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1 Global oceanic anoxia linked with the Capitanian (Middle Permian) marine mass

2 extinction

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27 ABSTRACT

28 The timing and causation of the Capitanian (late Middle Permian) biocrisis remain controversial. Here, a detailed uranium-isotopic (δ^{238} U) profile was generated for the 29 30 mid-Capitanian to lower Wuchiapingian of the Penglaitan section (the 31 Guadalupian/Lopingian Permian global stratotype) in South China for the purpose of 32 investigating relationships between the biocrisis and coeval oceanic anoxic events (OAEs). Negative δ^{238} U excursions indicate two distinct OAEs, a mid-Capitanian 33 (OAE-C1) and an end-Capitanian (OAE-C2) event. Mass balance modeling shows that 34 35 the anoxic sink of uranium (F_{anox} ; i.e., the fraction of the total U burial flux) and anoxic 36 seafloor area (Farea; i.e., the fraction of total seafloor area) increased during each OAE. 37 A dynamic mass balance model yields increases of F_{anox} from <30 % to >60 % and F_{area} from ~1% to ~4-7% during each OAE. These two OAEs coincided with two extinction 38 episodes during the Capitanian biocrisis, supporting a causal relationship between 39 oceanic anoxia and mass extinction during the Middle Permian. The most likely driver 40

41 of middle to late Capitanian global warming and oceanic anoxia was episodic
42 magmatism of the Emeishan Large Igneous Province.

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Keywords: Guadalupian–Lopingian boundary; biocrisis; uranium isotopes; Emeishan
LIP; global warming; Penglaitan

46

47 **1. Introduction**

48 The Capitanian mass extinction, characterized by genus-level extinction rates of 49 42-61% (Stanley and Yang, 1994; Clapham et al., 2009; Stanley, 2016), was 50 particularly severe for shallow-marine taxa including fusulinacean foraminifers, corals, 51 bivalves, and ammonoids (Shen and Shi, 2009; Wignall et al., 2009; Bond et al., 2010). 52 This event also affected terrestrial ecosystems and resulted in a mass extinction of land vertebrates (Retallack et al., 2006). The timing and number of episodes of the 53 54 Capitanian biocrisis remain controversial, with proposals for either a mid-Capitanian 55 (Wignall et al., 2009; Bond et al., 2010) or an end-Capitanian event (Shen and Shi, 56 2009; Shen et al., 2020). Although a plethora of mechanisms have been proposed for 57 this extinction, including marine anoxia, volcanism, sea-level fall, ocean acidification, 58 and climatic change (Zhou et al., 2002; Isozaki et al., 2007; Wignall et al., 2009; Bond 59 et al., 2010, 2015, 2020; Zhang et al., 2015; Wei et al., 2016), no consensus regarding 60 its causation has been reached to date.

61	Uranium isotopes in marine carbonates ($\delta^{238}U_{carb}$) permit quantitative
62	reconstruction of secular variation in average global-ocean redox conditions. The basis
63	for this proxy is the long residence time of U in seawater (~400 kyr) relative to the
64	oceanic mixing time (~1-2 kyr for the modern) (Ku et al., 1977). Microbially mediated
65	reduction of U(VI) to U(IV) under anoxic conditions at the seafloor results in a
66	substantial decrease in U solubility in seawater (Weyer et al., 2008; Andersen et al.,
67	2014). Because ²³⁸ U is preferentially reduced and immobilized relative to ²³⁵ U, the
68	δ^{238} U of seawater decreases as its burial flux increases. Thus, a global increase in the
69	extent of ocean anoxic seafloor is expected to cause simultaneous decreases in seawater
70	U concentration [U] and $\delta^{238} U.$ Primary marine carbonates record the $\delta^{238} U$ of
71	contemporaneous seawater with little fractionation, although precipitation of secondary
72	carbonate cement in the sulfate reduction zone can lead to isotopic shifts of $+0.2$ ‰ to
73	+0.4 ‰ in bulk-rock δ^{238} U (Chen et al., 2018; Tissot et al., 2018). Carbonate δ^{238} U
74	records have been used to quantitatively reconstruct the extent of seafloor anoxia during
75	several oceanic anoxic events (Lau et al., 2016; Song et al., 2017; Bartlett et al., 2018;
76	Zhang et al., 2018).

