

The Role of Lithosphere Thickness in the Formation of Ocean Islands and Seamounts: Contrasts between the Louisville and Emperor–Hawaiian Hotspot Trails

J. Godfrey Fitton¹*, Rebecca Williams², Tiffany L. Barry³ and Andrew D. Saunders³

¹School of GeoSciences, University of Edinburgh, Grant Institute, James Hutton Road, Edinburgh EH9 3FE, UK; ²Department of Geography, Geology and Environment, University of Hull, Cottingham Road, Hull HU6 7RX, UK; ³School of Geography, Geology and the Environment, University of Leicester, University Road, Leicester LE1 7RH, UK

*Corresponding author. Telephone: +44 0131 650 8529. E-mail: godfrey.fitton@ed.ac.uk

Received 22 May 2020; Accepted 5 December 2020

ABSTRACT

The Hawaii-Emperor and Louisville seamounts form the two most prominent time-progressive hotspot trails on Earth. Both formed over a similar time interval on lithosphere with a similar range of ages and thickness. The Hawaii-Emperor seamounts are large and magma productivity appears to be increasing at present. The Louisville seamounts, by contrast, are smaller and the trail appears to be waning. We present new major- and trace-element data from five of the older (74-50 Ma) Louisville seamounts drilled during Integrated Ocean Drilling Program (IODP) Expedition 330 and compare these with published data from the Emperor seamounts of the same age. Despite drilling deep into the shield-forming volcanic rocks at three of the Louisville seamounts, our data confirm the results of earlier studies based on dredge samples that the Louisville seamounts are composed of remarkably uniform alkali basalt. The basalt composition can be modelled by ~1.5-3% partial melting of a dominantly garnet-lherzolite mantle with a composition similar to that of the Ontong Java Plateau mantle source. Rock samples recovered by dredging and drilling on the Emperor seamounts range in composition from tholeiitic to alkali basalt and require larger degrees of melting (2-10%) and spinel- to garnet-lherzolite mantle sources. We use a simple decompression melting model to show that melting of mantle with a potential temperature of 1500 °C under lithosphere of varying thickness can account for the composition of the shield-forming tholeiitic basalts from the Emperor seamounts, whereas post-shield alkali basalt requires a lower temperature (1300-1400 °C). This is consistent with the derivation of Hawaii-Emperor shield-forming magmas from the hotter axis of a mantle plume and the post-shield magmas from the cooler plume sheath as the seamount drifts away from the plume axis. The composition of basalt from the Louisville seamounts shows no significant variation with lithosphere thickness at the time of seamount formation, contrary to the predictions of our decompression melting model. This lack of influence of lithospheric thickness is characteristic of basalt from most ocean islands. The problem can be resolved if the Louisville seamounts were formed by dehydration melting of mantle containing a small amount of water in a cooler plume. Hydrous melting in a relatively cool mantle plume (T_p = 1350–1400 °C) could produce a small amount of melt and then be inhibited by increasing viscosity from reaching the dry mantle solidus and melting further. The failure of the plume to reach the dry mantle solidus or the base of the lithosphere means that the resulting magmas would have the same composition irrespective of lithosphere thickness. A hotter mantle plume ($T_p \approx 1500$ °C) beneath the Emperor seamounts and the Hawaiian Islands would have lower viscosity before the

onset of melting, melt to a larger extent, and decompress to the base of the lithosphere. Thus our decompression melting model could potentially explain the composition of both the Emperor and Louisville seamounts. The absence of a significant lithospheric control on the composition of basalt from nearly all ocean islands suggests that dehydration melting is the rule and the Hawaiian islands are the exception. Alternatively, many ocean islands may not be the product of mantle plumes but may instead be formed by decompression melting of heterogeneous mantle sources composed of peridotite containing discrete bodies of carbonated and silica-oversaturated eclogite within the general upper mantle convective flow.

Key words: basalt; decompression melting; mantle plumes; Pacific Ocean; seamounts

INTRODUCTION

The idea that intraplate hotspots are caused by convective plumes originating in the deep mantle was proposed by Morgan (1971) to explain long, time-progressive chains of ocean islands and seamounts such as the Hawaii–Emperor seamount trail (HEST, Fig. 1). The Louisville seamount trail (LST, Fig. 1) in the SW Pacific Ocean is another example and was the subject of Integrated Ocean Drilling Program (IODP) Expedition 330. It has been suggested (e.g. Mahoney & Spencer, 1991) that the LST and the 122 Ma Ontong Java Plateau (OJP) represent, respectively, the tail and head of the same mantle plume.

The primary objective of IODP Expedition 330 was to determine whether the Louisville mantle plume had remained fixed with respect to the Earth's spin axis or whether it had drifted like the Hawaiian plume. Palaeomagnetic studies on drill core recovered from the Emperor seamounts during Ocean Drilling Program (ODP) Leg 197 showed an ~15° southward movement of the Hawaiian hotspot between 81 and 47 Ma (Tarduno et al., 2003). Five of the Louisville seamounts with similar ages were drilled during IODP Expedition 330 (Fig. 2a), and palaeomagnetic data show that there has been little or no motion relative to the Earth's spin axis for the Louisville hotspot during this time (Koppers et al., 2012a).

We present major- and trace-element data on samples recovered from the LST during IODP Expedition 330 and compare them with published data from the HEST (Regelous et al., 2003; Huang et al., 2005). The two suites of samples are of comparable age (74-50 Ma for LST; 85-47 Ma for HEST) and were erupted onto lithosphere with apparently similar age ranges (~13-76 Ma for LST; \sim 17-84 Ma for HEST). The ages of the seamounts and their underlying seafloor are shown in Fig. 2a and b. If the two suites of rocks represent basaltic magma generated by decompression melting in mantle plumes of similar composition, then systematic differences between them should reflect differences in the potential temperature (T_n) of their respective mantle plumes. The objective of the present study is to assess the extent to which decompression melting in plumes with different T_p can account for the difference in composition between the two seamount trails.

THE LOUISVILLE SEAMOUNT TRAIL

The Louisville seamount trail extends 4300 km southeastward from the Osbourn Seamount (Fig. 2a) and comprises at least 75 individual volcanoes, 50 of them more than 3km high (Lonsdale, 1988). Their age decreases from ~77 Ma (Koppers et al., 2004) at the NW end of the chain, where the seamount trail is being subducted beneath the Kermadec trench (Ballance et al., 1989). An extrapolation of the LST suggests that the present position of the hotspot is located near the intersection of the Pacific-Antarctic Ridge and the Eltanin Fracture Zone (Lonsdale, 1988; Watts et al., 1988; Koppers et al., 2012b) although the last magmatism associated with the hotspot is thought to have occurred 1.1 Myr ago (Koppers et al., 2004). Forty of the seamounts have flattened tops and form coral-free guyots, but none of the more recent (<12 Ma) seamounts reached sea level. Buchs et al. (2018) have shown that the flat tops of the LST guyots formed as shallowmarine shelves comprising lava-fed volcaniclastic deltas, unlike the HEST guyots whose flattened tops formed by erosion and truncation of subaerial volcanic edifices. The difference in mode of formation reflects the smaller size and lower rate of magma eruption in the LST. The LST appears to be in decline (Lonsdale, 1988) unlike the HEST, where magmatic productivity has increased considerably over the past 20 Myr (Wessel, 2016). King & Adam (2014) estimated that the HEST mantle buoyancy flux is nearly an order of magnitude higher than that for the LST. The two seamount trails record similar relative motions between heat source (plume) and the overlying plate during the periods of interest: $\sim 6.5 \, \text{cm a}^{-1}$ from 85 to 42 Ma for the HEST, and \sim 7.5 cm a⁻¹ from 74 to 50 Ma for the LST.

The LST is divided into two segments by the Wishbone Scarps (Fig. 2a): at $\sim 163^{\circ}$ W by the East Wishbone Scarp, which is interpreted to be a remanent transform fault, and at 167° W by the West Wishbone Scarp, which may be a former intra-oceanic arc, a fracture zone or a spreading centre (Luyendyk, 1995; Billen & Stock, 2000; Sutherland & Hollis, 2001; Mortimer *et al.*, 2006). To the SE of the scarps, the seamounts are smaller, more widely spaced and built on seafloor that was formed at ~ 50 –40 Ma (Lonsdale, 1988; Watts *et al.*, 1988; Lyons *et al.*, 2000). To the NW of the scarps,

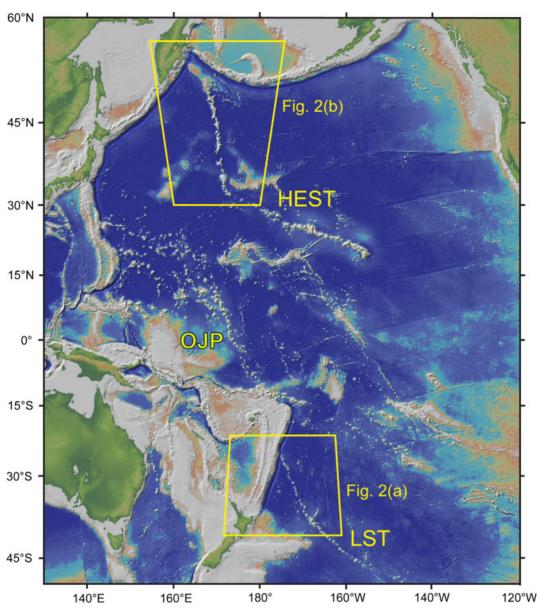


Fig. 1. Map of the Pacific Ocean showing the location of the Hawaii–Emperor seamount trail (HEST), the Louisville seamount trail (LST) and the Ontong Java Plateau (OJP). The two boxes show the locations of Fig. 2a and b. Figure made with GeoMapApp (www. geomapapp.org).

crustal ages are less well defined, and the boundaries between crustal areas created by different spreading centres are not well known (Billen & Stock, 2000). At the NW end of the chain, the Osbourn Trough crosses the LST close to Osbourn Seamount (25·9°S, 175·1°W; Fig. 2a). This trough is thought to be an ancient spreading centre that became extinct at 82–71 Ma (Billen & Stock, 2000), 93–86 Ma (Worthington et al., 2006; Downey et al., 2007), or ~115 Ma (Mortimer et al., 2006). Ferrobasalt recovered ~250 km north of the eastern termination of the Osbourn Trough at Deep Sea Drilling Project (DSDP) Sites 595 and 596 (Fig. 2a) yielded a minimum age of 100 Ma (Montgomery & Johnson, 1987), and Sutherland & Hollis (2001) estimated a crustal age at these sites of 132–144 Ma based on the

biostratigraphy of the deepest sediments cored. DSDP Site 204 lies less than 50 km north of the western part of the Osbourn Trough (Fig. 2a) but drilling at this site failed to reach basement and the cores yielded no reliable biostratigraphic evidence from which its age can be estimated (Billen & Stock, 2000). The seafloor age contours shown in Fig. 2a are based on Müller et al. (2008), who assumed that spreading on the Osbourn Trough ceased at $\sim 86\,\mathrm{Ma}$.

One of the objectives of Expedition 330 was to recover samples of relatively unaltered basalt for geochemical and isotopic analysis. The LST has been extensively dredged, and work on the dredge samples suggests that the volcanic rocks are predominantly alkaline in composition and remarkably homogeneous

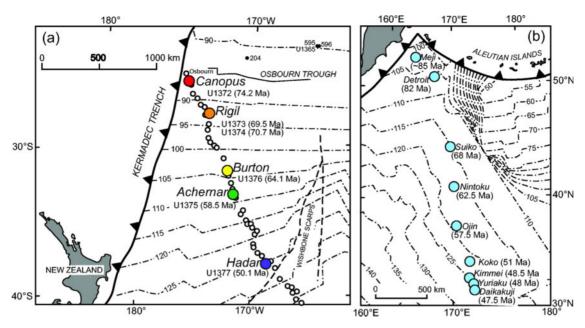


Fig. 2. (a) Map of the northwestern part of the Louisville seamount trail. Circles show the location of individual seamounts, with those drilled during IODP Expedition 330 highlighted in colour. The same colour coding is used for data from these seamounts in Figs 4, 5, 7 and 8. U1372–1377 are the drill sites; seamount ages from Koppers et al. (2012a). The locations of previous DSDP and IODP drill sites north of Osbourn Trough are also shown. (b) Map showing the location of the Emperor seamounts from which the data (from Regelous et al., 2003; Huang et al., 2005) used in this study were obtained. Seamount ages are from O'Connor et al. (2013). Ocean floor isochrons on both maps are based on Müller et al. (2008) and on the Agegrid data for ocean lithosphere from EarthByte.org.

through the 80 Myr lifespan of the chain (Cheng et al., 1987; Hawkins et al., 1987; Beier et al., 2011; Vanderkluysen, 2014). Dredge samples, however, allow only a rather limited assessment of the chemical and isotopic composition of the seamounts for two reasons. First, dredging can only sample the rocks exposed on the surface and these might represent only a late-stage, post-shield phase in the development of the seamounts. Second, dredge samples tend to be altered by long exposure to seawater. Drilling at all of the Expedition 330 sites recovered both fresh and variably altered basalt samples, and three of the sites (U1372, U1374 and U1376) were drilled deep enough to have sampled the shield-forming phase of volcanism, as was similarly achieved at sites along the HEST (Fig. 2b). Thus the chemical composition of samples recovered during Expedition 330 can be compared directly with the data from the HEST reported by Regelous et al. (2003) and Huang et al. (2005).

IODP Expedition 330 drilled five seamounts, spread out along an ~2000 km section of the older part of the LST (Fig. 2a). Only one of these (Burton) had previously been named, and the other four were given informal names by the Expedition 330 shipboard scientists after the four brightest southern hemisphere stars: Canopus, Rigil (from Rigil Kentauri, or Alpha Centauri), Achernar and Hadar. Six sites were drilled on the five seamounts: Canopus (U1372), Rigil (U1373, U1374), Burton (U1376), Achernar (U1375), and Hadar (U1377). In total, 1114 m of sedimentary cover and igneous basement were cored and 806 m were recovered, with drilling reaching,

respectively, to depths of 232.9, 65.7, 522.0, 182.8, 11.5, and 53.3 m below sea floor (mbsf). Retrieved core at each drill site was systematically logged, described and interpreted by shipboard scientists during Expedition 330. Scanned images and descriptions of the core recovered at each site, together with detailed interpretation of the stratigraphic sequences and lithologies, have been published in the Expedition 330 report (Koppers et al., 2012b). A summary of the stratigraphy at three of the sites, based on shipboard studies and subsequent shore-based work and the interpretation of downhole logging, is given in Fig. 3. The igneous basement recovered during IODP Expedition 330 is composed mostly of lava flows, monolithic volcaniclastic breccias, hyaloclastite and minor sheet intrusions. Because the flat tops of the guyots formed during the construction of the Louisville seamounts (Buchs et al., 2018) we can be confident that most of our rock samples represent the shield-forming stage. Only the latestage tuffs and breccias recovered from Burton, which are separated by a limestone unit from the main volcanic sequence (Fig. 3), could represent a post-shield phase of volcanism. A detailed description of the stratigraphy and an interpretation of the lithofacies at each of the drill sites have been published by Buchs et al. (2018).

PETROGRAPHY AND MINERALOGY

Images and descriptions of samples used for shipboard thin section analysis have been published in the Expedition 330 report (Koppers et al., 2012b). Here, we

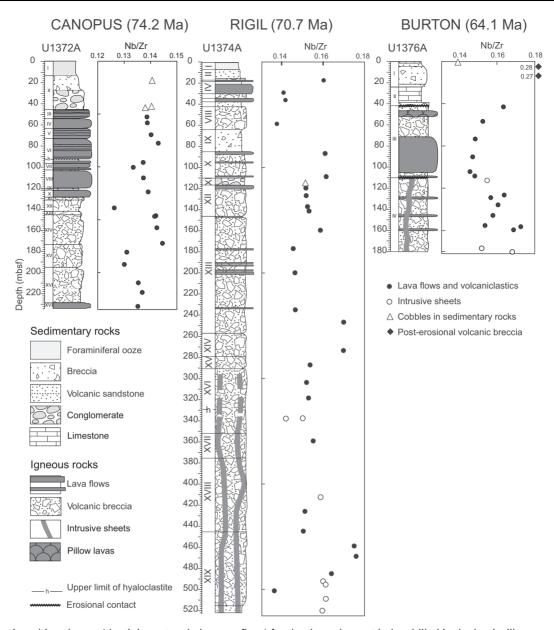


Fig. 3. Stratigraphic columns (depth in metres below sea floor) for the three deepest holes drilled in the Louisville seamounts during IODP Expedition 330 (based on detailed logs given by Koppers et al., 2012b). Data points show the location of the samples used in this study and the downhole variation in Nb/Zr.

summarize briefly the petrography based on studies carried out by members of the shipboard party (including J.G.F. and R.W.) and subsequent work by Dorais (2015) and Dorais & Buchs (2019).

The basalts range from aphyric to highly phyric with phenocryst assemblages commonly of olivine, plagioclase, olivine + clinopyroxene, plagioclase + clinopyroxene and olivine + clinopyroxene + plagioclase. Clinopyroxenes are often zoned, and are largely unaltered. Electron probe analysis shows that most of the pyroxenes are Mg-rich diopside, with some rare wollastonite-rich augite grains (Dorais, 2015). Several clinopyroxene antecrysts were found at Rigil (U1373A-7R1 83–85), the product of incorporation of earlier-formed crystals (e.g. cumulates) into later magmas.

Three distinct types of clinopyroxene in post-shield (rejuvenated phase) volcaniclastic rocks on Burton are interpreted as, respectively, phenocrysts and disaggregated crystals from cumulates and mantle xenoliths (Dorais & Buchs, 2019). Olivine is variably altered to iddingsite although unaltered crystals were frequently found (Tejada et al., 2015). Some olivines show strain lamellae, also suggestive of the incorporation of fragments from cumulates (Dorais, 2015). Plagioclase crystals and volcanic glass in the recovered core varied from fresh (Nichols et al., 2014) to heavily altered to clay. Zeolite- and calcite-filled amygdales are common throughout the core. The only medium-grained rock recovered during the expedition was a holocrystalline, moderately porphyritic dolerite cored in Hole U1375B.

Phenocryst phases and their relative proportions in each of the samples used in this study are summarized in Table 1.

