4628-4634.
7. J. García-Alonso, R. F. Fakhrullin and V. N. Paunov, *Biosens. Bioelectron.*, 2010, **25**, 1816-1819.
8. M. R. Dзамukova, A. I. Zamaleeva, D. G. Ishmuchametova, Y. N. Osin, A. P. Kiyasov, D. K. Nurgaliev, O. N. Ilinskaya and R. F. Fakhrullin, *Langmuir*, 2011, **27**, 14386-14393.
9. R. F. Fakhrullin and R. T. Minullina, *Langmuir*, 2009, **25**, 6617-6621.
10. S. H. Yang, E. H. Ko and I. S. Choi, *Langmuir*, 2012, **28**, 2151-2155.
11. J. García-Alonso, R. F. Fakhrullin, V. N. Paunov, Z. Shen, J. D. Hardege, N. Pamme, S. J. Haswell and G. M. Greenway, *Analytical and Bioanalytical Chemistry*, 2010, **400**, 1009-1013.
12. A. L. Hillberg, *Biomacromolecules*, 2006, **7**, 2742-2750.
13. D. Park, Y.-S. Yun and J. M. Park, *Biotechnology and Bioprocess Engineering*, 2010, **15**, 86-102.
14. J. Wang and C. Chen, *Biotechnol. Adv.*, 2009, **27**, 195-226.
15. I. Safarik, K. Pospiskova, K. Horska and M. Safarikova, *Soft Matter*, 2012, **8**, 5407-5413.
16. A. Biber, G. Antranikian and E. Heinze, *Appl. Microbiol. Biotechnol.*, 2002, **59**, 609-617.
17. J. Szejtli, *Chem. Rev.*, 1998, **98**, 1743-1754.
18. D. Landy, I. Mallard, A. Ponchel, E. Monflier and S. Fourmentin, *Environ Chem Lett*, 2012, **10**, 225-237.
19. T. V. Frank van de Manakker, C. F. van Nostrum, and W. E. Hennink, *Biomacromolecules*, 2009, **10**, 3157-3175.
20. M. Singh, R. Sharma and U. C. Banerjee, *Biotechnol. Adv.*, 2002, **20**, 341-359.
21. M. E. Brewster and T. Loftsson, *Advanced Drug Delivery Reviews*, 2007, **59**, 645-666.
22. R. Challa, A. Ahuja, J. Ali and R. Khar, *AAPS PharmSciTech*, 2005, **6**, E329-E357.
23. A. Vyas, S. Saraf and S. Saraf, *Journal of Inclusion Phenomena and Macrocyclic Chemistry*, 2008, **62**, 23-42.
24. R. Arun, *Scientia Pharmaceutica*, 2008, **76**, 567-598.
25. D. Duchêne, G. Ponchel and D. Wouessidjewe, *Advanced Drug Delivery Reviews*, 1999, **36**, 29-40.
26. L. Szenté and J. Szejtli, *Trends in Food Science & Technology*, 2004, **15**, 137-142.
27. G. Astray, C. Gonzalez-Barreiro, J. C. Mejuto, R. Rial-Otero and J. Simal-Gandara, *Food Hydrocolloids*, 2009, **23**, 1631-1640.
28. E. M. M. Del Valle, *Process Biochem.*, 2004, **39**, 1033-1046.
29. H. M. C. Marques, *Flavour and Fragrance Journal*, 2010, **25**, 313-326.
30. J. Szejtli, *J. Mater. Chem.*, 1997, **7**, 575-587.
31. Please note that FdA in this paper stands for Fluorescein Diacetate and should not be confused with the abbreviation for the US Food and Drug Administration which is also used in the literature.
32. E. Renard, A. Deratani, G. Volet and B. Sebillé, *Eur. Polym. J.*, 1997, **33**, 49-57.
33. S. Hamai and M. Handa, *Anal. Lett.*, 1999, **32**, 1037-1047.
34. W. H. Lawrence, M. Malik, J. E. Turner and J. Autian, *J. Pharm. Sci.*, 1972, **61**, 1712-1717.

-
35. A. J. Priya, S. P. Vijayalakshmi and A. M. Raichur, *J. Agric. Food. Chem.*, 2011, **59**, 11838-11845.