Although several studies have used elemental or mineralogical proxies to examine local environmental redox variation in Middle Permian sections, with possible implications for ocean anoxic events during the mid-Capitanian (Bond et al., 2015, 2020) and end-Capitanian (Zhang et al., 2015; Wei et al., 2016), global-ocean redox proxies such as carbonate δ^{238} U have not yet been generated for this event interval.

82	Here, we use the carbonate δ^{238} U proxy to document two discrete episodes of global-
83	ocean anoxia during the Capitanian and demonstrate their synchronicity with the two
84	phases of the Middle Permian mass extinction identified in earlier studies.
85	
86	2. Geological background
87	The Penglaitan section (23°41'43"N, 109°19'16"E; Fig. 1) is the Global Stratotype
88	Section and Point (GSSP) for the Guadalupian-Lopingian series boundary (GLB) as
89	well as of the Capitanian-Wuchiapingian stage boundary (Jin et al., 2006). The GLB is
90	defined by the first appearance datum of the conodont Clarkina postbitteri postbitteri
91	at the base of Bed 6k at Penglaitan (Jin et al., 2006). The presence of all biozones from

92 the *Jingondolella shannoni* Zone to the *Clarkina dukouensis* Zone indicates an absence

93 of major hiatuses at Penglaitan (Fig. 2A) (Jost et al., 2014). A total of 53 limestone

94 samples was collected over 22 m of section exposed along the Hongshui River,
95 beginning in the *Jingondolella shannoni* Zone, extending through the upper part of
96 Maokou Formation, and terminating at the base of *Clarkina dukouensis* Zone in the

97 lower Heshan Formation.

98

99 **3. Methods**

Weathered surfaces and veins of the samples were trimmed off, and the remaining
sample material was cut into small pieces and ground to a fine powder using a ball mill
at Wuhan Institute of Geology and Mineral Resources.

104 *3.1. Carbon and oxygen isotope analyses*

Carbon isotopic compositions were measured using a Thermo Fisher Gasbench II-105 106 MAT 253 stable isotope mass spectrometer at the State Key Laboratory of Biogeology 107 and Environmental Geology in the China University of Geosciences (Wuhan). About 108 150-400 µg of powder was placed in a 10 mL Na-glass vial, sealed with a butyl rubber septum, and reacted with 100 % phosphoric acid at 72 °C after flushing with helium 109 using the Gasbench II interface. The $\delta^{13}C_{carb}$ composition of the evolved CO₂ gas was 110 111 measured and reported as per mille variation relative to the Vienna Pee Dee Belemnite 112 (VPDB) standard. Data quality was monitored via repeated analysis of two Chinese national standards, GBW 04416 ($\delta^{13}C = +1.61 \%$, $\delta^{18}O = -11.59 \%$) and GBW 04417 113 $(\delta^{13}C = -6.06 \%, \delta^{18}O = -24.12 \%)$, which yielded analytical precisions (2 σ) of better 114 than ± 0.1 % for δ^{13} C and δ^{18} O. In addition, one sample out of every ten was re-analyzed 115 116 as a replicate to monitor instrumental precision.

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118 *3.2. Elemental and uranium isotope analyses*

Elemental concentrations and uranium isotopes were measured at Arizona State University. About 3-4 g of sample powder were dissolved in 1 M HNO₃ for 24 h and centrifuged to separate insoluble residue from the acid-soluble fraction. The trace element concentrations of carbonates in solution were analyzed using a Thermo iCAP-Q inductively coupled plasma mass spectrometer (ICP-MS). The solution, containing