ANALYTICAL TECHNIQUES

Ninety-one rock samples were selected from the visibly least altered sections of the drill core. Forty-two of the 91 samples studied are from lava flows, 31 are volcanic breccia, and 12 are from minor intrusions. The rest of the samples (six) are clasts in sedimentary units. One sample of volcaniclastic sediment from Burton (1376-03) consisting of relatively unaltered basaltic clasts cemented by zeolite was included despite having a 22 wt% loss on ignition (LOI; Table 1). Rock types for each of the samples used in this study are given in Table 1. The samples were crushed, hand-picked to remove secondary minerals and altered fragments where possible, and then ground in an agate-lined Tema mill. X-ray fluorescence (XRF) analyses were carried out on all samples at the University of Edinburgh using techniques similar to those described by Fitton et al. (1998), with modifications noted by Fitton & Godard (2004). Typical values for XRF precision have been given by Fitton et al. (1998). Major-element concentrations were determined after fusion with a lithium borate flux containing La₂O₃ as a heavy absorber, by a method similar to that described by Norrish & Hutton (1969). Traceelement concentrations were determined on pressedpowder samples. The fused and pressed samples were analyzed using a PANalytical PW 2404 automatic X-ray fluorescence spectrometer with a Rh-anode X-ray tube. Rare earth and other trace elements were determined by inductively coupled plasma mass spectrometry (ICP-MS) on a subset of 48 samples using an Agilent 7500 s system at the Open University, with a precision of better than 3%. Samples (including basalt standards BR and W-2 for reference) were prepared by dissolution of 0.1 g of each whole-rock powder with HF and HNO₃, and final dilution to 100 ml with 2% HNO3, following methods described in detail by Rogers et al. (2006). Drift and background were monitored using acid blanks and international reference standards BR and W-2. Analytical data for the rock samples are given in Table 1 and data for international geochemical standards analysed at the same time are given in Tables 2 and 3.

COMPOSITION OF ROCK SAMPLES FROM THE LOUISVILLE SEAMOUNT TRAIL

A total alkali–silica (TAS) diagram (Fig. 4) is used to compare the Louisville data given here with data from the Emperor seamounts (Regelous *et al.*, 2003; Huang *et al.*, 2005). Rock samples that are likely to be more altered (LOI > 2 wt%) are excluded from this plot. It is clear from this diagram that most of the LST samples are alkaline whereas more than a third of those from the HEST are tholeiitic. Two samples from Canopus and one from Burton plot clearly in the tholeiitic field, but

the rest of the Louisville samples are transitional to alkaline in composition. However, a suite of glass samples covering the same depth range as our samples plot entirely in the alkaline field (Nichols *et al.*, 2014). These data are shown as a grey field in Fig. 4. It is clear that alkali loss has affected some of our least altered basalt samples and that the TAS diagram should be used with caution when classifying volcanic rocks.

Three of the Louisville seamounts, Canopus (Site U1372), Rigil (Site U1374) and Burton (Site U1376), were drilled deep enough (232.9, 522.0 and 182.8 mbsf respectively) to look for changes of composition with depth. We use Nb/Zr plotted against depth (Fig. 3) to assess this because Nb/Zr is sensitive to degree and depth of melting but insensitive to the effects of low-pressure fractional crystallization. As can be seen in Fig. 3 there is little systematic stratigraphic variation in Nb/Zr in the drill core from Rigil and Burton seamounts, but the volcanic rocks cored at Canopus (Site U1372) show three cycles of upward-decreasing Nb/Zr. This could be interpreted as showing cycles of increasing degree of mantle melting but with no obvious overall trend with time. The general lack of systematic compositional variation with depth in the IODP Expedition 330 drill holes was also noted by Nichols et al. (2014). Dorais (2015) determined the rare earth element (REE) concentrations in clinopyroxene phenocrysts in basalt from the IODP Expedition 330 drill core and used these to calculate the composition of the liquid from which they crystallized. In most cases these match the bulk composition of glasses analysed by Nichols et al. (2014) and indicate an alkaline parentage. The cores of some clinopyroxene phenocrysts from Rigil seamount, however, have distinct compositions suggesting that they originated as antecrysts derived from a transitional magma, but Dorais (2015) found no evidence for the presence of tholeiitic magmas. If the Louisville seamounts have a tholeiitic basement then this must lie below the level sampled by the drill cores.

Incompatible-element concentrations, normalized to primitive mantle (PM) values, for samples from the Louisville and Emperor seamounts are shown in Fig. 5. Data from primitive olivine tholeiitic basalt (Kroenketype basalt) from the Ontong Java Plateau (OJP) are shown for comparison. Several features are obvious in Fig. 5. (1) The patterns for all five Louisville seamounts are remarkably similar to each other and to those for the post-shield basalt samples from the Emperor seamounts. (2) The Emperor shield lavas have similar patterns to those from the Louisville seamounts, but with more gentle negative slopes consistent with the former representing larger degrees of melting. (3) The Louisville seamounts show only a very weak trend towards overall enrichment with decreasing age from Canopus to Hadar. This remarkable homogeneity has previously been noted among dredge samples from the LST by Beier et al. (2011) and Vanderkluysen et al. (2014). (4) The Louisville and Emperor patterns are virtually linear, with relative depletion in only the most

Table 1: Major- and trace-element data for volcanic rocks from the Louisville seamounts recovered during IODP Expedition 330

· —						<u> </u>	· .	
Seamount:	Canopus	Canopus	Canopus	Canopus	Canopus	Canopus	Canopus	Canopus
Sample no.:	1372-1	1372-2	1372-3	1372-4	1372-5	1372-6	1372-7	1372-8
Site, hole:	U1372A	U1372A	U1372A	U1372A	U1372A	U1372A	U1372A	U1372A
· ·								
Core-section:	5R-1	8R-1	8R-2	9R-1	9R-6	10R-7	11R-6	13R-5
Interval (cm):	66–71	126–131	117–123	103–108	12–17	6–11	38–43	26–31
Depth (mbsf):	18-69	43.39	44.66	52.76	58.39	69-60	77.65	95.83
Rock type:	C in sed	C in sed	C in sed	L	L	L	L	L
Phenocrysts:	M-OI	M-OI	Aphyric	S-OI	H-OI	M-OI	Aphyric	Aphyric
XRF (wt %)								
SiO ₂	47.22	44.64	46.47	39.82	44.24	45.67	46.66	47.23
010 ₂	15.34	12.75	15.24	15.16	11.33	15·43	15.09	14.54
Al ₂ O ₃								
Fe ₂ O ₃ ^T	11.85	12.79	12.55	15.01	13.61	12.85	12.78	12.21
MgO	6.12	13⋅68	5⋅80	4.03	16.29	3.94	6.57	5.02
CaO	11.66	8.50	10.98	12.28	8.27	10∙96	11.40	11.24
Na ₂ O	2.77	2⋅07	2.93	2.59	1.82	2.74	2.68	3.07
K ₂ O	0.820	0.610	0.922	1.509	0.644	1.976	0.776	1.080
TiO ₂	2.690	2.325	3.266	3.069	2.325	3.040	2.877	3.018
MnŌ	0.144	0.168	0.162	0.172	0.176	0.127	0.169	0.205
P_2O_5	0.395	0.328	0.444	0.462	0.299	0.492	0.419	0.423
LOI	1.09	1.96	1.08	5.83	0.96	2.46	0.69	1.78
Total	100.10	99.82	99.84	99.93	99.96	99.69	100-11	99.82
XRF (ppm)	100 10	00 02	00 04	00 00	00 00	00 00	100 11	00 02
Zr	213.2	183.3	247.8	226-6	174-2	248-4	225.6	236.5
Nb	30.0	25.7	34.2	31.4	24.2	34.8	32.2	32·5
Y	28·4	23·7 23·7	31·0	28.8	22·7	31.7	30·0	30·5
Sr	483.8	361·9	557·8	540·1	387.3	513.7	485.5	467·7
Rb	14.4	10.3	15.4	16.7	11.0	24.5	11.4	18.7
Zn	98.1	89.4	100.6	121.5	94.6	112.5	93.9	105.4
Cu	66	42.2	78.3	63.2	63.6	72 ⋅1	86.1	70⋅1
Ni	74	376⋅9	60.7	293.8	512.9	82	72·9	107⋅3
Cr	130⋅8	779.2	111.1	504.3	803.9	110∙9	94.5	137.7
V	287.3	230.8	289.7	341	211.2	410⋅3	317⋅1	313⋅5
Ba	154.7	122⋅9	180∙4	174	120.4	195∙6	176⋅2	161⋅6
Sc	32⋅3	23.9	28.2	34	22.6	35⋅7	30⋅2	30⋅8
ICP-MS (ppm)								
Υ	28.30		32.50		23.60	34.05	30.63	
Zr	217.62		265.42		187.30	272.69	240.38	
Nb	28.26		34.56		24.29	36.20	32.69	
La	23.06		26.95		18.95	28.98	25.73	
Ce	53.77		64.24		45.45	67.72	60.45	
Pr	7.03		8.43		5.95	8.87	7.89	
Nd	30.46		36.44		25.73	38.01	33.64	
Sm	7.03		8.48		6.00	8.75	7.65	
Eu	2.27		2.71		1.91	2.82	2·51	
Gd	6·87		8·15		5.84	8.37	7·39	
Tb	1.03		1.21		0.87	1.23	1·11	
Dy	5.51		6.33		4·59	6.63	5.93	
Ho	1.06		1.19		0.87	1.25	1.13	
_	2.71		2.97		2.19	3.21	2.92	
Er Yb	2.09		2.25		1.70	2.46	2.26	
Lu	0.29		0.31		0.24	0.34	0.31	
Hf	4·97		5.99		4.28	6·14	5.32	
Ta	1.81		2.16		1.54	2.13	2.04	
Th	2.33		2.72		1.93	2.90	2.47	
Coomount	Concesso	Concess	Concess	Concess	Concerns	Concesso	Concesso	Cononii
Seamount:	Canopus	Canopus	Canopus	Canopus	Canopus	Canopus	Canopus	Canopus
Sample no.:	1372-9	1372-10	1372-11	1372-12	1372-13	1372-14	1372-15	1372-16
Site, hole:	U1372A	U1372A	U1372A	U1372A	U1372A	U1372A	U1372A	U1372A
Core-section:	14R-1	15R-1	16R-4	19R-1	21R-1	21R-2	23R-2	26R-2
Interval (cm):	71–78	118–123	30–35	64–69	124–130	44–50	5–11	44–50
Depth (mbsf):	100-45	110.51	123.73	138-87	146-27	146-92	157-66	172.36
Rock type:	L	L	L	НВ	C in B	C in B	C in B	C in B
Phenocrysts:	M-OI, S-PI	Aphyric	Aphyric	Aphyric	S-PI	S-PI	M-PI	Aphyric
XRF (wt %)								
SiO ₂	45.27	47.15	47.49	48-94	47.83	47.29	46.19	50.35
	12·88	14·64	15·19	15·55	47·63 15·85	47·29 15·73	15.31	16⋅39
Al ₂ O ₃								
$Fe_2O_3^T$	12·78	12·89	11·04	11·47	11·46	11.92	13·11	8·79
MgO	8.97	6·35	5·53	7.57	3·15	3.84 10.27	4·54	3·76
CaO	10.80	11·14	11.80	3.91	10·55	10·27	9.93 3.10	10.25
Na ₂ O	2.42	2.81	2.86	2.86	3.53	3.39	3⋅19	3.59

Table 1: Continued

Seamount:	Canopus	Canopus	Canopus	Canopus	Canopus	Canopus	Canopus	Canopus
Sample no.:	1372-9	1372-10	1372-11	1372-12	1372-13	1372-14	1372-15	1372-16
Site, hole:	U1372A	U1372A	U1372A	U1372A	U1372A	U1372A	U1372A	U1372A
Core-section:	14R-1	15R-1	16R-4	19R-1	21R-1	21R-2	23R-2	26R-2
Interval (cm):	71–78	118–123	30–35	64–69	124–130	44–50	5–11	44–50
Depth (mbsf):	100·45	110·51	123·73	138-87	146-27	146.92	157.66	172·36
Rock type:	L	L	L	HB	C in B	C in B	C in B	C in B
Phenocrysts:	M-OI, S-PI	Aphyric	Aphyric	Aphyric	S-PI	S-PI	M-PI	Aphyric
K_2O TiO_2 MnO P_2O_5 LOI $Total$ $XRF (ppm)$	0.739	0.827	0.855	0.829	1·139	0.934	0.985	1·150
	2.525	2.819	2.935	2.503	3·422	3.397	3.479	3·723
	0.167	0.171	0.162	0.106	0·170	0.200	0.214	0·132
	0.348	0.398	0.420	0.197	0·528	0.513	0.490	0·538
	2.75	0.65	1.21	5.78	1·84	2.12	2.41	1·10
	99.65	99.85	99.49	99.72	99·46	99.59	99.84	99·77
Zr	192·8	220.5	231.8	178.4	290·4	287·7	275.1	288.9
Nb	25·7	30.3	32.2	22.5	41·2	40·7	39.2	41.8
Y	25·9	29.6	30.5	17.1	37·7	37·1	35.9	38.6
Sr	416·3	517.2	528.1	244.5	540·7	557·1	503.7	519.0
Rb	12·1	14.5	12.8	7.1	17·6	11·5	12.9	16.4
Zn	90	89.6	97.3	96.1	122·7	119·2	112.8	118.8
Cu	48·9	83.1	87.4	78.7	55·3	54·9	54.4	58.8
Ni	291·5	54.6	74.9	63.3	62·9	57·7	34.9	86.5
Cr	465·5	85.9	38.4	130.6	31·1	29·8	50.1	60.0
V	254·6	276.6	314.3	332.4	342	326·2	321.9	375
Ba	134·5	162.8	174.7	63.5	200·6	220	199.6	207.4
Sc	27·9	30.4	30.1	37	28	26·4	25.7	30.7
ICP-MS (ppm) Y Zr Nb La Ce Pr Nd Sm Eu Gd Tb Dy Ho Er Yb Lu Hf Ta Th	25-80 200-70 25-44 20-31 48-64 6-39 27-67 6-41 2-08 6-22 0-93 5-03 0-95 2-45 1-87 0-26 4-54 1-65 2-03		30.84 243.19 31.91 25.58 60.38 7.82 33.55 7.72 2.50 7.46 1.12 5.92 1.14 2.91 2.22 0.31 5.46 2.02 2.60	18-02 197-89 22-73 14-11 34-02 4-67 20-18 4-79 1-60 4-62 0-71 3-73 0-70 1-83 1-37 0-20 4-52 1-64 1-76		38-37 301-49 39-90 31-54 73-06 9-43 40-04 9-22 2-94 8-92 1-34 7-23 1-42 3-61 2-86 0-40 6-69 2-51 3-32		39·17 300·30 40·72 31·78 74·11 9·70 41·20 9·44 3·03 9·14 1·38 7·47 1·45 3·72 2·95 0·41 6·82 2·57 3·33
Seamount: Sample no.: Site, hole: Core-section: Interval (cm): Depth (mbsf): Rock type: Phenocrysts:	Canopus	Canopus	Canopus	Canopus	Canopus	Rigil	Rigil	Rigil
	1372-17	1372-18	1372-19	1372-20	1372-21	1373-1	1373-2	1373-3
	U1372A	U1372A	U1372A	U1372A	U1372A	U1373A	U1373A	U1373A
	28R-1	30R-2	33R-1	35R-1	38R-3	1R-2	2R-3	7R-2
	53-58	105-110	76-81	20–25	86–92	142-148	46-51	68–75
	180-76	192-37	209-79	218-83	232·10	2·25	12-76	34-66
	HB	HB	B	C in B	L	C in B	C in B	L
	Aphyric	Aphyric	M-OI-PI	M-OI	M-OI-PI	H-OI	M-OI	H-Ol-Cp
XRF ($wt\%$) SiO_2 Al_2O_3 Fe_2O_3 MgO CaO Na_2O K_2O TiO_2 MnO P_2O_5 LOI $Total$ XRF (ppm)	49-86	51.74	48·87	49·02	46·26	45·29	43-53	43·02
	14-61	15.19	15·82	15·86	14·96	12·08	15-06	11·31
	12-22	10.70	9·06	8·97	10·78	12·32	13-98	12·61
	6-31	5.71	5·58	5·43	9·09	13·87	5-31	13·31
	5-71	5.05	13·16	13·00	11·83	11·39	10-42	11·86
	2-85	2.85	2·25	2·33	2·10	1·66	2-75	2·17
	0-911	0.996	0·426	0·482	0·540	0·513	1-629	0·716
	2-587	2.709	2·232	2·249	2·098	2·069	3-214	2·264
	0-113	0.078	0·139	0·141	0·143	0·170	0-143	0·181
	0-166	0.156	0·292	0·296	0·272	0·244	0-508	0·359
	4-28	4.45	1·71	1·63	1·31	0·35	3-21	2·21
	99-62	99.63	99·54	99·42	99·38	99·96	99-75	100·01