124	${\sim}500$ ng U, was then spiked with an IRMM 3636 uranium double spike to give a
125	233 U/ 235 U ratio of 2.5 and dried down to homogenize the spike-sample mixture. The
126	sample was redissolved in 3 M HNO3, and U was separated from sample matrix using
127	UTEVA Resin (http://www.eichrom.com/products/info/uteva_resin.aspx; see Weyer et
128	al., 2008), with extension of the 3 M HNO3 matrix elution step to 12 mL in order to
129	completely remove calcium. After column chemistry, the sample was treated twice with
130	a mixture of 2 mL concentrated HNO3 and 0.2 mL 30 $\%$ H2O2 to remove residual
131	organics from the resin, and it was then dissolved in 2 % HNO3 for isotopic analysis.
132	Uranium isotope measurements were performed on a Thermo Scientific Neptune
133	multicollector-inductively coupled plasma mass spectrometer (MC-ICP-MS) and
134	reported as $\delta^{238}U$ ($^{238}U/^{235}U)$ relative to the U-isotope standard CRM-145. This
135	standard was analyzed before and after every two samples to correct for minor
136	instrumental drift. The reproducibility of δ^{238} U for the standard CRM-145 was better
137	than \pm 0.11 ‰ (2 SD, N = 30). The accuracy of U isotope measurements was confirmed
138	by multiple analyses of the secondary standard CRM-129a, which yielded -1.70 ± 0.08 ‰
139	(2 SD, $N = 16$), conforming to values reported in previous studies (Chen et al., 2018,
140	2021; Zhang et al., 2020). The standard BCR-2 (Columbia River Basalt) yielded values
141	of $-0.24 \pm 0.12\%$ (2 SD, N = 14), in agreement with Lau et al. (2016).
142	

143 **4. Results**

The $\delta^{13}C_{carb}$ profile for Penglaitan is similar to that generated in a previous study 144 (Jost et al., 2014). $\delta^{13}C_{carb}$ fluctuates between +3.0 ‰ and +5.5 ‰ with moderate 145 146 positive excursions peaking at ~9.1 m (CPE1; CPE: carbon isotopic positive excursion) and ~ 20.0 m (CPE2). The GLB is marked by a distinct negative excursion to +3.2 ‰ 147 (Fig. 2A; Table S1). Carbonate δ^{238} U at Penglaitan ranges from -1.25 ‰ to +0.13 ‰ 148 149 with two distinct negative excursions: UNE-1 (UNE: uranium isotopic negative 150 excursion), from -0.20 ‰ at 3.1 m to -1.25 ‰ at 8.6 m, and UNE-2 from -0.20 ‰ at 151 19.7 m to -0.65 ‰ at 21.6 m, the latter consisting of two peaks (UNE2a at 20.7 m and 152 UNE2b at 21.6 m (Fig. 2B; Table S1). For each episode (i.e., mid-Capitanian versus end-Capitanian), the intervals of the $\delta^{13}C_{carb}$ and $\delta^{238}U$ excursions overlap broadly, 153 154 although their peaks are slightly offset: UNE1 precedes CPE1 by ~0.5 m but UNE2 lags CPE2 by ~1.5 m. However, second-order excursions in the $\delta^{13}C_{carb}$ and $\delta^{238}U$ 155 profiles at ~12-13 m and ~16-17 m are well-aligned (Fig. 2A-B), documenting a 156 pervasive coupling of the carbon and uranium cycles within the study section. 157

158

159 **5. Discussion**

160 5.1. Evidence for primary seawater $\delta^{238}U$ values

Previous studies have demonstrated that carbonate U isotopes can be altered by
various processes (e.g., marine diagenesis, meteoric diagenesis, and dolomitization),
yielding mixed primary and secondary signatures in pelagic carbonates (Romaniello et

al., 2013; Andersen et al., 2014; Chen et al., 2018; see other references review in Zhang et al., 2020). Here, we investigated relationships between δ^{238} U and the U concentration, CaCO₃ content, and Mg/Ca, Mn/Sr, U/Sr and U/Al ratios of the study samples to evaluate the robustness of our U-isotope dataset as a primary record of seawater U isotopes.