Table 1: Continu	ed							
Seamount: Sample no.: Site, hole: Core-section: Interval (cm): Depth (mbsf): Rock type: Phenocrysts:	Canopus 1372-17 U1372A 28R-1 53–58 180-76 HB Aphyric	Canopus 1372-18 U1372A 30R-2 105-110 192-37 HB Aphyric	Canopus 1372-19 U1372A 33R-1 76–81 209-79 B M-OI-PI	Canopus 1372-20 U1372A 35R-1 20–25 218-83 C in B M-OI	Canopus 1372-21 U1372A 38R-3 86-92 232·10 L M-OI-PI	Rigil 1373-1 U1373A 1R-2 142-148 2-25 C in B H-Ol	Rigil 1373-2 U1373A 2R-3 46–51 12-76 C in B M-OI	Rigil 1373-3 U1373A 7R-2 68-75 34-66 L H-OI-Cp
Zr Nb Y Sr Rb Zn Cu Ni Cr V Ba Sc ICP-MS (ppm) Y Zr Nb La Ce Pr Nd Sm Eu Gd Tb Dy Ho Er Yb Lu Hf Ta Th	181·4 23·8 17·4 315·0 10·0 92·2 93·4 83·5 364·7 294·9 52·6 37·2	197·0 25·6 14·5 276·1 11·5 84 162 80·9 347·6 342 87·4 41·9	164·1 22·2 23·3 476·9 3·9 104·4 84·2 87·8 532·8 255 104·9 33·9 24·97 178·49 23·91 18·39 43·62 5·74 24·98 5·92 1·97 5·81 0·88 4·80 0·93 2·37 1·85 0·26 4·09 1·46 1·90	166·1 22·7 24·0 482·7 4·3 104 83·8 83·8 544·0 252·6 114 33·7	155-9 21-1 22-3 437-0 8-9 76 79-3 175-6 481-3 232-9 113-5 31-4 23-39 168-60 21-28 17-69 41-35 5-52 23-88 5-63 1.87 5-56 0.84 4-53 0.87 2-23 1-73 0.24 3-90 1-33 1-77	137·5 20·1 20·4 377·2 10·8 79·4 77 325·3 794·1 269·6 101·8 33·2	243.4 41.3 28.2 502.8 20.1 115 44.3 190.1 260.8 347 227.8 28	168·0 30·4 20·7 525·7 13·6 84·5 73 368·7 717·3 234·4 194·1 27·3
Seamount: Sample no.: Site, hole: Core-section: Interval (cm): Depth (mbsf): Rock type: Phenocrysts:	Rigil 1373-4 U1373A 9R-1 71-73 42·82 L Aphryic	Rigil 1373-5 U1373A 11R-3 3-8 54-02 L Aphryic	Rigil 1373-6 U1373A 13R-2 77-83 62-96 L Aphryic	Rigil 1374-1 U1374A 3R-3 11-14 17-41 L H-OI-Cp	Rigil 1374-2 U1374A 5R-4 33-39 29-06 L Aphryic	Rigil 1374-3 U1374A 7R-2 46-51 36-13 L Aphryic	Rigil 1374-4 U1374A 12R-1 20-25 58-63 B M-OI	Rigil 1374-5 U1374A 16R-2 65-70 86-70 L Aphryic
XRF (wt %) SiO ₂ Al ₂ O ₃ Fe ₂ O ₃ MgO CaO Na ₂ O K ₂ O TiO ₂ MnO P ₂ O ₅ LOI Total XRF (ppm) Zr Nb Y Sr Rb Zn	45·09 15·11 13·03 4·94 11·56 2·84 1·244 2·975 0·215 0·450 2·37 99·82 232·4 34·5 32·1 442·7 18·3	46.63 14.83 12.73 5.89 11.49 2.86 0.763 3.012 0.172 0.445 0.68 99.50 234.4 35.6 32.1 467.8 9.1	46·79 15·02 12·70 5·80 11·57 2·83 0·701 3·034 0·163 0·447 0·87 99·93 234·4 35·6 31·8 484·7 6·3 95·8	45-80 14-71 11-70 9-11 9-01 2-78 0-818 2-614 0-181 0-618 2-04 99-38 295-4 47-3 32-0 586-1 8-9 104-4	45.83 15.47 14.86 5.82 8.96 3.12 0.860 3.288 0.210 0.443 1.03 99.89 243.5 34.3 34.9 441.9 14.5 105.8	46·00 15·44 14·36 5·57 8·95 3·16 0·880 3·346 0·191 0·457 1·34 99·69 252·1 35·8 35·6 444·7 13·3 112·2	42-65 16-04 13-87 4-10 11-68 2-60 0-956 2-687 0-173 0-553 4-11 99-42 220-6 30-4 31-6 490-7 15-7 117-2	45.03 17.00 13.57 5.33 9.25 3.08 0.784 3.647 0.183 0.558 1.07 99.50 289.6 46.6 35.0 703.0 8.6 97.3

Table 1: Continue	ed							
Seamount: Sample no.: Site, hole: Core-section: Interval (cm): Depth (mbsf): Rock type: Phenocrysts:	Rigil 1373-4 U1373A 9R-1 71-73 42·82 L Aphryic	Rigil 1373-5 U1373A 11R-3 3-8 54-02 L Aphryic	Rigil 1373-6 U1373A 13R-2 77-83 62·96 L Aphryic	Rigil 1374-1 U1374A 3R-3 11-14 17·41 L H-OI-Cp	Rigil 1374-2 U1374A 5R-4 33–39 29·06 L Aphryic	Rigil 1374-3 U1374A 7R-2 46-51 36·13 L Aphryic	Rigil 1374-4 U1374A 12R-1 20–25 58-63 B M-OI	Rigil 1374-5 U1374A 16R-2 65-70 86-70 L Aphryic
Cu Ni Cr V Ba Sc ICP-MS (ppm)	66 58·4 65·3 347 185 34·5	100.5 60.6 51.5 332 187.1 32.9	100·4 60·5 50·8 330·5 184·7 31·6	41.6 191.7 442.1 204.5 230.5 17.4	37·2 31·4 25·3 340·7 173·9 26·7	36·9 30·1 26·6 356 180·3 28·5	54·6 156·2 400·4 348·7 156·1 35·8	36·8 36·2 12·6 282·2 230·6 20·5
Y Zr Nb La Ce Pr Nd Sm Eu Gd Tb Dy Ho Er Yb Lu Hf Ta Th	31·29 234·25 36·22 24·80 56·68 7·29 31·41 7·09 2·32 6·92 1·03 5·65 1·08 2·82 2·19 0·32 4·85 1·85 2·19	31·03 229·79 35·58 24·55 56·95 7·27 31·35 7·12 2·27 6·93 1·04 5·64 1·08 2·81 2·19 0·32 4·90 1·85 2·24			33.68 235.57 34.34 24.29 55.71 7.33 31.77 7.35 2.36 7.11 1.09 5.96 1.15 2.99 2.40 0.34 5.12 1.80 2.22			33.47 279.70 46.23 31.08 70.87 9.07 38.46 8.35 2.65 7.65 1.15 6.06 1.17 3.01 2.37 0.35 5.80 2.35 2.65
Seamount: Sample no.: Site, hole: Core-section: Interval (cm): Depth (mbsf): Rock type: Phenocrysts:	Rigil 1374-6 U1374A 21R-2 77-82 108-69 L Aphyric	Rigil 1374-7 U1374A 22R-4 68–73 115-72 C in B H-Ol-Cp	Rigil 1374-8 U1374A 23R-3 106-111 119-52 L M-PI-OI-Cp	Rigil 1374-9 U1374A 25R-1 109–112 126-71 C in B H-PI-OI-Cp	Rigil 1374-10 U1374A 27R-2 53-58 137-12 C in B H-PI-OI-Cp	Rigil 1374-11 U1374A 28R-1 132-135 141·34 C in B H-PI-OI-Cp	Rigil 1374-12 U1374A 30R-5 18-25 159-49 L M-OI-PI	Rigil 1374-13 U1374A 32R-3 128-135 176-89 L S-OI
XRF (wt %) SiO ₂ Al ₂ O ₃ Fe ₂ O ₃ MgO CaO Na ₂ O K ₂ O TiO ₂ MnO P ₂ O ₅ LOI Total XRF (ppm)	45·04 16·79 14·29 5·39 9·37 3·08 0·969 3·623 0·183 0·555 0·23 99·52	44.96 14.58 13.15 8.97 9.19 2.60 0.925 2.798 0.179 0.450 1.93 99.73	45·42 14·93 13·00 8·92 10·07 2·59 0·666 2·739 0·179 0·427 1·06 100·00	45.02 15.00 13.02 7.60 10.62 2.66 0.840 2.758 0.192 0.441 1.61 99.77	44·41 15·02 12·75 6·93 11·15 2·64 0·774 2·751 0·205 0·449 2·38 99·45	45·42 15·09 13·09 7·56 10·32 2·68 0·875 2·812 0·190 0·456 1·37 99·86	45·28 14·68 12·53 8·74 10·07 2·73 0·791 2·644 0·204 0·455 1·69 99·81	46·37 15·93 13·04 6·51 8·61 3·42 1·080 2·972 0·197 0·588 0·98 99·70
Zr Nbb Y Sr Rb Zn Cu Ni Cr V Ba	289·3 46·7 34·6 692·5 15·8 98·7 35 34·4 13·9 278·8 214·8 22·4	249·0 37·7 30·6 467·6 16·0 87·2 46·9 173·6 265·1 247·7 170·4 22·7	240·5 36·5 29·9 520·0 10·9 92·7 50·8 154·1 279·2 252·1 175·3 25·4	242·3 36·8 30·0 534·3 14·9 98·9 50·3 165·4 289·7 260·2 157·8 26·3	246·7 37·6 30·0 539·0 13·5 99·3 48·9 156·6 253·6 260·7 156·8 25·3	250.9 38.4 30.8 541.9 16.2 99.5 49.3 154 262.7 263 168.7 25.8	260·5 41·3 31·3 522·0 12·8 95·9 46·4 155·9 316·1 237 184·1 23·3	321-6 46-8 34-8 603-9 15-4 92-7 35-4 78-8 101-0 226-5 227-5 20-8

Table 1: Continue	ed							
Seamount: Sample no.: Site, hole: Core-section: Interval (cm): Depth (mbsf): Rock type: Phenocrysts:	Rigil	Rigil	Rigil	Rigil	Rigil	Rigil	Rigil	Rigil
	1374-6	1374-7	1374-8	1374-9	1374-10	1374-11	1374-12	1374-13
	U1374A	U1374A	U1374A	U1374A	U1374A	U1374A	U1374A	U1374A
	21R-2	22R-4	23R-3	25R-1	27R-2	28R-1	30R-5	32R-3
	77-82	68-73	106-111	109-112	53–58	132-135	18-25	128-135
	108-69	115-72	119-52	126-71	137-12	141-34	159-49	176-89
	L	C in B	L	C in B	C in B	C in B	L	L
	Aphyric	H-OI-Cp	M-PI-OI-Cp	H-Pl-Ol-Cp	H-PI-OI-Cp	H-PI-OI-Cp	M-OI-PI	S-OI
ICP-MS (ppm)	r , -				-			
Y Y Zr Nb La Ce Pr Nd Sm Eu Gd Tb Dy Ho Er Yb Lu Hf Ta Th	33.74 283.61 47.10 31.28 70.91 9.08 38.44 8.39 2.63 7.71 1.13 6.02 1.15 3.00 2.35 0.34 5.73 2.34 2.60						30.67 251.09 40.76 27.42 62.28 7.84 32.94 7.20 2.28 6.77 1.01 5.43 1.05 2.73 2.14 0.31 5.15 2.07 2.57	
Seamount: Sample no.: Site, hole: Core-section: Interval (cm): Depth (mbsf): Rock type: Phenocrysts:	Rigil	Rigil	Rigil	Rigil	Rigil	Rigil	Rigil	Rigil
	1374-14	1374-15	1374-16	1374-17	1374-18	1374-19	1374-20	1374-21
	U1374A	U1374A	U1374A	U1374A	U1374A	U1374A	U1374A	U1374A
	37R-3	41R-4	42R-5	45R-4	46R-7	48R-5	50R-2	52R-2
	39-44	55–58	105-111	58-63	73-76	97-102	31-36	39-45
	199-86	234-92	246-63	273-54	287-18	303-79	318-54	337-69
	L	L	L	C in B	C in B	C in B	L	Int
	S-OI	S-OI	M-PI-Cp	S-PI	M-PI-Cp	M-PI-Cp-OI	M-PI-Cp-OI	Aphyric
XRF (wt%) SiO ₂ Al ₂ O ₃ Fe ₂ O ₃ MgO CaO Na ₂ O K ₂ O TiO ₂ MnO P ₂ O ₅ LOI Total	45·71	46·42	44·89	45·20	46-02	45.75	46·34	45-70
	15·88	15·92	15·72	15·88	14-89	15.69	15·82	14-98
	13·13	13·09	12·68	12·18	13-63	12.09	12·01	13-00
	6·54	7·14	5·85	5·67	5-29	5.55	5·97	5-96
	8·93	8·42	12·08	12·28	11-09	12.70	12·36	8-91
	3·27	3·24	2·60	2·62	2-87	2.32	2·40	3-17
	0·922	0·895	0·816	0·863	0-755	0.712	0·582	1-022
	3·006	3·046	3·290	3·275	3-502	2.727	2·779	3-271
	0·180	0·164	0·180	0·192	0-272	0.185	0·189	0-275
	0·573	0·570	0·525	0·526	0-531	0.393	0·389	0-408
	1·37	1·21	1·13	1·24	0-77	1.57	0·83	2-73
	99·51	100·10	99·76	99·93	99-62	99.69	99·67	99-42
XRF (ppm) Zr Nb Y Sr Rb Zn Cu Ni Cr V Ba Sc ICP-MS (ppm)	312·0	313-0	265·2	261·1	277-7	208-7	210·1	219.5
	45·7	45-9	45·1	44·3	42-7	31-7	32·1	33.0
	34·5	34-7	31·1	30·9	34-6	26-7	27·1	29.8
	609·0	596-9	643·4	637·3	562-9	507-1	514·6	453.8
	11·6	11-3	12·8	14·2	10-7	18-0	12·2	12.7
	95·3	95-9	97·5	98·1	118-9	98	95·3	122.4
	34·1	33-8	81	82·1	50-2	64-1	64·4	104.4
	94	106-7	57·9	62·7	34-8	71-2	73·6	51.9
	120·2	125-5	87·2	98·1	12-6	180-0	175·7	40.2
	236·4	226-3	296·9	304	359-8	295-4	289·9	425.4
	216·7	214-1	214·1	211·5	218	121	139·6	168.5
	21·9	20-5	27	27	29-2	32-3	31·7	35.3
Y Zr Nb La Ce					32.57 264.38 41.87 28.70 66.23		26.88 202.96 32.20 22.29 51.19	

Table 1: Continu	ed							
Seamount: Sample no.: Site, hole: Core-section: Interval (cm): Depth (mbsf): Rock type: Phenocrysts:	Rigil	Rigil	Rigil	Rigil	Rigil	Rigil	Rigil	Rigil
	1374-14	1374-15	1374-16	1374-17	1374-18	1374-19	1374-20	1374-21
	U1374A	U1374A	U1374A	U1374A	U1374A	U1374A	U1374A	U1374A
	37R-3	41R-4	42R-5	45R-4	46R-7	48R-5	50R-2	52R-2
	39-44	55-58	105-111	58–63	73–76	97-102	31-36	39-45
	199·86	234-92	246-63	273-54	287·18	303-79	318-54	337-69
	L	L	L	C in B	C in B	C in B	L	Int
	S-OI	S-OI	M-PI-Cp	S-PI	M-PI-Cp	M-PI-Cp-OI	M-PI-Cp-OI	Aphyric
Pr Nd Sm Eu Gd Tb Dy Ho Er Yb Lu Hf Ta					8·50 36·56 8·12 2·55 7·62 1·13 6·06 1·14 2·92 2·20 0·31 5·65 2·15 2·34		6·56 28·36 6·41 2·07 6·11 0·90 4·85 0·91 2·32 1·74 0·25 4·39 1·79 2·01	
Seamount: Sample no.: Site, hole: Core-section: Interval (cm): Depth (mbsf): Rock type: Phenocrysts:	Rigil	Rigil	Rigil	Rigil	Rigil	Rigil	Rigil	Rigil
	1374-22	1374-23	1374-24	1374-25	1374-26	1374-27	1374-28	1374-29
	U1374A	U1374A	U1374A	U1374A	U1374A	U1374A	U1374A	U1374A
	52R-2	54R-4	60R-1	61R-3	63R-3	64R-6	65R-7	67R-5
	78-83	30–36	16-21	118-123	112–117	101-104	104-110	70-75
	338-08	359·21	412-59	426.00	444-77	458-80	468-60	485-04
	Int	C in HB	Int	C in B	L	C in B	C in B	C in B
	Aphyric	Aphyric	Aphyric	M-Pl	M-PI	Aphyric	Aphyric	Aphyric
XRF (wt %) SiO ₂ Al ₂ O ₃ Fe ₂ O ₃ MgO CaO Na ₂ O K ₂ O TiO ₂ MnO P ₂ O ₅ LOI	45-85	45.65	44.79	45.00	44.35	45·17	45·18	45·43
	14-67	14.76	15.31	16.78	16.37	17·19	17·30	15·15
	12-85	13.99	12.29	12.87	12.48	12·82	12·67	12·24
	5-87	4.83	6.09	5.66	5.83	4·86	4·77	7·17
	10-11	11.53	11.33	12.72	12.46	9·52	9·77	13·14
	3-07	2.64	2.65	2.37	2.36	3·81	3·76	2·37
	0-967	0.812	0.894	0.370	0.317	0·865	0·858	0·461
	3-092	3.323	2.691	2.887	2.787	3·682	3·709	2·864
	0-262	0.206	0.147	0.162	0.160	0·187	0·190	0·162
	0-433	0.491	0.390	0.362	0.355	0·734	0·731	0·429
	2-33	1.27	3.01	0.38	2.50	0·89	1·08	0·48
	99-50	99.50	99.59	99.56	99.97	99·73	100·02	99·90
XRF (ppm) Zr Nb Y Sr Rb Zn Cu Ni Cr V Ba Sc	232-8	259·0	206.3	200·2	198.5	329.6	333·1	222·4
	33-1	40·2	32.8	30·3	29.9	57.7	58·6	36·5
	33-8	32·5	25.9	26·7	26.9	34.5	34·4	25·7
	386-7	526·1	646.1	558·6	556.2	924.8	932·7	566·3
	12-8	24·2	12.9	5·6	3.4	16.2	15·9	9·5
	111-7	122·4	96.5	98·5	95.7	98.1	100·3	96·3
	48-4	57·3	60.1	90·7	91.0	32.4	32·9	115·2
	39-9	46·1	48.5	58·2	58.0	12	12·3	94·5
	26-5	29·1	34.7	57·4	58.6	6.5	4·7	86·3
	363-2	357	286.6	332	330.4	230.1	226·2	295
	172-4	157·5	245.9	143	153.1	293.8	301·8	172
	24-7	29·7	23.9	28·5	28.0	15.8	16·4	31·4
ICP-MS (ppm) Y Zr Nb La Ce Pr Nd Sm Eu	33·13 229·71 33·51 23·08 54·31 7·08 31·09 7·23 2·28	32.82 253.22 40.60 27.18 63.16 8.10 34.80 7.83 2.49	25.61 205.14 33.00 21.61 51.52 6.59 28.38 6.44 2.09			35.86 345.42 59.98 44.32 96.92 12.24 50.34 10.58 3.31		25·50 215·61 36·21 24·41 57·05 7·16 30·72 6·85 2·19

Table 1: Continue		D: :1	D: :1	D: :1	D: 1	D: 1	D: :1	D: 11
Seamount: Sample no.:	Rigil 1374-22	Rigil 1374-23	Rigil 1374-24	Rigil 1374-25	Rigil 1374-26	Rigil 1374-27	Rigil 1374-28	Rigil 1374-29
Site, hole:	U1374A	U1374A	U1374A	U1374A	U1374A	U1374A	U1374A	U1374A
Core-section:	52R-2	54R-4	60R-1	61R-3	63R-3	64R-6	65R-7	67R-5
Interval (cm):	78–83	30–36	16–21	118–123	112–117	101–104	104–110	70–75
Depth (mbsf):	338.08	359-21	412.59	426.00	444.77	458.80	468-60	485.04
Rock type:	Int	C in HB	Int	C in B	L	C in B	C in B	C in B
Phenocrysts:	Aphyric	Aphyric	Aphyric	M-PI	M-PI	Aphyric	Aphyric	Aphyric
Gd Tb	7·01 1·08	7⋅34 1⋅10	6·09 0·91			9·42 1·33		6·29 0·92
Dy	1.08 5.88	1·10 5·84	4·76			6·86		0.92 4.78
Ho	1.14	1.10	0.89			1·27		0.88
Er	2.98	2.81	2.26			3.25		2.23
Yb	2.33	2.10	1.71			2.44		1.63
Lu	0.34	0.30	0.25			0.34		0.23
Hf	4⋅88	5.29	4.36			7.08		4.59
Ta	1.75	2.08	1.74			3.48		1.89
Th	1.94	2.47	1.89			4.27		2.10
Seamount:	Rigil 1374-30	Rigil	Rigil 1374-32	Rigil	Rigil	Achernar	Achernar 1375-02	Burton
Sample no.: Site, hole:	U1374A	1374-31 U1374A	U1374A	1374-33 U1374A	1374-34 U1374A	1375-01 U1375A	U1375B	1376-01 U1376A
Core-section:	68R-3	68R-5	69R-2	70R-1	72R-2	2R-1	1R-1	1R-1
Interval (cm):	23–28	82–89	127–137	60–65	86–92	30–33	10–15	140–145
Depth (mbsf):	492.22	495.44	501.30	508.93	520.10	8.82	0.13	1.43
Rock type:	Int	Int	В	Int	Int	C in sed	L?	C in sed
Phenocrysts:	Aphyric	Aphyric	Aphyric	Aphyric	Aphyric	Aphyric	M-OI	Aphyric
XRF (wt %) SiO ₂	44.98	42.65	42.37	45.15	44.86	46-30	44.22	41.44
Al ₂ O ₃	13.98	13.12	13.62	16·72	16.88	16.76	15·69	13.60
Fe ₂ O ₃ ^T	13.11	13.56	11.41	14.18	14.03	12.08	13.97	12.00
MgO	6.07	6.29	8.77	5.86	5.90	4.05	3.76	9.80
CaO	9.32	10.06	8.37	8.94	8.79	8.84	10.44	8.50
Na ₂ O	3.71	3.63	3.05	3.38	3.25	3.26	3.36	2.29
K ₂ O	1.157	1.162	0.921	1.000	0.967	1.722	1.456	1.462
TiO ₂ MnO	2⋅961 0⋅191	3·133 0·222	2·126 0·171	3⋅590 0⋅187	3⋅560 0⋅193	3.754 0.136	4·126 0·163	2⋅564 0⋅175
P ₂ O ₅	0.436	0.487	0.338	0.187	0·193 0·546	0.768	0.548	0.175
LOI	3.67	5.33	9.04	0.41	0.71	1.93	1.82	7.76
Total	99.59	99.64	100-19	99.97	99.68	99.60	99.56	99.95
XRF (ppm)								
Zr	229.9	244.2	133.2	288.9	285.5	317·6	271.0	176.5
Nb Y	36⋅8 28⋅3	39⋅4 30⋅3	18⋅2 21⋅3	46⋅6 34⋅6	45⋅6 34⋅4	59⋅5 34⋅3	45⋅6 32⋅0	24·6 24·2
Sr	682·3	586·2	206.9	693·6	686·4	946.4	753·8	265.9
Rb	16.1	13.7	14.5	19.0	17.9	37.3	26.1	64.3
Zn	102.8	118-4	100	92.3	91.9	94.9	90	127⋅5
Cu	51 ⋅8	55⋅1	78	36	35.5	22.1	71	86.2
Ni	42.3	44.5	68-6	33.7	32.2	43.7	59.6	319.7
Cr	22.0	18.1	103.8	16·0	12⋅0 263⋅9	6.7	40·6	219.7
V Ba	303 236·1	340·7 172·3	334·1 142·9	267 225⋅7	263·9 225·7	202⋅8 372⋅1	328⋅5 291⋅1	296⋅3 139⋅9
Sc	24.7	28.2	37.9	18.4	18·3	13.4	16.8	32.1
ICP-MS (ppm)	247		0, 0	10 4	100	10 4	100	02 1
Y Zr		29·97 236·33		33⋅41 271⋅90	36⋅51 310⋅67	34⋅37 343⋅28	32·43 290·01	
Nb		38.93		45.19	49.88	68.69	51.03	
La		25.74		30.75	35.20	44.42	33.45	
Ce		60.15		71.27	80.66	96.93	77.04	
Pr		7.66		8.90	10.30	12⋅06	9.96	
Nd		32.83		37.86	43.33	49.99	42.03	
Sm		7.37		8.25	9·51	10.67	9.39	
Eu Gd		2·33 6·92		2⋅64 7⋅50	3⋅02 8⋅78	3⋅42 9⋅50	3⋅03 8⋅82	
Tb		1.03		7.50 1.12	8.78 1.30	9.50 1.36	8·82 1·26	
Dy		5.44		5.93	6.89	6.96	6.45	
Ho		1.03		1.14	1.33	1.28	1.18	

Table 1: Continue Seamount:	Rigil	Rigil	Rigil	Rigil	Rigil	Achernar	Achernar	Burton
Sample no.:	1374-30	1374-31	1374-32	1374-33	1374-34	1375-01	1375-02	1376-01
Site, hole: Core-section:	U1374A 68R-3	U1374A 68R-5	U1374A 69R-2	U1374A 70R-1	U1374A 72R-2	U1375A 2R-1	U1375B 1R-1	U1376A 1R-1
Interval (cm):	23–28	82–89	127–137	60–65	86–92	30–33	10–15	140–145
Depth (mbsf):	492.22	495.44	501.30	508.93	520·10	8.82	0.13	1.43
Rock type: Phenocrysts:	Int Aphyric	Int Aphyric	B Aphyric	Int Aphyric	Int Aphyric	C in sed Aphyric	L? M-Ol	C in sed Aphyric
Er		2.59		2.96	3.40	3⋅15	2.94	
Yb Lu		1.90 0.28		2⋅34 0⋅34	2⋅72 0⋅39	2⋅32 0⋅32	2·16 0·29	
Hf		5·05		5·76	6·55	7·14	6·32	
Та		2.04		2.46	2.81	4.14	3.07	
Th		2.42		2.84	3.57	4.46	3.28	
Seamount:	Burton	Burton	Burton	Burton	Burton	Burton	Burton	Burton
Sample no.: Site, hole:	1376-02 U1376A	1376-03 U1376A	1376-04 U1376A	1376-05 U1376A	1376-06 U1376A	1376-07 U1376A	1376-08 U1376A	1376-09 U1376A
Core-section:	1R-4	2R-3	5R-4	6R-7	8R-5	11R-4	13R-3	14R-3
Interval (cm):	50-53	123-127	74-79	98-102	111-116	66-72	60-65	3-8
Depth (mbsf):	4⋅72 C in sed	13⋅48 sed	42·83 HB	56.73 C in HB	73⋅35 L	90·43 L	104·27 L	108⋅50 I
Rock type: Phenocrysts:	H-OI	H-OI	H-OI-Cp	Aphyric	H-OI-Cp	H-OI-Cp	H-OI-Cp	H-OI-Cp
XRF (wt%)	00.74	04.44	40.00	47.40	45.47	45.00	44.00	45.00
SiO_2 Al_2O_3	39.74 10.72	24·44 6·93	42⋅62 9⋅66	47·46 14·68	45⋅17 10⋅18	45·00 9·88	44⋅99 9⋅76	45⋅69 11⋅69
Fe ₂ O ₃ [†]	13.67	9·10	13.61	12.52	13.61	13.59	13.80	13.16
MgO	13.04	7.52	16.34	5.46	16.87	17.01	17.90	13.15
CaO Na ₂ O	11⋅31 2⋅23	25⋅23 1⋅49	3⋅35 1⋅88	10⋅33 2⋅78	8·13 1·73	8⋅60 1⋅75	7·98 1·69	9·16 2·08
K ₂ O	1.250	1.098	1.027	1.368	0.608	0.590	0.580	0.699
TiO ₂	2.363	1.538	1.978	2.991	2.082	2.042	2.036	2.373
MnO	0·165 0·900	0.082 0.540	0·163 0·194	0·171 0·414	0·178 0·280	0·179 0·271	0·181 0·273	0·171 0·322
P ₂ O ₅ LOI	4.06	22.06	8·95	1·55	1.25	1.01	0.76	0.322
Total XRF (ppm)	99.45	100.03	99.77	99.72	100.09	99.93	99.95	99-42
Zr	218.9	118-3	121.4	217.6	151.7	148-1	148-2	177.7
Nb	60·9	32·5	19·8	33·2	22.5	21.8	21.6	26.3
Y Sr	26⋅7 620⋅8	14⋅5 377⋅5	17⋅7 43⋅9	28⋅9 479⋅0	21⋅4 296⋅1	20⋅5 318⋅1	20·6 292·5	24·1 365·4
Rb	21.3	15.9	16.9	33.7	11.9	11.6	11.8	13.6
Zn	125.3	111.8	135.4	128.2	108.7	105.1	105.4	107.9
Cu Ni	60·1 396·6	58⋅2 330⋅7	78·1 602·9	41⋅8 133⋅6	69·2 498·7	67.6 525.6	66∙4 549∙1	77⋅8 349⋅5
Cr	489·1	361.3	781·2	114·9	808.1	868.3	824.3	609.6
V	199.8	98	181.5	295.1	207.8	195.6	192.4	225.9
Ba Sc	425⋅8 22⋅2	144·1 26·2	37⋅1 26⋅3	186⋅3 28⋅9	124⋅6 22⋅7	123⋅3 20⋅8	116⋅1 21⋅7	139⋅4 23⋅3
ICP-MS (ppm)	22 2	202	200	200	22 /	200	217	
Y			18.71	31.11	21.62			24.32
Zr Nb			134⋅30 20⋅45	241·19 34·46	159⋅09 22⋅11			182⋅60 25⋅36
Ва			25.92	187.92	114.09			133.20
La			14·75	26.61	17·17			19·53
Ce Pr			33⋅78 4⋅54	60·26 7·89	39⋅12 5⋅19			45.75 5.90
Nd			19.65	33.58	22.23			25.06
Sm			4.64	7.66	5.24			5.87
Eu Gd			1⋅47 4⋅53	2⋅52 7⋅51	1.70 5.19			1.91 5.83
Tb			0.67	7·51 1·11	0.78			0.88
Dy			3.57	5.94	4.21			4.74
Ho Er			0·67 1·72	1⋅13 2⋅86	0⋅81 2⋅05			0.90 2.29
Yb			1·72 1·29	2·16	1.56			1.75
Lu			0.17	0.30	0.22			0.24

Table 1: Continue	ed							
Seamount:	Burton	Burton	Burton	Burton	Burton	Burton	Burton	Burton
Sample no.:	1376-02	1376-03	1376-04	1376-05	1376-06	1376-07	1376-08	1376-09
Site, hole:	U1376A	U1376A	U1376A	U1376A	U1376A	U1376A	U1376A	U1376A
Core-section:	1R-4	2R-3	5R-4	6R-7	8R-5	11R-4	13R-3	14R-3
Interval (cm):	50-53	123-127	74-79	98-102	111-116	66-72	60-65	3-8
Depth (mbsf):	4-72	13-48	42-83	56-73	73-35	90·43	104-27	108-50
Rock type:	C in sed	sed	HB	C in HB	L	L	L	L
Phenocrysts:	H-OI	H-OI	H-OI-Cp	Aphyric	H-OI-Cp	H-OI-Cp	H-OI-Cp	H-OI-Cp
Hf Ta Th			3·25 1·30 1·56	5·37 - 2·62	3·70 1·48 1·74			4·16 1·69 1·99
Seamount: Sample no.: Site, hole: Core-section: Interval (cm): Depth (mbsf): Rock type: Phenocrysts:	Burton	Burton	Burton	Burton	Burton	Burton	Burton	Burton
	1376-10	1376-11	1376-12	1376-13	1376-14	1376-15	1376-16	1376-17
	U1376A	U1376A	U1376A	U1376A	U1376A	U1376A	U1376A	U1376A
	15R-2	17R-1	17R-3	18R-1	19R-1	21R-2	21R-2	21R-5
	60-63	135-140	75-80	95-100	130-136	2-7	102-107	17-22
	112-62	126-58	128-88	135-78	145-73	155-38	156-38	159-44
	Int	HB	L	L	L	C in B	L	L
	Aphyric	Aphyric	Aphyric	Aphryic	S-OI	Aphyric	Aphyric	H-OI
XRF (wt%) SiO ₂ Al ₂ O ₃ Fe ₂ O ₃ MgO CaO Na ₂ O K ₂ O TiO ₂ MnO P ₂ O ₅ LOI Total XRF (ppm)	48.75	51.39	47·23	47·37	46.37	47·10	45·13	41-56
	14.51	13.21	14·83	14·86	14.17	13·48	14·01	7-10
	8.92	11.65	11·40	11·00	11.95	12·42	11·93	14-14
	8.13	6.45	4·91	5·15	6.24	7·78	9·74	21-69
	9.10	9.82	11·82	11·98	11.41	10·80	9·19	6-80
	3.22	2.74	2·93	3·05	2.69	2·68	2·16	0-87
	1.156	0.844	1·027	1·016	0.974	0·679	1·113	0-399
	2.933	2.554	3·148	3·214	3.086	2·491	2·988	1-531
	0.162	0.148	0·152	0·127	0.166	0·175	0·167	0-171
	0.436	0.312	0·457	0·465	0.446	0·313	0·432	0-223
	2.53	0.55	1·63	1·20	1.99	1·76	2·94	5-54
	99.85	99.67	99·53	99·43	99.49	99·68	99·81	100-02
Zr	207·0	181·1	232·9	237·7	231·3	166-0	203-6	110·2
Nb	32·0	29·7	36·5	38·1	36·5	25-5	35-1	18·6
Y	27·6	24·4	29·4	30·0	29·2	27-0	26-1	13·7
Sr	459·3	397·0	578·6	591·6	515·7	406-3	416-9	243·4
Rb	11·3	15·3	17·8	15·9	17·3	9-7	14-7	7·7
Zn	107·6	96	79·7	102·8	99·4	110-6	87-2	103·6
Cu	88·8	84	67	59·9	84	82-7	81	43·9
Ni	318·9	103·5	129	98·9	81·3	168-5	141-9	888
Cr	116·0	497·6	151·9	149·1	207·0	383-9	428-2	1131·8
V	313·3	269·9	280·1	294·2	267·1	234-8	282-1	135·2
Ba	171·2	183·6	218·3	225·1	194·4	150-5	206	72·4
Sc	32·1	31·3	26·2	28·5	24·8	23-6	29-9	14·1
ICP-MS (ppm) Y Zr Nb La Ce Pr Nd Sm Eu Gd Tb Dy Ho Er Yb Lu Hf Ta Th	28·28 219·60 31·03 24·93 57·86 7·41 31·04 7·11 2·31 6·96 1·03 5·50 1·05 2·65 2·03 0·28 5·00 2·01 2·51	24·57 188·73 28·53 20·72 47·89 6·06 25·65 5·89 1·94 5·91 0·89 4·80 0·93 2·38 1·86 0·25 4·35 1·81 2·20	30·35 249·79 37·26 28·85 65·78 8·40 35·72 8·07 2·60 7·69 1·13 5·94 1·11 2·79 2·07 0·28 5·63 2·51 3·00		28.80 233.88 35.48 26.55 61.29 7.79 33.16 7.49 2.42 7.16 1.07 5.65 1.08 2.67 2.01 0.28 5.25 2.31 2.65	28.59 177.90 25.60 19.62 45.56 5.87 25.10 6.12 2.03 6.34 0.99 5.45 1.05 2.68 2.06 0.29 4.17 1.66 2.00		

 K_2O

Table 1: Continued Seamount: Burton Burton Hadar Hadar Hadar Hadar Hadar Hadar Sample no.: 1376-18 1376-19 1377-1 1377-2 1377-3 1377-4 1377-5 1377-6 U1377A Site, hole: U1376A U1376A U1377A U1377A U1377A U1377A U1377B 23R-4 23R-6 4R-2 2R-2 Core-section: 3R-2 5R-1 6R-1 6R-3 46-53 130-137 1-6 6-11 62-67 89-94 26-32 Interval (cm): 55-61 Depth (mbsf): 177.23 180.07 16.67 26.49 34.75 44.62 46.57 11.77 Rock type: C in B Int 1 П 1 ı Phenocrysts: Aphyric M-OI-Cp M-OI Aphyric **Aphyric Aphyric** M-OI Aphyric XRF (wt%) SiO₂ 46.89 46.75 51.16 45.15 47.17 46.57 40.06 48.41 Al₂O₃ 14.40 17.66 13.95 11.38 16.29 14.10 15.77 15.57 11.65 10.28 8.17 12.98 12.44 11.08 10.99 9.83 Fe₂O MgO 5.54 5.50 4.09 6.14 7.27 3.46 6.59 4.65 CaO 11.63 12.80 6.76 6.32 5.25 7.66 14.76 10.29 2.84 2.50 3.64 3.22 2.53 2.84 Na₂O 3.51 2.17 K_2O 0.935 0.898 1.70 1.15 1.53 2.49 1.51 1.76 2.901 3.500 3.658 TiO₂ 2.942 3.725 3.462 2.771 3.467 0.153 0.055 0.079 0.085 MnŌ 0.1490.073 0.154 0.107 0.401 0.411 0.524 0.535 0.549 0.578 0.560 0.551 $P_{2}O_{5}$ LÕI 8.91 2.67 3.12 2.73 3.97 3.41 7.46 1.38 Total 100.05 99.41 99.86 99.46 99.95 99.51 99.85 99.56 XRF (ppm) 266.7 278.4 285.1 271.9 280.3 Zr 211.4 213.1 272.0 Nb 32.5 35.4 43.0 45.4 45.8 47.5 46.2 44.4 30.5 30.9 27.9 26.8 24.0 25.2 25.9 21.4 Sr 496-1 564.6 497.0 663.3 546.7 467.2 653.2 574.9 Rb 16.3 10.8 27.8 17.5 28.8 36.6 27.3 43.8 Zn 99.2 118 232.6 149.3 208.7 141.5 276.2 129 77.5 68.6 Cu 89.8 85.5 55.4 56.7 61.1 91.1 205 367.1 Ni 110.4 153.2 440.7 104.3 60.7 69 Cr 97.8 384.7 649.7 109.1 338.5 25.1 702.6 99.5 ٧ 255.5 293.3 301.6 349.1 292.4 332.9 189.1 286.5 Ba 187.3 197.9 239.6 256.1 243.9 299.6 271.4 257 Sc 23.8 32.3 37.4 31.9 31.9 25.1 18.8 24.8 ICP-MS (ppm) 29.24 27.86 23.00 24.87 27.71 30.96 32.23 Zr 228.10 226.90 258-66 299.88 299.30 299.90 268-21 Nb 32.79 35.26 43.06 44.74 48.14 48.63 45.21 25.16 26.55 29.08 29.68 33.83 34.71 34.07 La 60.93 69.89 75.59 78.85 Ce 58.38 65.84 78.70 Pr 7.43 7.75 7.93 8.76 9.40 9.93 9.90 38.45 Nd 31.53 32.87 32.92 37.27 41.23 41.48 Sm 7.22 7.44 7.05 8.25 8.43 9.08 9.22 Fu 2.35 2.42 2.31 2.66 2.62 2.87 2.88 Gd 7.02 7.11 6.25 7.32 7.58 8.34 8.49 Tb 1.06 1.04 0.91 1.03 1.09 1.21 1.24 Dy 5.57 5.51 4.67 5.21 5.63 6.20 6.39 Ho 1.06 1.02 0.84 0.91 1.02 1.15 1.17 2.69 2.61 2.04 2.22 2.56 Fr 2.77 2.92 Yb 2.07 1.95 1.47 1.57 1.95 2.05 2.12 0.28 0.27 0.20 0.22 0.26 0.28 0.29 Lu 6.58 Hf 5.09 5.17 5.97 5.87 6.70 6.59 Ta 2.24 2.44 2.60 2.92 2.98 2.85 2.52 2.66 2.75 3.07 3.76 3.77 3.62 Th Seamount: Hadar Hadar Hadar Sample no.: 1377-7 1377-8 1377-9 U1377B U1377B U1377B Site, hole: Core-section: 4R-2 4R-3 5R-4 Interval (cm): 8-12 72-79 15-18 20.83 Depth (mbsf): 19.44 31.97 Rock type: Int Int Int M-PI M-OI-PI Phenocrysts: Aphyric XRF (wt%) SiO₂ 49.24 49.82 45.02 Al₂O₃ 16.42 16.18 15.18 $Fe_2O_3^T$ 9.61 8.68 12.18 MgO 4.16 4.27 3.72 CaO 10.14 11.30 10.68 3.21 3.20 2.89 Na₂O