169 Post-depositional solution-precipitation reactions in the host carbonate sediment 170 can leach U from calcite, altering Th/U ratios (Chung and Swart, 1990). However, U leaching generally affects U concentrations without altering δ^{238} U values (Weyer et al., 171 172 2008) due to a lack of fractionation among U isotopes during weathering. Detrital U can mask carbonate U-isotope signals (Stirling et al., 2007; Asael et al., 2013), but there 173 is no significant relationship between U concentration and δ^{238} U (Fig. 3A). The study 174 175 samples also do not show a significant relationship between CaCO₃ content and δ^{238} U (Fig. 3B). Dolomitization can alter δ^{238} U (Romaniello et al., 2013), but most samples 176 177 in this study (50 of 53 total) exhibit Mg/Ca ratios <0.1, indicating that dolomitization was insignificant, and there is no significant correlation of Mg/Ca ratios with δ^{238} U (Fig. 178 179 3C).

180 Th is delivered to marine sediments in detrital siliciclastics (mainly clays), and it 181 is relatively immobile in the diagenetic environment. Uranium can accumulate in 182 marine sediments in several forms, including detrital, carbonate-bound, phosphate-183 bound, and organic-bound fractions (Cumberland et al., 2016). Given typical 184 concentrations in upper continental crust-derived sediments (McLennan et al., 2001)

185	(10.7 ppm Th, 2.8 ppm U), Th/U ratios should be \sim 4 in sediments without authigenic
186	U enrichment. Wignall and Twitchett (1996) cited Th/U ratios of 2 to 7 for oxic facies
187	versus Th/U ratios of ≤ 2 for anoxic facies in which measurable authigenic U enrichment
188	has occurred. The degree of authigenic U enrichment depends on seawater U
189	concentrations, however, and if widespread seafloor anoxia results in seawater U
190	drawdown, then the authigenic U fraction of the sediment will also decline (cf. Algeo,
191	2004). Following initial deposition, U can adsorb/desorb from some phases (especially
192	organics) due to redox changes in sediment porewaters (Cumberland et al., 2016). U in
193	carbonate and phosphate is structurally bound and less subject to secondary
194	remobilization. Diagenetic alteration of U-isotope signals is potentially a concern in
195	carbonate sediments (Hood et al., 2018), but Mn/Sr, U/Sr and U/Al ratios show no
196	correlation with δ^{238} U (Fig. 3D-F). In addition, no significant correlation of δ^{238} U and
197	U concentrations exists with Al/Ca and Th/Ca (see S1. Supplemental assessment of
198	diagenetic effects in Supplementary material). These observations provide no evidence
199	of diagenetic alteration of the U-isotopic compositions of the study samples and are
200	thus consistent with well-preserved primary marine geochemical signatures in the
201	Penglaitan section. Leaching of detrital U due to use of 1 N HNO3 is possible, but
202	testing of different carbonate dissolution protocols (i.e., using acetic acid, HCl and
203	HNO ₃) did not yield any significant differences in $\delta^{238}U_{carb}$ (see S2. Tests of multiple
204	dissolution protocols in Supplementary material).

205 Chen et al. (2021) inferred that the uranium isotope compositions of calcite can be 206 affected by anoxic seawater conditions. However, aragonite is relatively less 207 susceptible to this influence, and Permian seas favored aragonite precipitation, as 208 shown by fluid inclusions in marine halite (Lowenstein et al., 2005), although the original carbonate mineralogy of the present study samples is uncertain. In addition, 209 there is no correlation between our δ^{238} U profile and local redox variation, as 210 determined from framboidal pyrite (Wei et al., 2016). In summary, there are no 211 212 relationships of δ^{238} U to other geochemical proxies that would imply alteration of U-213 system chemistry by siliciclastic inputs, dolomitization, or other diagenetic processes. Hence, the δ^{238} U profile of this study is regarded as a robust record of variations in 214 215 contemporaneous seawater U-isotopic composition.