1.62

(continued)

2.11

1.33

Table 1: Continued			
Seamount:	Hadar	Hadar	Hadar
Sample no.:	1377-7	1377-8	1377-9
Site, hole:	U1377B	U1377B	U1377B
Core-section:	4R-2	4R-3	5R-4
Interval (cm):	8-12	72-79	15-18
Depth (mbsf):	19.44	20.83	31.97
Rock type:	Int	Int	Int
Phenocrysts:	M-OI-PI	M-PI	Aphyric
TiO ₂	3.048	2.953	3.445
MnO	0.106	0.093	0.193
P_2O_5	0.437	0.372	0.569
LOI	1.92	1⋅39	3.80
Total	99-91	99.59	99.78
XRF (ppm)			
Zr	211.7	210.8	277.7
Nb	32.3	32.1	46.9
Y	28.7	28.4	29.6
Sr	449.3	448.8	628.5
Rb	28.1	28.1	93.7
Zn	120.5	119.4	143.8
Cu	85.5	85.6	69.3
Ni	104	102.8	63.8
Cr	188-1	186·2	49.4
V	276.9	277.8	295.4
Ba	203.5	199·3	285
Sc (CD MC (CD CD)	32⋅5	31.9	28.7
ICP-MS (ppm)		00.40	26.65
Y Zr		26·46 208·51	20·05 271·22
Nb		32.03	271·22 46·94
		32·03 21·34	30·33
La Ce		51·44	70·43
Pr		6.39	8·68
Nd		27.51	36·54
Sm		6.39	7.91
Eu		2.09	2.49
Gd		6.24	7·26
Tb		0.96	1.05
Dy		5·15	5.47
Ho		0.97	1.01
Er		2.50	2.52
Yb		1.88	1.87
Lu		0.27	0.26
Hf		4·81	6·02
Ta		1.87	2.67
Th		1.90	2.26

Rock type abbreviations: L, lava; C, lava clast; sed, sedimentary conglomerate or breccia; B, volcaniclastic breccias; HB, hyaloclastite breccias; Int, intrusive sheet. Phenocryst mineralogy abbreviations: OI, olivine; PI, plagioclase; Cp, clinopyroxene; Aphyric, <1% phenocrysts; H, highly phyric, >10% phenocrysts; M, moderately phyric, >2%-10% phenocrysts; S, sparsely phyric, 1-2% phenocrysts. Fe₂O₃^T is total Fe reported as Fe₂O₃; LOI is loss on ignition at 1100 °C.

incompatible elements. This feature is important because it implies that magmas in both seamount trails could have had a source resembling primitive (chondritic) mantle that had been only slightly depleted through the loss of a small melt fraction at some time in its history. (5) The Louisville and Emperor patterns are similar in shape to those for primitive OJP basalt, but are rotated about the least incompatible element (Lu) suggesting that the LST and HEST magmas could have been generated by variable but smaller-degree melting of a mantle source similar to the one that produced the OJP magmas. The nature of the LST mantle source and its partial melting will be discussed in the next section.

FORMATION OF THE LOUISVILLE AND EMPEROR MAGMAS

As was noted earlier, the LST and HEST have all the characteristics predicted for mantle plumes. The HEST is a product of a vigorous mantle plume and its output is increasing, in contrast to the LST whose output was always less vigorous (much smaller seamounts) and may well have ceased altogether in the last million years. Noble-gas isotopic studies on the LST (Hanyu, 2014) show that $^3\text{He}/^4\text{He}$ ranges up to 10-6 Ra, much lower than in basalt from Hawaii (up to $\sim\!30\,\text{Ra}$), but two LST basalt samples have a primordial Ne isotopic signature that implies a deep-mantle origin.

Table 2: Average values obtained during this study by XRF spectrometry for international standards BCR-1, BHVO-1, BHVO-2, BIR-1 and BE-N

	BCR-1 Mean (ppm)	RSD %	BHVO-1 Mean (ppm)	RSD %	BHVO-2 Mean (ppm)	RSD %	BIR-1 Mean (ppm)	RSD %	BE-N Mean (ppm)	RSD %
n					5					
Nb	13.0		19.8			0.6	0.6		120.5	
Zr	192.5		173·6		172·6	0.2	17·9		273.1	
Y	38.1		27.1		26.8	0.6	16.5		30.2	
Śr	328.4		382.7		380.8	0.2	107.1		1371.6	
Rb	47⋅3		9.3		9.3	2.1	0.2		47.3	
n	3		3		3		3		3	
Zn	117⋅3	0.35	111.1	0.09	101.5	0.44	65⋅1	0.54	122.7	0.21
Cu	19∙4	1.4	140.0	0.33	122.0	0.36	123.1	0.25	73.0	1.4
Ni	12.7	3.1	127.5	0.20	120.2	0.10	153⋅6	0.55	272.0	0.30
Cr	14.0	5.4	311.7	0.27	304.7	0.05	381.7	0.29	383.4	0.26
V	394.7	0.20	323.9	0.64	318.3	1.1	307.2	0.33	246.8	0.47
Ba	669.7	0.55	144.0	1.6	134.7	1.3	9.7	2⋅1	1038-3	0.16
Sc	30.7	1.7	34.5	1.2	33.4	2.4	40⋅3	0.76	24.5	3⋅1

Table 3: Average values obtained during this study by ICP-MS for international standards BR and W2

	DD		14/2	
	BR Mean (ppm)	RSD %	W2 Mean (ppm)	RSD %
n	11		11	
Υ	30.9	3⋅2	22.5	2.8
Zr	278.4	5.1	89.8	2.8
Nb	122.6	4.6	7.6	2.7
La	85.0	7.7	10⋅7	5⋅1
Ce	155.9	6⋅0	23.6	2.8
Pr	17⋅6	6⋅2	3.0	4.4
Nd	66.8	5.6	13.0	4.0
Sm	12⋅2	5.6	3.3	4.1
Eu	3.7	5.9	1.1	4.9
Gd	10⋅2	5.1	3.7	5⋅1
Tb	1.3	6⋅0	0.63	4.9
Dy	6.3	5.9	3.8	4.7
Ho	1.1	6⋅2	0.80	5⋅0
Er	2.6	5.9	2⋅3	4.8
Yb	1.8	6.7	2.0	5⋅2
Lu	0.24	6.0	0.30	4.2
Hf	5.5	6.6	2⋅3	4.1
Ta	5.6	7.6	0.49	3.0
Th	10⋅6	9.9	2.2	6.2

Both seamount chains have been active over a similar time interval (\sim 80–0 Ma) in the Pacific Ocean. The age of the underlying ocean crust for each increases with decreasing seamount age (Fig. 2), and so both the LST and HEST seamounts were built on ocean lithosphere whose age, at the time of seamount formation, varied considerably over approximately the same range (\sim 15–80 Ma). Thus, the two sets of seamount data represent magmas formed from hotspots of varied vigour beneath lithosphere of widely varying thickness. If the relative vigour of the two hotspots is due largely to differences in mantle $T_{\rm p}$, then the composition of their respective magmas should be predictable by a simple decompression melting model.

The nature of the mantle source supplying magmas to Hawaii and other ocean islands has recently been the subject of considerable debate. Several researchers have used the major-element composition of basalts

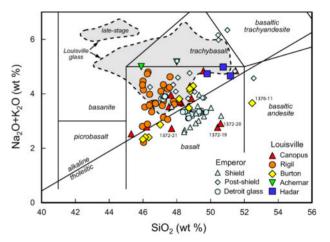


Fig. 4. A plot of total alkalis against silica (after Le Bas *et al.*, 1986), showing the range of composition of volcanic rock samples with <2 wt% LOI recovered from the Louisville seamounts during IODP Expedition 330, and by dredging and drilling on the Emperor seamounts (from Regelous *et al.*, 2003; Huang *et al.*, 2005). All analyses have been recalculated to sum to 100 wt% with Fe as FeO. The line separating Hawaiian tholeitic and alkaline basalts is from Macdonald & Katsura (1964). The field of analyses of Louisville glass samples (from Nichols *et al.*, 2014)) covering the same drilled interval as the wholerock samples is shown in grey. It should be noted that many of the least altered whole-rock samples appear to have lost alkalis.

(e.g. Herzberg, 2011; Jackson et al., 2012) and the Ni content of their olivine phenocrysts (Sobolev et al., 2005) to argue that the source is at least partly composed of pyroxenite or eclogite rather than entirely peridotite. The return of oceanic crust to the mantle through subduction provides the source of this pyroxenite or eclogite. Other researchers, however, have argued that pyroxenite or eclogite forms only a minor component of the mantle and that the composition of Hawaiian and other ocean island basalts can be explained adequately with a peridotite source. For example, Putirka et al. (2011) argued that the high Ni content of olivine

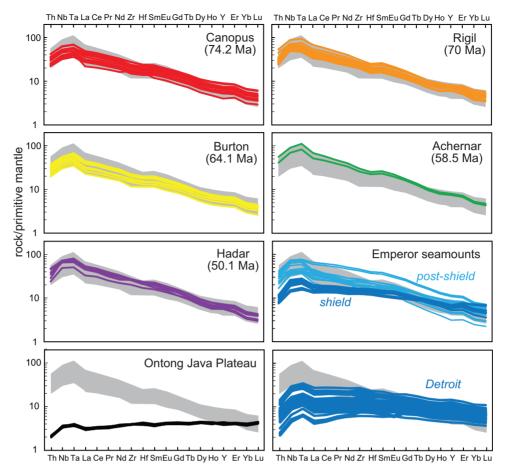


Fig. 5. Incompatible trace element concentrations, normalized to primitive mantle values (McDonough & Sun, 1995), in basaltic rock samples recovered from the Louisville seamounts during IODP Expedition 330 and from the Emperor seamounts by dredging and DSDP and ODP drilling (data from Regelous *et al.*, 2003; Huang *et al.*, 2005). The grey field in all panels shows the range of composition of all the Louisville samples for comparison. Data for primitive (Kroenke-type) Ontong Java Plateau tholeiitic basalt are from Fitton & Godard (2004).

phenocrysts in Hawaiian basalt results from high temperatures of crystallization rather than from the olivinefree mantle source proposed by Sobolev et al. (2005). Putirka et al. (2011) also showed that partial melting experiments conducted on peridotite bulk compositions at pressures of 4-5 GPa can reproduce the range of composition of Hawaiian basalt. Rhodes (2015) showed that the isotopic diversity in basalt erupted over the magmatic history of Mauna Loa contrasts with a lack of major-element variability, suggesting that there is no need for a major pyroxenite or eclogite component in the mantle source. There is less uncertainty in the composition of the mantle source of the LST magmas. Tejada et al. (2015) showed that basalt samples recovered during IODP Expedition 330 from Canopus, Rigil and Burton seamounts have remarkably constant initial ¹⁸⁷Os/¹⁸⁸Os, with values indistinguishable from that of primitive mantle and therefore precluding a significant recycled oceanic crust component in the LST mantle source.

To model the LST and HEST magma composition by decompression melting we consider two mantle compositions: pure peridotite and a 50:50 mixture of

peridotite and pyroxenite–eclogite. The peridotite has a mineralogical composition given by McKenzie & O'Nions (1991): (1) 57·8% olivine, 27·0% orthopyroxene, 11·9% clinopyroxene and 3·3% spinel (spinel-lherzolite); (2) 59·8% olivine, 21·1% orthopyroxene, 7·6% clinopyroxene and 11·5% garnet (garnet-lherzolite). Our mixed source consists of either spinel-lherzolite mixed with pyroxenite (100% clinopyroxene), or garnet-lherzolite mixed with eclogite (50% clinopyroxene, 50% garnet).

Magma composition is a function of mantle composition, the nature of the melting process, and degree and depth of partial melting. To model it we also need to know partition coefficients (*D*) between solid and liquid phases. For the present purposes we use a simple batch melting equation

$$C_L/C_0 = 1/(D + F - PF)$$

where C_0 is the initial concentration of some element in the mantle source, C_L is its concentration in the liquid, F is the melt fraction, D is the average distribution coefficient for the mantle phases weighted by their respective mass fractions before the onset of melting, and P is the

Downloaded from https://academic.oup.com/petrology/article/61/11-12/egaa111/6062409 by VIB Plant Systems Biology Center user on 24 May 2021

Table 4: Distribution coefficients (D) used in this study

			Orthopyroxene	oxene			Clinopyro	xene			Garnet				Spinel		
_	ر	ref.	mean D	1 a	и	ref.	mean D	10	и	ref.	mean D	<u>J</u> a	и	ref.	D	и	ref.
	က	4, 5, 8	0.0057	0.0032	∞	4, 5, 8	0.0741	0.0383	23	1-8	0.0205	0.0166	11	2, 3, 5, 7, 8	0.01	-	6
	4	4, 5, 8	0.0246	0.0186	œ	4, 5, 8	0.2423	0.1550	23	1-8	0.2934	0.2714	1	2, 3, 5, 7, 8	0.01	_	ဝ
	4	4, 5, 8	0.1201	0.0673	∞	4, 5, 8	0.3685	0.1541	22	1-8	3.7042	1.4169	7	'n,	0.01	_	6
	7	4, 5	0.0034	0.0021	9	4, 5	0.0122	0.0137	21	1–7	0.0222	0.0195	10	2, 3, 5, 7	0.08	_	10
	က	4, 5	0.0180	0.0069	9	4, 5	0.0938	0.0664	21	1–7	0.3394	0.1311	6	3, 5, 7	90.0	_	10
	7	4, 5	0.0623	0.0181	9	4, 5	0.3382	0.1654	19	1, 3–7	2.0380	0.6311	6	3, 5, 7	0.002	_	Ξ

n, number of individual values of D. References: 1, Hart & Dunn (1993); 2, Hauri et al. (1994); 3, Johnson (1998); 4, McDade et al. (2003); 5, Salters et al. (2002); 6, Skulski et al. (1994); 7, Tuff & Gibson (2007); 8, McKenzie & O'Nions (1991); 9, Irving (1978); 10, Horn et al. (1994); 11, Elkins et al. (2008).

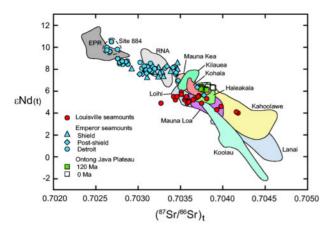


Fig. 6. Sr- and Nd-isotope ratios in basalt samples from the Louisville seamounts collected by dredging (Vanderkluysen et al., 2014) compared with data from the Emperor seamounts (Regelous et al. 2003; Huang et al., 2005) and the Ontong Java Plateau (Tejada et al., 2004). The two sets of OJP data points show the effect of age-correction. Site 884 is located on the outer flank of Detroit seamount. Data fields for basalt from the Hawaiian Islands (RNA, rejuvenated and North Arch) and East Pacific Rise (EPR) are taken from Huang et al. (2013).

average distribution coefficient for the mantle phases weighted by their respective contribution to the melt. We use the following mantle melting proportions for spinel- and garnet-lherzolite, respectively:

```
liquid = -0.22 ol + 0.38 opx + 0.71 cpx + 0.13 sp (Baker & Stolper, 1994)
liquid = 0.08 ol - 0.19 opx + 0.81 cpx + 0.30 gt (Walter, 1998).
```

These values are used for both the peridotite source and the mixed source as they are determined by the composition of the primary magma, not the source. Partition coefficients (*D*) vary significantly with pressure, temperature and liquid composition, and this limits the reliability of partial melting models as the choice of values is always somewhat subjective. Here we have attempted to be more objective by using averages of all appropriate *D* values (Table 4).

It has been suggested that the LST and the 122 Ma Ontong Java Plateau (OJP) represent, respectively, the tail and head of the same mantle plume (e.g. Mahoney & Spencer, 1991). Basalt samples from the two have PM-normalized incompatible trace element patterns that are similar in shape (Fig. 5), implying that the LST magmas could have been produced by small-degree melting from a mantle source similar to that of the OJP magmas. The two also have similar initial ⁸⁷Sr/⁸⁶Sr and ¹⁴³Nd/¹⁴⁴Nd (Mahoney & Spencer, 1991; Vanderkluysen et al., 2014), and Tejada et al (2004) have shown that basalt samples from the OJP and Hawaii have nearidentical Sr-, Nd-, Pb- and Hf-isotope ratios. The similarity in initial 87Sr/86Sr and Nd between basalt samples from the LST, Hawaii and OJP (Fig. 6) suggests that these have compositionally similar mantle sources. Data from the other Hawaiian islands and the HEST, however, show a much larger range indicating

considerable heterogeneity in the Hawaiian plume (e.g. Huang et al., 2013; Harrison et al., 2020). The array of HEST data extending towards the composition of East Pacific Rise basalt (Fig. 6) suggests either mixing with ambient upper mantle or the presence of a depleted, refractory plume component that is sampled only by large degrees of melting when the plume was situated close to a mid-ocean ridge (Regelous et al, 2003). Recent work by Harrison et al. (2020) has shown that the HEST has sampled both ambient upper mantle and a depleted plume component in variable proportions over time. Despite heterogeneity in the HEST mantle source, the isotopic similarity of basalt from the OJP, LST, Hawaii and some of the HEST (Fig. 6) suggests that these have similar mantle sources. It should therefore be possible to model their incompatible-element abundances using a common mantle source based on the composition of primitive OJP basalt. We use this approach to calculate the composition of the peridotite mantle source. The pyroxenite and eclogite components of our mixed source are assumed to have the incompatible-element abundances of average normal mid-ocean ridge basalt given by Sun & McDonough (1989).