216

217 5.2. Steady-state and dynamic modeling of the U cycles

218 Seawater U concentration ([U]) and isotopic compositions are controlled by several factors including the [U] and δ^{238} U of rivers and the sink flux of U to anoxic facies. Lau 219 220 et al. (2016) demonstrated that the principal mechanism for producing large, rapid, and sustained decreases in [U] and $\delta^{238}U$ (such as those observed in the present study 221 222 section) is an enhanced flux of U to anoxic facies (i.e., higher F_{anox}). In this study, we 223 employed both dynamic and steady-state U cycle models to quantitatively estimate F_{anox} and concurrent changes in seawater [U] and F_{area} (see S3. Marine uranium cycle 224 modeling in Supplementary material). The dynamic model results show that F_{anox} 225

226	increased from $<30\%$ to $\sim70\%$ during UNE-1, decreased to 0-30% between UNE-1 and
227	UNE-2, and then increased again to ~70% during UNE-2 (Fig. 4B; Table S2; note: F_{anox}
228	is 10 in the modern ocean, Montoya-Pino et al., 2010). Seawater [U] decreased from
229	11 nM to 5 nM during UNE-1 and from 13 nM to 8 nM during UNE-2 (Fig. 4A). Mass
230	balance calculations yield F_{area} estimates of ~7% during UNE-1 and ~4% during UNE-
231	2 (Fig. 4D; Table S2; this value is 0.2% in the modern ocean and \sim 1% in the Middle
232	Permian prior to OAE-C1). The steady-state model yields similar trends for F_{anox} and
233	F_{area} but relatively higher values than the dynamic model (Fig. 4); sensitivity tests
234	demonstrate the general robustness of the modeling output (see S4. Sensitivity testing
235	of model simulations in Supplementary material).

237 5.3. Two phases of oceanic anoxia during Capitanian

The Penglaitan U-isotope profile and U cycle model reveal two phases of 238 239 widespread global-ocean anoxia during the middle and late Capitanian (Fig. 2), 240 consistent with the findings of earlier studies making use of local redox proxies (Bond 241 et al., 2015, 2020; Zhang et al., 2015; Wei et al., 2016; Zhang et al., 2021). Pyrite 242 framboid size distributions, sulfur isotopes, and redox-sensitive elements in South 243 China, Spitsbergen, and Arctic Canada provided evidence of a mid-Capitanian anoxic 244 event, although its precise age in the Boreal region is uncertain due to poor age control (Bond et al., 2015, 2020; Zhang et al., 2015; Wei et al., 2016; Zhang et al., 2021). Latest 245 246 Capitanian anoxia has been inferred on the basis of pyrite framboids and multiple sulfur

isotopes in South China and the western USA (Zhang et al., 2015; Wei et al., 2016;
Bond et al., 2020). Our U-isotope record, combined with these local redox proxy data,
serves to document the existence of two discrete Capitanian OAEs, a finding not
previously demonstrated.

251 The specific mechanism triggering oceanic anoxia during the Capitanian remains 252 under investigation. Recent studies have constrained the timing of onset and 253 termination of Emeishan Large Igneous Province (ELIP) magmatism and demonstrated 254 the existence of several eruption phases (reviewed by Shellnutt et al., 2020). ELIP 255 eruptions began during the J. altudaensis Zone (~263 Ma) and increased in volume 256 during the J. xuanhanensis Zone (~262 Ma) (Sun et al., 2010). New U-Pb ages from zircons in claystones support extensive submarine eruptions during the middle 257 258 Capitanian (~262.5 Ma) (Yan et al., 2020), with termination of the main phase of ELIP activity was at ~260.0 \pm 0.9 Ma (Zhong et al., 2014). Coeval ELIP effects have been 259 260 inferred from spikes in Hg/TOC profiles of the Middle Permian in Spitsbergen, South 261 China, Arctic Canada (Grasby et al., 2016; Huang et al., 2019; Bond et al., 2020) and a 262 positive Hg isotope excursion at Penglaitan (Huang et al., 2019). Recently, Liu et al 263 (2021) used surface-wave tomography to document the existence of a huge magma 264 reservoir and hidden hotspot track related to the ELIP, noting that such hotspot 265 volcanism can liberate potentially catastrophic volumes of greenhouse gases (Fig. 2). 266 Our results demonstrate that the Penglaitan section records two discrete OAEs 267 dating to the middle (J. altudaensis to lower J. xuanhanensis zones) and the late