Calculating the composition of the OJP mantle source is straightforward because primitive (Kroenketype) basalt, containing sparse microphenocrysts of olivine and Cr-spinel, was recovered from the OJP during ODP Leg 192 (Mahoney et al., 2001). Using forward and inverse mantle-melting modelling, Herzberg (2004) showed that these basalts were produced by around 30% melting, and Fitton & Godard (2004) obtained the same value using incompatible trace element concentrations measured on the same samples. Tejada et al. (2004) showed that the isotopic characteristics of OJP basalts can be explained by removing ~1 % by mass of a small melt fraction (e.g. continental crust) from primitive mantle at about 3 Ga, a process that can also account for the incompatible-element composition of Kroenke-type basalt (Fitton & Godard, 2004). In the present study we calculate the peridotite mantle composition from that of Kroenke-type basalt. First, we calculate the composition of the primary magma by incrementally adding equilibrium olivine to the composition of six unaltered samples of Kroenke-type basalt from ODP Sites 1185 and 1187 until the olivine composition reached Fo_{91.6} (Herzberg, 2004). The amount of olivine added ranges from 29.6 to 33.4 wt% and the calculated primary magmas contain 18-9-19-4 wt% MgO. We then calculate the composition of the OJP mantle source by assuming that the primary magmas were produced by 30% melting, leaving a harzburgite residue composed of 75% olivine and 25% orthopyroxene. The composition of the mantle source was calculated by mass balance:

$$C_0 = C_1 (D' + F - D'F)$$

where C_L is the average primary magma composition and D' is the bulk D of the residual harzburgite,

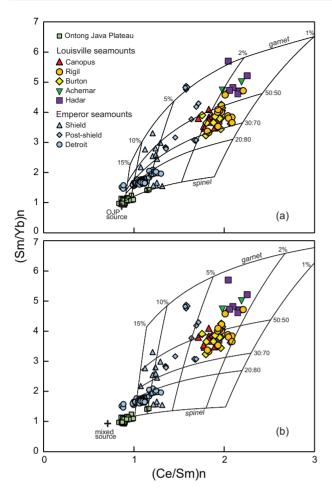


Fig. 7. Chondrite-normalized Sm/Yb plotted against chondrite-normalized Ce/Sm for basalt samples from the Emperor (data from Regelous et al., 2003; Huang et al., 2005) and Louisville seamounts. Data (from Fitton & Godard, 2004) for basalt from the Ontong Java Plateau (OJP) are included for comparison. (a) Partial melting curves are based on a mantle source composition derived from the composition of primitive OJP tholeiitic (Kroenke-type) basalt and are labelled with melt per cent and proportion of spinel- and garnet-lherzolite in the source. (b) The melting curves are for a mantle source composed of a 50:50 mixture of peridotite and pyroxenite–eclogite. Details of the melting calculations are given in the text, and the D values used are given in Table 4.

calculated from the D values given in Table 4. Because $D\approx 0$ for incompatible elements in olivine and orthopyroxene, $C_0\approx 0.3~C_L$ for 30 % melting. The calculated mantle composition and further details of the calculations have been given by Hastie *et al.* (2016).

The results of modelling REE ratios in the Louisville and Emperor data are shown in Fig. 7. The LST data define a trend in which decreasing degree of partial melting (~3 to ~1.5%) is accompanied by increasing proportion of garnet (30–100%) in the case of the peridotite mantle source (Fig. 7a). A mixed peridotite + pyroxenite-eclogite mantle source (Fig. 7b) can also reproduce the REE ratios, but with larger implied melt fractions. The younger seamounts (Achernar and Hadar) appear to have been formed by smaller degrees of melting than do the older seamounts (Canopus, Rigil

and Burton). Data from the HEST generally indicate higher degrees of melting (\sim 2 to \sim 10%) of a spinel- to garnet-lherzolite source. Although it is common practice to use the batch melting equation to model geochemical data, melting accompanied by melt segregation is a dynamic process that lies between batch and fractional melting. Consequently, the melt fractions implied by the model curves in Fig. 7 can give only an approximation to reality. However, the use of the batch melting equation is justified here because the results of batch and fractional melting converge as melt fraction (F) tends to zero, and the relative variation in depth and melt fraction will be robust. We use a more sophisticated incremental melting model later to assess effects of lithosphere thickness on composition.

An attempt to model the concentrations of Nb and Zr in the LST and HEST data is shown in Fig. 8. The LST data define an array that extrapolates to the origin, suggesting that it is mostly controlled by low-pressure fractional crystallization of phases (olivine, plagioclase and augite) containing low concentrations of Nb and Zr, but it is clear that the younger seamounts have slightly higher Nb/Zr than the older seamounts. The modelling results for a peridotite source (Fig. 8a) are broadly consistent with the degrees of melting implied by the REE ratios (Fig. 7a), but the attempt to model Zr and Nb concentrations using a mixed source (Fig. 8b) failed to account for the LST data because of the high Zr and Nb contents of the pyroxenite component. Taken together, Figs 7 and 8 show clearly that the HEST was formed by generally higher degrees of melting of a peridotite mantle source containing smaller proportions of garnet than were the LST. It is possible that the HEST and LST formed from mantle sources with significantly different compositions rather than from the common OJP source that we assume here, but the generally similar isotopic composition of Louisville, Hawaiian and OJP basalts (Fig. 6; Mahoney & Spencer, 1991; Tejada et al., 2004, 2015; Vanderkluysen et al., 2014) makes this unlikely. Because the lithosphere age, and therefore thickness, at the time of seamount formation varied over the same range for the LST and HEST (Figs 1 and 2), the differences in degree of melting and mantle mineralogy between the two trails is most probably due to differences in the T_p of their respective sources. This hypothesis is tested in the next section.

VARIATION OF MAGMA COMPOSITION WITH LITHOSPHERE THICKNESS

Decompression melting in mantle plumes is controlled mostly by $T_{\rm p}$ and lithosphere thickness. The depth at which melting starts increases with increasing $T_{\rm p}$, and degree of melting increases steadily with falling pressure until upwelling and decompression are stopped by the lithosphere lid. A simple model of decompression melting in mantle plumes is shown in Fig. 9. In this model melting starts on the dry mantle solidus (F=0)

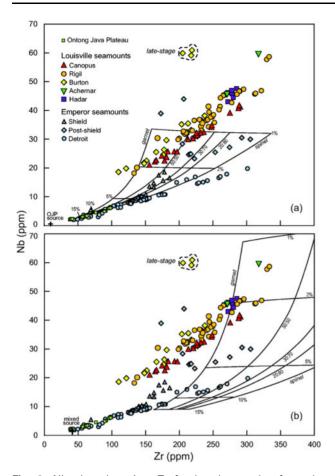


Fig. 8. Nb plotted against Zr for basalt samples from the Emperor (data from Regelous *et al.*, 2003; Huang *et al.*, 2005) and Louisville seamounts. Data for basalt from the Ontong Java Plateau (OJP) are included for comparison. (a) Partial melting curves are based on a peridotite mantle source composition derived from the composition of primitive OJP tholeitic (Kroenke-type) basalt and are labelled with melt per cent and proportion of spinel- and garnet-lherzolite in the source. (b) The melting curves are for a mantle source composed of a 50:50 mixture of peridotite and pyroxenite–eclogite. Details of the melting calculations are given in the text, and the *D* values used are given in Table 4. The late-stage samples from Burton seamount include two glass analyses from Nichols *et al.* (2014).

and reaches a maximum (F_{max}) at the base of the lithosphere. Increasing lithosphere thickness suppresses melting such that no melt can form in volatile-free mantle with $T_p = 1300$ °C when lithosphere thickness exceeds 57 km, whereas it would take a lithosphere thickness of ~120 km to prevent melting of mantle with $T_{\rm p} = 1500\,^{\circ}{\rm C}$ (Fig. 9). The rate at which melt is produced with falling pressure above the mantle solidus is governed by the latent heat of fusion. Here we use a value of 1.2 % kbar⁻¹ (0.4 % km⁻¹), which is the value proposed by Klein & Langmuir (1987) and is within the range of values calculated by McKenzie & Bickle (1988). We use the same value for both our peridotite and mixed mantle sources because it relates mostly to the conversion of excess temperature above the solidus to the latent heat of fusion of clinopyroxene and garnet to produce basaltic melt. The rate of melt production should

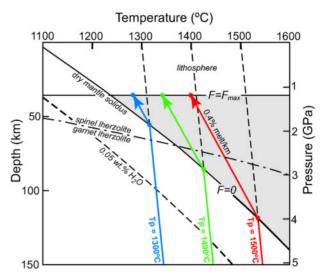


Fig. 9. A decompression melting model for mantle with potential temperature ($T_{\rm p}$) of 1300, 1400 and 1500 °C. The mantle has an adiabatic gradient of $0.3\,^{\circ}{\rm C\,km^{-1}}$ (Turcotte & Schubert, 1982) until it crosses the dry solidus and cools at a faster rate, producing melt at $0.4\,^{\circ}{\rm km^{-1}}$ (Klein & Langmuir, 1987). Decompression and melting stop at the base of the lithosphere. The dry mantle solidus and the solidus of mantle with $0.05\,^{\circ}{\rm km^{-1}}$ (water are from Katz *et al.* (2003), and the depth of the transition from spinel- to garnet-lherzolite is from Klemme & O'Neill (2000).

therefore be insensitive to the proportion of olivine and orthopyroxene in the source. We also assume that the garnet–spinel transition is the same in both our peridotite and mixed mantle sources. In the absence of volatiles, melts from mantle with $T_{\rm p}\!<\!1350\,^{\circ}{\rm C}$ will form entirely from spinel-lherzolite whereas hotter mantle will melt initially in the garnet-lherzolite stability field. The addition of small amounts of volatiles (H₂O or CO₂) to the mantle will result in the formation of small amounts of melt at greater depth (Fig. 9), with the amount of melt limited by the volatile content and its solubility in the melt.

We can use the model in Fig. 9 to calculate the chemical composition of magmas from the mantle T_p and the lithosphere thickness. To do this we assume that the melt is the aggregate of all partial melts in a column of mantle extending from the depth to the mantle solidus (F=0) to the base of the lithosphere $(F=F_{max})$. We use Nb/Zr and Ce/Y to illustrate the effect of T_p and lithosphere thickness on melt composition (Fig. 10). These ratios are useful because they represent the whole range of incompatible elements (Fig. 5) and are unaffected by low-temperature alteration or low-pressure fractional crystallization, but both are sensitive to the presence of garnet in the source. We calculate Nb/Zr and Ce/Y in the aggregate melt by calculating the composition of individual contributions to the melt in 1km steps (0.4% increments) through the thickness of the mantle column from the solidus to the base of the lithosphere (Fig. 9). The aggregate melt composition is the average of all the individual melts weighted by their mass fraction. The procedures and parameters used are

as described in the previous section. Figure 10 shows that Nb/Zr and Ce/Y in the aggregate melt are relatively insensitive to mantle source composition, especially in the case of Nb/Zr where the peridotite and mixed source curves are virtually identical. This is not surprising given that our two mantle sources differ only in the modal abundance (but only slightly in relative proportion) of clinopyroxene and garnet, and it is these phases that contribute most to the melt.

To compare our analytical data with the model curves in Fig. 10 requires that we know the thickness of the lithosphere beneath each seamount at the time the seamount formed. The age of the lithosphere at the time of an individual seamount's formation is the difference between the age of the seamount and the age of the seafloor on which it grew, and this can easily be found from the age data in Fig. 2. Converting lithosphere age into lithosphere thickness is much less straightforward, largely because of uncertainties in the appropriate definition of thickness. To what depth can mantle plumes rise before further upwelling is inhibited by a rigid lid? Olson et al. (1988) have shown that thermal plumes do not effectively intrude and erode the lithosphere in a moving plate, so we can use steadystate lithosphere thicknesses. Here we choose the 1100°C isotherm as the likely base of the rigid lithosphere, as found by Watson & McKenzie (1991) in their modelling of melting in the Hawaiian plume, and we use the plate model of McKenzie et al. (2005) to relate lithosphere age to the depth of this isotherm. Lithosphere thicknesses at the time of eruption of individual seamounts within the LST and HEST were estimated and plotted with Nb/Zr and Ce/Y data in Fig. 10. Data for the shield-forming lavas of Mauna Kea, sampled by the Hawaii Scientific Drilling Project (Rhodes & Vollinger, 2004), are included for comparison. Adjustments were made to the data for the HEST (from Regelous et al., 2003; Huang et al., 2005) and Hawaii to allow for small differences in the values reported for USGS reference standard BHVO-2 from those we obtained when analysing the LST samples (Table 2).

The model curves in Fig. 10 fit the HEST and Hawaii data remarkably well considering uncertainties in lithosphere thickness and the assumptions we made in our melting calculations. The low-Nb/Zr, low-Ce/Y HEST samples, including those from Hawaii, are all from the shield-forming stages of their respective volcanoes, and these can be explained by decompression melting of mantle with $T_p = 1500$ °C. This temperature is similar to that for the centre of the Hawaiian plume obtained through numerical modelling (Watson & McKenzie, 1991) and from Fe-Mg relationships in Hawaiian picrite (Herzberg et al., 2007). Samples with higher Nb/Zr and Ce/Y represent the post-shield stage (Regelous et al., 2003) and data from these plot between the curves for $T_{\rm p} = 1300$ and $1400\,^{\circ}$ C, implying that they formed from the cooler sheath around the plume as the seamount drifted away from the plume axis. Thus, the HEST and

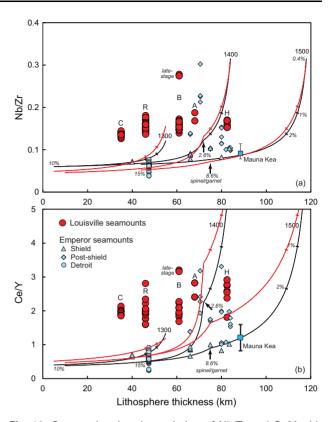


Fig. 10. Curves showing the variation of Nb/Zr and Ce/Y with lithosphere thickness in accumulated partial melts from mantle with $T_p = 1300$, 1400 and 1500 °C, based on the melting model shown in Fig. 9. The black curves are for a peridotite mantle source; red curves are for a 50:50 mixture of peridotite and pyroxenite-eclogite. Nb/Zr and Ce/Y are calculated by summing the composition of partial melts in 1 km (0.4%) increments in a column of mantle extending from the depth of the dry solidus to the base of the lithosphere. The small arrows mark the points at which the curves cross the garnet-spinel transition, and the numbers give the melt per cent at various points along the curves. Lithosphere thickness for the data points is calculated from the age of the lithosphere at the time of eruption and the relationship between age and the depth to the 1100 °C isotherm in the plate model of McKenzie et al. (2005). Data from Detroit Seamount (Huang et al., 2005) and for samples from the shield-forming stage of the other Emperor seamounts (Regelous et al., 2003) plot close to the curve for $T_p = 1500 \,^{\circ}$ C, and data from the post-shield stage plot between the curves for $T_p = 1400 \,^{\circ}\text{C}$ and $1300 \,^{\circ}\text{C}$. This is consistent with the derivation of shield-forming magmas from the hotter axis of a mantle plume and the post-shield magmas from the cooler plume sheath as the seamount drifts away from the plume axis. The mean composition (error bars show the full range of the data) of shield-forming lavas of Mauna Kea, sampled by the Hawaii Scientific Drilling Project (Rhodes & Vollinger, 2004), plots close to the curve for $T_p = 1500$ °C. Data from the Louisville seamounts, by contrast, show no significant relationship between Nb/Zr and Ce/Y and lithosphere thickness at the time of seamount formation. C, Canopus; R, Rigil; B, Burton; A, Achernar; H, Hadar.

Hawaii data can be explained adequately by decompression melting in a mantle plume.

The Louisville data show a slight increase in Nb/Zr and Ce/Y, implying a small decrease in degree of melting, with increasing lithosphere thickness but the variations in Nb/Zr and Ce/Y are far less than predicted by the model (Fig. 10). Adjusting the input parameters

(mantle mineralogy, D values, depth to the spinel-garnet transition, and latent heat of fusion) of our model would change the position of the curves but not their form and would therefore not improve the fit to the LST data. The composition of the two oldest Louisville seamounts (Canopus and Rigil) is particularly difficult to explain by simple decompression melting because they were apparently formed on young lithosphere that was only 35-45 km thick. Figure 10 shows that a mantle plume with $T_p > 1300$ °C impinging on such thin lithosphere ought to melt far more than the 1.5-3% implied by the composition of basalt from these two seamounts (Figs 7 and 8) and produce tholeiitic magmas with much lower Nb/Zr than is observed. It is, of course, possible that such rocks exist at depths beyond the range of drilling, but drilling at Site U1374 on Rigil Seamount penetrated about 15% of the distance to the base of the seamount (a minimum value because some material would have been removed by erosion) without any systematic change in basalt composition (Fig. 3). Shieldforming tholeiite forms >95% of the volume of Hawaiian volcanoes (Clague, 1987). Possible reasons for the misfit between the predicted and observed composition of the Louisville seamounts are discussed in the next section.

ORIGIN OF THE LOUISVILLE SEAMOUNTS

Our decompression melting model can explain the compositional variation of the HEST very well and yet appears unable to account for the relatively constant composition of the LST. One possible reason for this could be that the seafloor ages that we have used, and hence the estimated lithosphere thicknesses, are in error. The seafloor beneath the northwesterly part of the LST formed during the Cretaceous normal superchron (121-83 Ma; Cande & Kent, 1995) and so its age is unconstrained by magnetic anomalies. Consequently, the seafloor ages given in Fig. 2 were based on inferred ages for the initiation and cessation of spreading on the Osbourn Trough. The age of its initiation is based on the assumption that the Early Cretaceous Manihiki and Hikurangi plateaux originally formed part of the Greater Ontong Java Plateau and were rifted apart by spreading on the Osbourn Trough. Its demise is dated through the tentative identification of magnetic anomalies 33 and 32 around the trough (Billen & Stock, 2000; Worthington et al., 2006; Downey et al., 2007). It is possible that spreading on the Osbourn Trough ceased much later than 86 Ma, but this would require higher spreading rates. Billen & Stock (2000) estimated that a full spreading rate of 20 cm a⁻¹ would be required if spreading ceased at 105 Ma. The relative constancy in the composition of the seamounts would require that they were all emplaced on lithosphere of similar thickness if the magmas were generated by simple decompression melting to the base of the lithosphere. This would require a lithosphere age of at least 75 Ma (e.g. McKenzie et al., 2005) and therefore a present-day age of \sim 150 Ma

for the ocean crust beneath Canopus Seamount, an age that falls a long way outside all current estimates and would therefore require a radical reassessment of the history of the western Pacific Ocean. We must conclude that the lithosphere ages given in Fig. 2 are at least approximately correct and therefore find an alternative explanation for the misfit.

Seamounts on the HEST are generally much larger than those on the LST, implying a much higher rate of magma production in the former. As noted earlier, King & Adam (2014) estimated that the HEST mantle buoyancy flux is nearly an order of magnitude higher than that for the LST. If the two seamount trails formed on seafloor with the same range of ages (Fig. 2) then the lithosphere beneath the two has the same range of thickness. The greater magma productivity in the HEST must, therefore, imply a higher T_p in the Hawaiian hotspot than in that responsible for the LST. Nichols et al. (2014) have used the volatile content of glassy margins of intrusive sheets in Burton and Hadar to estimate depth of emplacement and hence the uplift and subsequent subsidence history of the two seamounts. They showed that these can be explained by a thermal anomaly beneath the Louisville seamounts that is no more than 100 °C hotter than normal upper mantle temperature, implying a T_p of 1300–1400 °C beneath these seamounts. This is at least 100 °C cooler than our estimate for the temperature of the mantle beneath the Emperor seamounts. Attempts to verify this independently with PRIMELT3 (Herzberg & Asimow, 2015) failed because even the most primitive LST magmas, represented by basalt samples with Mg# > 70, had crystallized clinopyroxene as well as olivine.

We show that mantle with $T_p = 1350-1400$ °C decompressing to depths as shallow as 40-50 km should melt to a much larger extent than the 1.5-3% implied by our data (Fig. 10). Furthermore, mantle with $T_p \le 1350 \,^{\circ}\text{C}$ would melt entirely within the spinel-lherzolite stability field (Fig. 9), whereas the composition of our LST samples requires that a large proportion of the magma formed from garnet-lherzolite (Figs 7 and 8). Earlier we posed the question: to what depth can mantle plumes rise before further upwelling is inhibited by a rigid lid? The answer in the case of the Louisville seamounts is clearly not at the base of the lithosphere as defined by plate models (e.g. McKenzie et al., 2005). We need a mechanism for generating small-degree melt, largely in the garnet-lherzolite stability field, and then preventing further melting irrespective of lithosphere thickness. Melting in the presence of a small amount of volatile components (H₂O or CO₂) could provide one. The addition of as little of 0.05% of water to the mantle increases the depth of onset of melting by ~45 km (Katz et al., 2003; Fig. 9). This value is similar to that determined by Sarafian et al. (2017), who showed that the addition of 0.045% of water increases the depth of onset of mantle melting by 34 km. The amount of melt produced at depths greater than that of the dry solidus

would be limited by the concentration of volatile components in the mantle and their solubility in the melt.

The presence of water has been shown to lower peridotite viscosity by 2-3 orders of magnitude relative to anhydrous peridotite (Hirth & Kohlstedt, 1996). Ito et al. (1999) showed that dehydration of the mantle through partial melting results in a 50-fold increase in its viscosity, which they argued is enough to restrict further upwelling of a hot ($T_p = 1530$ °C) mantle plume beneath a mid-ocean ridge, as in Iceland. Hirth & Kohlstedt (1996) calculated that 125 ppm of water in the mantle beneath mid-ocean ridges would result in 1-2% melting between the wet and dry solidi. Thus it seems likely that the production of a small amount of volatile-rich melt in a relatively cool mantle plume ($T_p = 1350-1400$ °C), such as that responsible for the LST, would increase the viscosity of the residual mantle and prevent it from reaching the dry mantle solidus and melting further. A mantle plume with $T_p = 1350$ °C would start to melt at a depth of ~120 km (Fig. 9) and produce a small amount of melt in the garnet-lherzolite stability field. Further ascent and decompression would be limited by increasing viscosity, and this could prevent the mantle plume from reaching the dry mantle solidus and the garnet-spinel boundary at ~70 km. Varying the lithosphere thickness between 35 and 60 km, as in the case of Canopus, Rigil and Burton seamounts (Fig. 10), would have little if any effect on magma composition. The very slight reduction of melt fraction with decreasing age apparent in the LST data (Figs 7, 8 and 10) could reflect a very reduced influence of lithosphere thickness or could be due to a small reduction of $T_{\rm p}$. A hotter mantle plume ($T_{\rm p} \approx 1500\,^{\circ}$ C) beneath the HEST would have lower viscosity before the onset of melting, melt to a larger extent, and decompress to the base of the lithosphere as in the model in Figs 9 and 10. The initial 1-2% of melt produced by dehydration melting of garnet-lherzolite would be diluted by much larger degrees of melting of spinel-lherzolite. In this way our simple decompression melting model can account for the composition of both the HEST and LST.

COMPARISON WITH OTHER OCEAN-ISLAND BASALTS

The lack of variation in the composition of the LST with lithosphere age and thickness raises the question of whether the same is true of ocean-island basalt (OIB) in general. Figure 11 shows the variation of Nb/Zr and Ce/Y with lithosphere age in a large OIB dataset (from Fitton, 2007). The data show no correlation between Nb/Zr and lithosphere age and only a very weak correlation for Ce/Y. Islands formed on the oldest lithosphere (e.g. the Canary, Comores and Caroline Islands) have a similar range of composition to those formed close to spreading centres (e.g. Ascension, Azores and Rodrigues). This appears to contradict Ellam (1992), who showed that the average Ce/Y values for individual islands vary with lithosphere age and therefore thickness, but this discrepancy is due largely to the much

larger dataset used in the present study. The correlation reported by Ellam (1992) between lithosphere age and Ce/Y in OIB defined a convex-upward data array, whereas a true relationship would produce a convexdownward array (Fig. 11). In a more recent study, Niu et al. (2011) averaged a large set of OIB data in 10 km lithosphere thickness intervals and showed linear variations of several geochemical parameters (notably REE ratios) with lithosphere thickness. The data averages show a twofold increase in Sm/Yb as lithosphere thickness increases from 0 to 90 km, and they ascribed this to a progressive increase in depth and decrease in degree of melting ('the lid effect'). Our data show a similar increase in Ce/Y between the Ascension and Canaries data (Fig. 11) although the data scatter obscures any linear trend. However, our melting model predicts a much greater dependence on lithosphere thickness for ratios such as Ce/Y and Sm/Yb that are sensitive to the abundance of garnet in the mantle (Figs 10 and 11). Decompression melting of volatile-free peridotite is, therefore, incapable of accounting for the composition of most OIB and its weak dependence on lithosphere thickness (Fig. 11).

Basalts from Iceland, the Galapagos Islands, and the shield-forming stage of the Hawaiian Islands and Emperor seamounts have lower Nb/Zr and Ce/Y than other OIB (Fig 11), implying larger degrees of melting consistent with our melting model. These basalts are tholeiitic, in contrast to the alkaline basalt forming most ocean islands and, along with basalt from Samoa (also with low Nb/Zr), have the highest ³He/⁴He measured in OIB (Class & Goldstein, 2005). These islands are clearly different from other ocean islands and are the products of vigorous mantle plumes decompressing under lithosphere of varying thickness. Iceland and the Galapagos Islands are located on or close to spreading centres, whereas the Hawaiian Islands are located on lithosphere that is ~90 Myr old. The Hawaiian Islands are probably unique among ocean islands in being largely composed of tholeiitic shield-forming basalt that formed by melting beneath thick lithosphere. This requires large degrees of mantle melting within the hottest mantle plume observed on Earth today (Watson & McKenzie, 1991; Herzberg et al., 2007).

Other ocean islands are most probably either the product of less vigorous mantle plumes that lack the buoyancy to overcome the stalling effect of dehydration melting or not the product of mantle plumes at all. Some intraplate volcanic provinces, including ocean islands, are demonstrably not related to plume activity (e.g. Guimarães et al., 2020), and this may also be true of the majority of ocean islands. The wide compositional range of OIB magmas generated beneath lithosphere of all thicknesses requires a heterogeneous mantle source (e.g. Dasgupta et al., 2010). Dasgupta et al. showed that the major-element composition of OIB can be explained with a source composed of peridotite mixed with discrete bodies of carbonated eclogite and a silica-oversaturated eclogite. The more easily fusible character

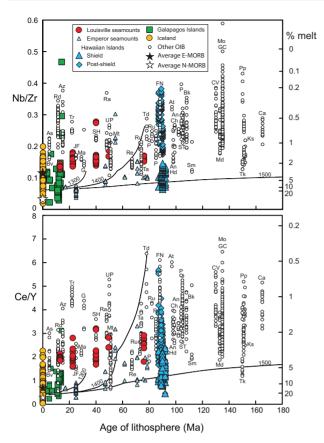


Fig. 11. Plots of Nb/Zr and Ce/Y versus lithosphere age at the time of emplacement for the Louisville and Emperor seamounts compared with data from ocean island basalt (OIB) and mid-ocean ridge basalt (MORB). OIB data from Fitton (2007), except for Bouvet data, which are from Prestvik et al (1999). Hawaiian Islands: Loihi, Hawaii, Kahoolawe, Lanai, Mauai, Oahu, Kauai. Atlantic Ocean: An, Annobon; As, Ascension; Az, Azores; Bk, Bioko; Bv, Bouvet; Ca, Canaries; CV, Cape Verde; FN, Fernando de Noronha; G, Gough; Md, Madeira; P, Principe; SH, St Helena; ST, São Tomé; Td, Trindade; Tr, Tristan da Cunha. Indian Ocean: Ch, Christmas; GC, Grande Comore; Hd, Heard; Ma, Marion; Mo, Moheli; Mt, Mauritius; Rd, Rodriguez; Re, Réunion. Pacific Ocean: At, Aitutaki; JF, Juan Fernandez; Ks, Kosrae; Pp, Pohnpei; Ra, Rapa; Ru, Rurutu; Sm, Samoa; Ta, Tahiti; Tk, Truk; UP, Ua Pou. Principe and Rurutu are identified more than once in the diagram because they have a long history of volcanism. Average normal and enriched MORB (N-MORB and E-MORB) compositions are from Sun & McDonough (1989). The melting curves for $T_p = 1300$, 1400 and 1500 °C are for a peridotite mantle source as in Fig. 10 but with lithosphere thickness converted to age. The curves for a peridotite-pyroxenite plus eclogite mix shown in Fig. 10 have been omitted from this diagram for clarity. Melt percentages along the right-hand side of the diagram are calculated for garnet-lherzolite with the composition of the Ontong Java Plateau mantle source. Details of the melting calculations are given in the text.

of these bodies removes the need for mantle plumes in the majority of ocean islands, which may be formed instead by decompression melting of enriched components within the general upper mantle convective flow.

CONCLUSION

Drilling during IODP Expedition 330 provided volcanic rock samples from five of the Louisville seamounts, and

three of the holes penetrated deep enough to have sampled rocks from the shield-forming phase of seamount growth. Geochemical data on the rock samples confirm the conclusion from earlier dredging expeditions that the Louisville seamounts are mostly composed of alkali basalt with a remarkably uniform composition.

The composition of the Louisville basalt samples can be modelled by $\sim\!1.5\text{--}3\,\%$ melting of garnet-lherzolite with only a small contribution from spinel-lherzolite. By contrast, the composition of basalt from the Emperor seamounts requires larger degrees of melting (2–10%) and mantle sources ranging from spinel- to garnet-lherzolite. A mantle source composition calculated from the composition of primitive tholeiitic basalt from the Ontong Java Plateau (Hastie *et al.*, 2016) was used in all melting calculations.

Decompression melting in a mantle plume impinging on the base of the lithosphere ought to produce magmas whose composition is determined by mantle $T_{\rm p}$ and lithosphere thickness. The ages of the Louisville and Emperor seamounts and of their underlying ocean crust are known, and so the age of the lithosphere at the time of seamount formation can be calculated by subtracting the former from the latter. This is then used to estimate lithosphere thickness from the plate model of McKenzie et al. (2005). Both seamount trails formed on lithosphere with the same range of lithosphere thickness (\sim 35–85 km). The composition of the accumulated melt produced by decompression melting in a column of mantle extending from the mantle solidus to the base the lithosphere is calculated for $T_{\rm p}\!=\!1300$, 1400 and 1500 °C and compared with the composition of basalt from the Emperor and Louisville seamounts. The composition of basalt from the tholeiitic shield-forming phase of the Emperor seamounts and Hawaii compares well with that predicted for mantle with $T_p = 1500$ °C whereas the composition of post-shield alkali basalt fits the prediction for $T_p = 1300-1400$ °C. This is consistent with the generation of the early, shield-forming magmas in the core of a mantle plume and the later, postshield magmas in the cooler plume sheath as the seamount drifts away from the plume axis. Differences between shield and post-shield basalt in radiogenic isotope ratios (e.g. Hanano et al., 2010) and ΔNb (the relative enrichment in Nb on a plot of Nb/Y against Zr/Y; Fitton, 2007; Greene et al., 2010; Harrison et al., 2020) limit the reliability of modelling the two with a common mantle source. However, this is unlikely to affect the conclusion that post-shield magmas were produced by smaller degrees of melting of cooler mantle.

Basalt from the Louisville seamounts is remarkably uniform in composition even though lithospheric thickness at the time of seamount formation varied considerably. Such homogeneity is not predicted by our decompression melting model but the problem can be resolved if the Louisville seamounts were formed by dehydration melting of mantle containing a small amount of volatiles in a cooler plume. Hydrous melting in a

relatively cool mantle plume ($T_p = 1350-1400$ °C) could produce a small amount of melt and then be inhibited by increasing viscosity from reaching the dry mantle solidus and melting further. The failure of the plume to reach the dry mantle solidus or the base of the lithosphere means that the resulting magmas would have the same composition irrespective of lithosphere thickness. This lack of influence of lithospheric thickness on basalt composition is a general feature of most ocean islands. Only a few ocean islands (e.g. Iceland, Galapagos and the Hawaiian Islands) are composed of tholeiitic basalt whose composition can potentially be explained by dry decompression melting of a peridotite mantle source beneath lithosphere of the respective thickness. The Hawaiian Islands are unique in being composed dominantly of tholeiitic basalt formed by decompression melting beneath thick lithosphere, and this requires an exceptionally hot mantle plume. A mantle plume with $T_p \approx 1500\,^{\circ}\text{C}$ beneath the Emperor seamounts and Hawaiian Islands would have lower viscosity before the onset of melting, melt to a larger extent, and decompress to the base of the lithosphere. Thus, our simple decompression melting model could explain the composition of both the Emperor and Louisville seamounts, but probably not ocean islands in general. Most islands may not be the product of mantle plumes but may instead be formed by decompression melting of heterogeneous mantle within the general upper mantle convective flow.

ACKNOWLEDGEMENTS

This research used samples provided by the Integrated Ocean Discovery Program (IODP). We are grateful to Nic Odling and Sam Hammond for assistance with XRF and ICP-MS analyses, respectively, and to Fred Frey, Mike Garcia, Keith Putirka, Cin-Ty Lee and Shichun Huang for their constructive criticism of this paper.

FUNDING

Funding was provided by the UK Natural Environment Research Council through grants to J.G.F. (grant number NE/1022205/1) and A.D.S. (grant number NE/J005401/1).

REFERENCES

- Baker, M. B. & Stolper, E. M. (1994). Determining the composition of high-pressure mantle melts using diamond aggregates. Geochimica et Cosmochimica Acta 58, 2811–2827.
- Ballance, P. F., Scholl, D. W., Vallier, T. L., Stevenson, A. J., Ryan, H. & Herzer, R. H. (1989). Subduction of a Late Cretaceous seamount of the Louisville Ridge at the Tonga Trench: A model of normal and accelerated tectonic erosion. *Tectonics* 8, 953–962.
- Beier, C., Vanderkluysen, L., Regelous, M., Mahoney, J. J. & Garbe-Schönberg, D. (2011). Lithospheric control on geochemical composition along the Louisville Seamount Chain.

- Geochemistry, Geophysics, Geosystems 12, Q0AM01, doi: 10.1029/2011GC003690.
- Billen, M. & Stock, J. (2000). Morphology and origin of the Osbourn Trough. *Journal of Geophysical Research* 105, 481–489.
- Buchs, D. M., Williams, R., Sano, S. & Wright, V. P. (2018). Non-Hawaiian lithostratigraphy of Louisville seamounts and the formation of high-latitude oceanic islands and guyots. *Journal of Volcanology and Geothermal Research* 356, 1–23.
- Cande, S. C. & Kent, D. V. (1995). Revised calibration of the geomagnetic polarity timescale for the Late Cretaceous and Cenozoic. *Journal of Geophysical Research: Solid Earth* 100, 6093–6095
- Cheng, Q., Park, K.-H., Macdougall, J. D., Zindler, A., Lugmair, G. W., Staudigel, H., Hawkins, J. & Lonsdale, P. (1987).
 Isotopic evidence for a hotspot origin of the Louisville Seamount Chain. In: Keating, B. H., Fryer, P., Batiza, R. & Boehlert, G. W. (eds) Seamounts, Islands, and Atolls.
 Washington, DC: American Geophysical Union, pp. 283–296.
- Clague, D. A. (1987). Hawaiian alkaline volcanism. In: Fitton, J. G. & Upton, B. G. J. (eds) *Alkaline Igneous Rocks. Geological Society, London, Special Publications* **30**, 227–252.
- Class, C. & Goldstein, S. L. (2005). Evolution of helium isotopes in the Earth's mantle. *Nature* **436**, 1107–1112.
- Dasgupta, R., Jackson, M. G. & Lee, C.-T. A. (2010). Major element chemistry of ocean island basalts—Conditions of mantle melting and heterogeneity of mantle source. *Earth and Planetary Science Letters* 289, 377–392.
- Dorais, M. J. (2015). Exploring the mineralogical heterogeneities of the Louisville Seamount Trail. *Geochemistry*, *Geophysics*, *Geosystems* 16, 2884–2899, doi:10.1002/2015G C005917.
- Dorais, M. J. & Buchs, D. M. (2019). Mineralogical characterization of rejuvenated magmatism at Burton Guyot, Louisville Seamount trail. *Contributions to Mineralogy and Petrology* **174**. 66.
- Downey, N. J., Stock, J. M., Clayton, R. W. & Cande, S. C. (2007). History of the Cretaceous Osbourn spreading center. *Journal of Geophysical Research* **112**, B04102.
- Elkins, L. J., Gaetani, G. A. & Sims, K. W. W. (2008). Partitioning of U and Th during garnet pyroxenite partial melting: Constraints on the source of alkaline ocean island basalts. *Earth and Planetary Science Letters* **265**, 270–286.
- Ellam, R. M. (1992). Lithospheric thickness as a control on basalt geochemistry. *Geology* **20**, 153–156.
- Fitton, J. G. (2007). The OIB paradox. In: Foulger, G. R. & Jurdy, D. M. (eds) Plates, Plumes, and Planetary Processes. Geological Society of America, Special Papers 430, 387–412.
- Fitton, J. G. & Godard, M. (2004). Origin and evolution of magmas on the Ontong Java Plateau. In: Fitton, J. G., Mahoney, J. J., Wallace, P. J. & Saunders, A. D. (eds) *Origin and Evolution of the Ontong Java Plateau. Geological Society, London, Special Publications* 229, 151–178.
- Fitton, J. G., Saunders, A. D., Larsen, L. M., Hardarson, B. S. & Norry, M. J. (1998). Volcanic rocks from the southeast Greenland margin at 63°N: composition, petrogenesis and mantle sources. In: Saunders, A. D., Larsen, H. C., & Wise, S. W., Jr. (eds) *Proceedings of the Ocean Drilling Program, Scientific Results* 152. College Station, TX: Ocean Drilling Program, pp. 331–350.
- Greene, A. R., Garcia, M. O., Weis, D., Ito, G., Kuga, M., Robinson, J. & Yamasaki, S. (2010). Low-productivity Hawaiian volcanism between Kaua'i and O'ahu. Geochemistry, Geophysics, Geosystems 11, Q0AC08, doi:10.1029/ 2010GC003233.