268	Capitanian (upper J. granti to lower C. dukouensis zones). Each OAE may have
269	coincided with a discrete ELIP eruption phase that triggered global climatic warming
270	of ~3-5 $^\circ\!\!\mathbb{C}$ (Chen et al., 2011; Wang et al., 2020; Zhang et al., 2021) (Fig. 2D), and
271	associated development of marine anoxia through: (1) reduced overturning circulation
272	and increased water-column stratification (Zhang et al., 2021); (2) enhanced subaerial
273	weathering and riverine nutrient fluxes, stimulating high marine primary productivity
274	that consumed dissolved oxygen; and (3) decreased oxygen solubility in seawater. A
275	lessening of oceanic anoxia between UNE1 and UNE2 implies a relative lull in ELIP
276	activity between the two main eruption stages, during which lower and more stable
277	seawater temperatures existed (Chen et al., 2011; Wang et al., 2020) (Fig. 2D).
278	Patterns of $\delta^{13}C_{carb}$ variation during the Capitanian and their underlying
278 279	Patterns of $\delta^{13}C_{carb}$ variation during the Capitanian and their underlying significance remain in debate (Bond et al., 2010; Jost et al., 2014). In this study, CPE1
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279 280	significance remain in debate (Bond et al., 2010; Jost et al., 2014). In this study, CPE1 (~1.5 ‰) spans the upper <i>J. altudaensis</i> to basal <i>J. xuanhanensis</i> zones, and CPE2
279 280 281	significance remain in debate (Bond et al., 2010; Jost et al., 2014). In this study, CPE1 (~1.5 ‰) spans the upper <i>J. altudaensis</i> to basal <i>J. xuanhanensis</i> zones, and CPE2 (~1.0 ‰) spans the upper <i>J. granti</i> to the top of <i>C. postbitteri hongshuiensis</i> zones.
279280281282	significance remain in debate (Bond et al., 2010; Jost et al., 2014). In this study, CPE1 (~1.5 ‰) spans the upper <i>J. altudaensis</i> to basal <i>J. xuanhanensis</i> zones, and CPE2 (~1.0 ‰) spans the upper <i>J. granti</i> to the top of <i>C. postbitteri hongshuiensis</i> zones. Whereas CPE2 is well-documented in South China and Japanese sections (Isozaki et
 279 280 281 282 283 	significance remain in debate (Bond et al., 2010; Jost et al., 2014). In this study, CPE1 (~1.5 ‰) spans the upper <i>J. altudaensis</i> to basal <i>J. xuanhanensis</i> zones, and CPE2 (~1.0 ‰) spans the upper <i>J. granti</i> to the top of <i>C. postbitteri hongshuiensis</i> zones. Whereas CPE2 is well-documented in South China and Japanese sections (Isozaki et al., 2007; Jost et al., 2014; Chen et al., 2011; Wang et al., 2004), CPE1 is less well-
 279 280 281 282 283 284 	significance remain in debate (Bond et al., 2010; Jost et al., 2014). In this study, CPE1 (~1.5 ‰) spans the upper <i>J. altudaensis</i> to basal <i>J. xuanhanensis</i> zones, and CPE2 (~1.0 ‰) spans the upper <i>J. granti</i> to the top of <i>C. postbitteri hongshuiensis</i> zones. Whereas CPE2 is well-documented in South China and Japanese sections (Isozaki et al., 2007; Jost et al., 2014; Chen et al., 2011; Wang et al., 2004), CPE1 is less well-established, possibly being represented by a small positive excursion in the <i>J.</i>

large negative excursion (Wignall et al., 2009), but this feature has been attributed to
local diagenesis (Jost et al., 2014).

290 Systematic negative covariation between the C- and U-isotope profiles records 291 linkage between organic carbon burial fluxes and global-ocean redox conditions, and the relative timing of δ^{13} C versus δ^{238} U shifts may indicate patterns of forcings (Fig. 292 293 2A-B). The UNE1 excursion leads CPE1 with respect to both its start (at ~3.6 m vs \sim 5.8 m) and its peak (at 8.6 m vs 9.1 m), which is consistent with expanded oceanic 294 295 anoxia triggering increased organic burial. The stratigraphic interval between the two main isotopic events is marked by small negative δ^{238} U excursions at ~13 m and ~17 296 297 m ("minor UNEs"; Fig. 2) that correspond to the termini of rising segments of the δ^{13} C profile. The observation that the δ^{13} C profile is trending positive throughout this 298 interval while negative δ^{238} U excursions are short, discrete events suggests that rising 299 300 marine productivity and organic burial were now driving anoxia, and not the other way 301 around as during the OAE-C1 event. The onset of UNE2 at 18.6 m leads that of CPE2 302 at 19.6 m, suggesting that, once again, oceanic redox changes drove organic carbon 303 burial. This pattern (i.e., redox changes leading organic burial) is consistent with an LIP 304 driver, as massive ocean-surface warming due to emissions of greenhouse gases would 305 cause water-column stratification and stagnation, producing deepwater anoxia that 306 enhanced organic carbon burial (Zhang et al., 2021; cf. Song et al., 2013). Thus, the interplay of our C- and U-isotope records supports two main eruption stages of the ELIP, 307

308 the first in the mid-Guadalupian (*J. prexuanhanensis* to early *J. xuanhanensis* zones)
309 and the second at the GLB (Fig. 2).

The OAE-C2 event was terminated by a rapid decline in δ^{13} C, whose onset at 20.2 310 m preceded the start of the positive shift in δ^{238} U at 21.6 m. This relationship may 311 312 indicate that, despite continued water-column stagnation related to LIP greenhouse gas 313 release (Zhang et al., 2021; cf. Song et al., 2013), organic C burial fluxes began to falter, probably because of a general deterioration of marine ecosystems and reduced primary 314 production around the GLB. Although other interpretations of the $\delta^{13}C-\delta^{238}U$ 315 316 relationships may be possible, the scenario above is consistent with available data 317 regarding ELIP activity and marine biotic patterns during the GLB transition.

318

319 5.4. The link between oceanic anoxia and biotic crisis

320 The cause of the biotic crisis during the Capitanian has not been resolved in part 321 because the number and timing of extinctions and their relationships to coeval marine 322 environmental changes are not well established. Our U-isotope records document two 323 OAEs, each coincident with a discrete episode of the biocrisis (Shen and Shi, 2009; 324 Wignall et al., 2009; Bond et al., 2010; Shen et al., 2020) (Fig. 5). During the first 325 episode (i.e., mid-Capitanian OAE-C1), various calcareous algae, fusulinids, and 326 marine invertebrates disappeared in South China (Wignall et al., 2009; Yang et al., 2004) (Fig. 5), Spitsbergen and Arctic Canada (Bond et al., 2015, 2020). At Penglaitan, this 327 328 event was followed by an interval in which reefs flourished (Huang et al., 2019). During

329	the second episode (i.e., end-Capitanian OAE-C2), reef organisms (including corals,
330	fusulinids, and alataconchids) disappeared together with many ammonoid taxa (Wang
331	and Sugiyama, 2001; Yang et al., 2004; Jin et al., 2006) (Fig. 5). The lesser severity of
332	the Capitanian biocrisis relative to the 'Big Five' Phanerozoic mass extinctions
333	(Clapham et al., 2009; Shen et al., 2020) is consistent with its relatively small areas of
334	seafloor anoxia (~4-7%) compared to the end-Permian (>20%) (Zhang et al., 2018) and
335	Late Ordovician mass extinctions (~15%) (Bartlett et al., 2018). In summary, the
336	Capitanian crisis occurred in two episodes, separated by an interlude of improved
337	environmental conditions. Each episode was linked to an oceanic anoxic event that may
338	have been triggered by a phase of intense ELIP magmatism.