- Guimarães, A. R., Fitton, J. G., Kirstein, L. A. & Barfod, D. N. (2020). Contemporaneous intraplate magmatism on conjugate South Atlantic margins: A hotspot conundrum. *Earth and Planetary Science Letters* 536, 116147.
- Hanano, D., Weis, D., Scoates, J. S., Aciego, S. & DePaolo, D. J. (2010). Horizontal and vertical zoning of heterogeneities in the Hawaiian mantle plume from the geochemistry of consecutive postshield volcano pairs: Kohala–Mahukona and Mauna Kea–Hualalai. Geochemistry, Geophysics, Geosystems 11, Q01004, doi:10.1029/2009GC002782.
- Hanyu, T. (2014). Deep plume origin of the Louisville hotspot: Noble gas evidence. Geochemistry, Geophysics, Geosystems 15, 565–576, doi:10.1002/2013GC005085
- Harrison, L. N., Weis, D. & Garcia, M. O. (2020). The multiple depleted mantle components in the Hawaiian–Emperor chain. Chemical Geology 532, 119324.
- Hart, S. R. & Dunn, T. (1993). Experimental cpx/melt partitioning of 24 trace elements. Contributions to Mineralogy and Petrology 113, 1–8.
- Hastie, A. R., Fitton, J. G., Kerr, A. C., McDonald, I., Schwindrofska, A. & Hoernle, K. (2016). The composition of mantle plumes and the deep Earth. *Earth and Planetary Science Letters* 444, 13–25.
- Hauri, E. H., Wagner, T. P. & Grove, T. L. (1994). Experimental and natural partitioning of Th, U, Pb and other trace elements between garnet, clinopyroxene and basaltic melts. *Chemical Geology* 117, 149–166.
- Hawkins, J. W., Lonsdale, P. F. & Batiza, R. (1987). Petrologic evolution of the Louisville Seamount Chain. In: Keating, B. H., Fryer, P., Batiza, R. & Boehlert, G. W. (eds) Seamounts, Islands, and Atolls. Washington, DC: American Geophysical Union, pp. 235–254.
- Herzberg, C. (2004). Partial melting below the Ontong Java Plateau. In: Fitton, J. G., Mahoney, J. J., Wallace, P. J. & Saunders, A. D. (eds) Origin and Evolution of the Ontong Java Plateau. Geological Society, London, Special Publications 229, 179–184.
- Herzberg, C. (2011). Identification of Source Lithology in the Hawaiian and Canary Islands: Implications for Origins. *Journal of Petrology* **52**, 113–146.
- Herzberg, C. & Asimow, P. D. (2015). PRIMELT3 MEGA.XLSM software for primary magma calculation: Peridotite primary magma MgO contents from the liquidus to the solidus. Geochemistry, Geophysics, Geosystems 16, 563–578, doi: 10.1002/2014GC005631.
- Herzberg, C., Asimow, P. D., Arndt, N., Niu, Y., Lesher, C. M., Albarède, F., Fitton, J. G., Cheadle, M. J. & Saunders, A. D. (2007). Temperatures in ambient mantle and plumes: constraints from basalts, picrites and komatiites. *Geochemistry, Geophysics, Geosystems* 8, Q02006, doi:10.1029/2006GC0 01390.
- Hirth, G. & Kohlstedt, D. L. (1996). Water in the oceanic upper mantle: implications for rheology, melt extraction, and the evolution of the lithosphere. *Earth and Planetary Science Letters* 144, 93–108.
- Horn, I., Foley, S. F., Jackson, S. E. & Jenner, G. A. (1994). Experimentally determined partitioning of high field strength- and selected transition elements between spinel and basaltic melt. *Chemical Geology* 117, 193–218.
- Huang, S., Regelous, M., Thordarson, T. & Frey, F. A. (2005). Petrogenesis of lavas from the Detroit Seamount: geochemical differences between Emperor Chain and Hawaiian volcanoes. *Geochemistry, Geophysics, Geosystems* 6, Q01L06, doi:10.1029/2004GC000756.
- Huang, S., Blichert-Toft, J., Fodor, R. V., Bauer, G. R. & Bizimis, M. (2013). Sr, Nd, Hf and Pb isotope systematics of

- postshield-stage lavas at Kahoolawe, Hawaii. *Chemical Geology* **360–361**, 159–172.
- Irving, A. J. (1978). A review of experimental studies of crystal/liquid trace element partitioning. *Geochimica et Cosmochimica Acta* 42, 743–770.
- Ito, G., Shen, Y., Hirth, G. & Wolfe, C. J. (1999). Mantle flow, melting, and dehydration of the Iceland mantle plume. *Earth and Planetary Science Letters* **165**, 81–96.
- Jackson, M. G., Weis, D. & Huang, S. (2012). Major element variations in Hawaiian shield lavas: Source features and perspectives from global ocean island basalt (OIB) systematics. *Geochemistry, Geophysics, Geosystems* 13, Q09009.doi: 10.1029/2012GC004268.
- Johnson, K. T. M. (1998). Experimental determination of partition coefficients for rare earth and high-field-strength elements between clinopyroxene, garnet, and basaltic melt at high pressures. Contributions to Mineralogy and Petrology 133, 60–68.
- Katz, R. F., Spiegelman, M. & Langmuir, C. H. (2003). A new parameterization of hydrous mantle melting. *Geochemistry*, *Geophysics*, *Geosystems* 4, 1073, doi:10.1029/2002GC000433.
- King, S. C. & Adam, C. (2014). Hotspot swells revisited. *Physics of the Earth and Planetary Interiors* **235**, 66–83.
- Klein, E. M. & Langmuir, C. H. (1987). Global correlations of ocean ridge basalt chemistry with axial depth and crustal thickness. *Journal of Geophysical Research* 92, 8089–8115.
- Klemme, S. & O'Neill, H. S. C. (2000). The near-solidus transition from garnet lherzolite to spinel lherzolite. *Contributions to Mineralogy and Petrology* **138**, 237–248.
- Koppers, A. A. P., Duncan, R. A. & Steinberger, B. (2004). Implications of a nonlinear ⁴⁰Ar/³⁹Ar age progression along the Louisville seamount trail for models of fixed and moving hot spots. *Geochemistry Geophysics Geosystems* 5, Q06L02, doi:10.1029/2003GC000671.
- Koppers, A. A. P., Yamazaki, T., Geldmacher, J., et al. (2012a). Limited latitudinal mantle plume motion for the Louisville hotspot. *Nature Geoscience* 5, 911–917.
- Koppers, A. A. P., Yamazaki, T. & Geldmacher, J. & the Expedition 330 Scientists (2012b). Proceedings of the Integrated Ocean Drilling Program 330. Tokyo: Integrated Ocean Drilling Program Management International. doi: 10.2204/iodp.proc.330.2012. http://publications.iodp.org/proceedings/330/330toc.htm
- Le Bas, M. J., Le Maitre, R. W., Streckeisen, A. & Zanettin, B; IUGS Subcommission on the Systematics of Igneous Rocks (1986). A chemical classification of volcanic rocks based on the total alkali–silica diagram. *Journal of Petrology* 27, 745–750.
- Lonsdale, P. (1988). Geography and history of the Louisville Hotspot Chain in the southwest Pacific. *Journal of Geophysical Research* **93**, 3078.
- Luyendyk, B. P. (1995). Hypothesis for Cretaceous rifting of east Gondwana caused by subducted slab capture. *Geology* 23, 373–376.
- Lyons, S. N., Sandwell, D. T. & Smith, W. H. F. (2000). Three-dimensional estimation of elastic thickness under the Louisville Ridge. *Journal of Geophysical Research: Solid Earth* 105, 13239–13252.
- Macdonald, G. A. & Katsura, T. (1964). Chemical composition of Hawaiian lavas. *Journal of Petrology* **5**, 82–133.
- Mahoney, J. J. & Spencer, K. J. (1991). Isotopic evidence for the origin of the Manihiki and Ontong Java oceanic plateaus. *Earth and Planetary Science Letters* **104**, 196–210.
- Mahoney, J. J., Fitton, J. G. & Wallace, P. J. et al. (2001). Proceedings of the Ocean Drilling Program, Initial Reports

- **192.** College Station, TX: Ocean Drilling Program. doi: 10.2973/odp.proc.ir.192.2001
- McDade, P., Blundy, J. D. & Wood, B. J. (2003). Trace element partitioning on the Tinaquillo Iherzolite solidus at 1-5 GPa. *Physics of the Earth and Planetary Interiors* **139**, 129–147.
- McDonough, W. F. & Sun, S-S. (1995). The composition of the Earth. *Chemical Geology* **120**, 223–253.
- McKenzie, D. & Bickle, M. J. (1988). The volume and composition of melt generated by extension of the lithosphere. *Journal of Petrology* **29**, 625–629.
- McKenzie, D.O. & Nions, R. K. (1991). Partial melt distributions from inversion of rare earth element concentrations. *Journal of Petrology* **32**, 1021–1091.
- McKenzie, D., Jackson, J. & Priestley, K. (2005). Thermal structure of oceanic and continental lithosphere. *Earth and Planetary Science Letters* **233**, 337–349.
- Montgomery, A. F. & Johnson, P. H. (1987). Paleomagnetic studies on Leg 91 basalts and sediments. In: Menard, H., Natland, J., Jordan, T. & Orcutt, J. (eds) *Initial Reports of the Deep Sea Drilling Program* 91. Washington DC: US Government. Printing Office, pp. 475–482.
- Morgan, W. J. (1971). Convection plumes in the lower mantle. *Nature* **230**, 42–43, doi:10.1038/230042a0.
- Mortimer, N., Hoernle, K., Hauff, F., Palin, J. M., Dunlap, W. J., Werner, R. & Faure, K. (2006). New constraints on the age and evolution of the Wishbone Ridge, southwest Pacific Cretaceous microplates, and Zealandia–West Antarctica breakup. *Geology* 34, 185.
- Müller, R. D., Sdrolias, M., Gaina, C. & Roest, W. R. (2008). Age, spreading rates, and spreading asymmetry of the world's ocean crust. *Geochemistry Geophysics Geosystems* 9, Q04006, doi:10.1029/2007GC001743.
- Nichols, A. R. L., Beier, C., Brandl, P. A., Buchs, D. M. & Krumm, S. H. (2014). Geochemistry of volcanic glasses from the Louisville Seamount Trail (IODP Expedition 330): Implications for eruption environments and mantle melting. Geochemistry, Geophysics, Geosystems 15, 1718–1738, doi: 10.1002/2013GC005086.
- Niu, Y., Wilson, M., Humphreys, E. R. & O'Hara, M. J. (2011). The origin of intra-plate ocean island basalts (OIB): the lid effect and its geodynamic implications. *Journal of Petrology* 52, 1443–1468.
- Norrish, K. & Hutton, J. T. (1969). An accurate X-ray spectrographic method for the analysis of a wide range of geological samples. *Geochimica et Cosmochimica Acta* 33, 431–453.
- O'Connor, J. M., Steinberger, B., Regelous, M., Koppers, A. A. P., Wijbrans, J. R., Haase, K. M., Stoffers, P., Jokat, W. & Garbe-Schönberg, D. (2013). Constraints on past plate and mantle motion from new ages for the Hawaiian–Emperor Seamount Chain. *Geochemistry, Geophysics, Geosystems* 14, 4564–4584, doi:10.1002/ggge.20267.
- Olson, P., Schubert, G., Anderson, C. & Goldman, P. (1988). Plume formation and lithosphere erosion: a comparison of laboratory and numerical experiments. *Journal of Geophysical Research* **93**, 15065–15084.
- Prestvik, T., Goldberg, S. & Goles, G. G. (1999). Petrogenesis of the volcanic suite of Bouvetoya (Bouvet Island), South Atlantic. Norsk Geologisk Tidsskrift 79, 205–218.
- Putirka, K., Ryerson, F. J., Perfit, M. & Ridley, W. I. (2011). Mineralogy and composition of the oceanic mantle. *Journal of Petrology* 52, 279–313.
- Regelous, M., Hofmann, A. W., Abouchami, W. & Galer, S. J. G. (2003). Geochemistry of lavas from the Emperor Seamounts, and the geochemical evolution of Hawaiian

- magmatism from 85 to 42 Ma. Journal of Petrology 44, 113-140.
- Rhodes, J. M. (2015). Major-element and isotopic variations in Mauna Loa magmas over 600 ka: implications for magma generation and source lithology as Mauna Loa transits the Hawaiian Plume. In: Carey, R., Cayol, V., Poland, M. & Weis, D. (eds) Hawaiian Volcanoes: From Source to Surface. American Geophysical Union, Geophysical Monograph 208, 59–78.
- Rhodes, J. M. & Vollinger, M. J. (2004). Composition of basaltic lavas sampled by phase-2 of the Hawaii Scientific Drilling Project: Geochemical stratigraphy and magma types. *Geochemistry, Geophysics, Geosystems* 5, Q03G13, doi: 10.1029/2002GC000434.
- Rogers, N. W., Thomas, L. E., Macdonald, R., Hawkesworth, C. J. & Mokadem, F. (2006). ²³⁸U-²³⁰Th disequilibrium in recent basalts and dynamic melting beneath the Kenya rift. *Chemical Geology* **234**, 148–168.
- Salters, V. J. M., Longhi, J. E. & Bizimis, M. (2002). Near mantle solidus trace element partitioning at pressures up to 3–4 GPa. *Geochemistry, Geophysics, Geosystems* 3, doi: 10.1029/2001GC000148.
- Sarafian, E., Gaetani, G. A., Hauri, E. H. & Sarafian, A. R. (2017). Experimental constraints on the damp peridotite solidus and oceanic mantle potential temperature. *Science* 355, 942–945.
- Skulski, T., Minarik, W. & Watson, E. B. (1994). High-pressure experimental trace-element partitioning between clinopyroxene and basaltic melts. *Chemical Geology* 117, 127–147.
- Sobolev, A. V., Hofmann, A. W., Sobolev, S. V. & Nikogosian, I. K. (2005). An olivine-free mantle source of Hawaiian shield basalts. *Nature* 434, 590–597.
- Sun, S.-S. & McDonough, W. F. (1989). Chemical and isotopic systematics of oceanic basalts: implications for mantle composition and processes. In: Saunders, A. D. & Norry, M. J. (eds) Magmatism in the Ocean Basins. Geological Society, London, Special Publications 42, 313–345.
- Sutherland, R. & Hollis, C. (2001). Cretaceous demise of the Moa plate and strike-slip motion at the Gondwana margin. *Geology* **29**, 279–282.
- Tarduno, J. A., Duncan, R. A. & Scholl, D. W. (2003). The Emperor Seamounts: southward motion of the Hawaiian hotspot plume in Earth's mantle. Science 301, 1064–1069.
- Tejada, M. L. G., Mahoney, J. J., Castillo, P. R., Ingle, S. P., Sheth, H. C. & Weis, D. (2004). Pin-pricking the elephant: evidence on the origin of the Ontong Java Plateau from Pb-Sr-Hf-Nd isotopic characteristics of ODP leg 192 basalts. In: Fitton, J. G., Mahoney, J. J., Wallace, P. J. & Saunders, A. D. (eds) Origin and Evolution of the Ontong Java Plateau. Geological Society, London, Special Publications 229,133-150.
- Tejada, M. L. G., Hanyu, T., Ishikawa, A., Senda, R., Suzuki, K., Fitton, G. & Williams, R. (2015). Re–Os isotope and platinum group elements of a FOcal ZOne mantle source, Louisville Seamounts Chain, Pacific Ocean. Geochemistry, Geophysics, Geosystems 16, 486–504.doi:10.1002/2014GC0 05629.
- Tuff, J. & Gibson, S. (2007). Trace-element partitioning between garnet, clinopyroxene and Fe-rich picritic melts at 3 to 7 GPa. Contributions to Mineralogy and Petrology 153, 369–387.
- Turcotte, D. L. & Schubert, G. (1982). Geodynamics: Applications of Continuum Physics to Geological Problems. New York: Wiley, 450 pp.

- Vanderkluysen, L., Mahoney, J. J., Koppers, A. A. P., Beier, C., Regelous, M., Gee, J. S. & Lonsdale, P. F. (2014). Louisville Seamount Chain: Petrogenetic processes and geochemical evolution of the mantle source. *Geochemistry, Geophysics, Geosystems* 15, 2380–2400.doi:10.1002/ 2014GC005288.
- Walter, M. J. (1998). Melting of garnet peridotite and the origin of komatiite and depleted lithosphere. *Journal of Petrology* **39**, 29–60.
- Watson, S. & McKenzie, D. (1991). Melt generation by plumes: a study of Hawaiian volcanism. *Journal of Petrology* **32**, 501–537.
- Watts, A. B., Weissel, J. K., Duncan, R. A. & Larson, R. L. (1988). Origin of the Louisville Ridge and its relationship to the Eltanin Fracture Zone System. *Journal of Geophysical Research* **93**, 3051.
- Wessel, P. (2016). Regional–residual separation of bathymetry and revised estimates of Hawaii plume flux. *Geophysical Journal International* **204**, 932–947.
- Worthington, T. J., Hekinian, R., Stoffers, P., Kuhn, T. & Hauff, F. (2006). Osbourn Trough: Structure, geochemistry and implications of a mid-Cretaceous paleospreading ridge in the South Pacific. *Earth and Planetary Science Letters* **245**, 685–701.