340 **6.** Conclusions

341 We present a new high-resolution carbonate U-isotope record from the middle Capitanian to earliest Lopingian from the Penglaitan section of South China. The δ^{238} U 342 343 profile and marine uranium cycle modeling based thereon reveal two separate global 344 oceanic anoxic events (OAEs), the first during the middle Capitanian J. altudaensis to lower J. xuanhanensis zones, and the second during the latest Capitanian upper J. granti 345 to lower C. dukouensis zones. Each OAE coincided with one phase of the Capitanian 346 marine biotic crisis, suggesting a direct causal relationship between anoxia and mass 347 extinction during the Middle Permian. Our results are consistent with eruptions of the 348

349 Emeishan Large Igneous Province as the trigger for contemporaneous oceanic and350 climatic changes.

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352 Acknowledgments

We thank Professor Shen Shuzhong for discussion of an early version of this manuscript. We thank Du Yong for technical assistance with the carbon and oxygen isotope analyses. This work was supported by the National Natural Science Foundation of China (grants 42172032, 41872033, 41402302 to H-Y.S.) and the Natural Environment Research Council (NE/J01799X/1 to D.P.G.B.).

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359 Author contributions: H-Y.S., A.D.A. designed this research; H-Y.S., H-J.S., H.W.
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- 360 collected the samples; H-Y.S., W.Z., T.J.A., and X.C. performed the elemental and
- 361 isotopes analysis. R.J.S., T.J.A., W.Z. and H-Y.S. performed the uranium modeling. H-
- 362 Y.S., H-J.S., P.B.W., D.P.G.B. and J.T. evaluated mass extinction scenarios. H-Y.S.

363 prepared the manuscript with contributions from all co-authors.

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Fig. 1. Global paleogeography of Middle Permian (~260 Ma). Adapted from Ron
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is the Penglaitan section. Red circles are sites of oceanic anoxia area during the
Guadalupian. A: South China (Zhang et al., 2015; Wei et al., 2016); B: Spitsbergen
(Bond et al., 2020); C: Sverdrup Basin (Bond et al., 2015); D: Nevada (Zhang et al.,
2015).

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Fig. 2. Chemostratigraphy of the Penglaitan study section: (A) $\delta^{13}C_{carb}$, (B) 549 carbonate δ^{238} U; (C) percent anoxic seafloor area (F_{area}); (D) Tropical sea surface 550 551 temperature (SST) curves. Abbreviations: Fm.: Formation; J.: Jingondolella; alt.: 552 altudaensis; pre.: prexuanhanensis; C.p.h.: Clarkina postbitteri hongshuiensis; C.p.p.: Clarkina postbitteri postbitteri; C.d.: Clarkina dukouensis. CPE: carbon isotope 553 positive excursion; UNE: uranium isotope negative excursion, ELIP: Emeishan Large 554 555 Igneous Province. The geochronologic ages are from Henderson et al. (2012) and Shen 556 et al. (2020). Tropical SST curves are from Chen et al. (2011) and Wang et al. (2020); 557 Hg peak intervals in mid-Capitanian from Grasby et al. (2016) and Bond et al. (2020) and in end-Capitanian from Huang et al. (2019); hotspot volcanism from Liu et al. 558 559 (2021); ELIP interval from Shellnutt et al. (2020).

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- 561 Fig. 3. Comparison of δ²³⁸U ratios to U concentration (A), CaCO₃ (B), Mg/Ca ratios
 562 (C), Mn/Sr ratios (D), U/Sr (E), and U/Al (F).
- 563

Fig. 4. Marine U-cycle mass balance model under steady state and dynamic state: 564 (A) Seawater U concentrations (nM); (B) Fractional U removal to anoxic sink, Fanox; 565 (C) Seawater δ^{238} U data (gray), LOWESS-smoothed curve (black, overlapped by the 566 dynamic model curve), and dynamic model curve (blue); (D) Modeled changes in 567 568 anoxic seafloor changes, Farea. 569 570 Fig. 5. Relationship between anoxic seafloor (F_{anoxic}) and two stages of mass extinction during the Capitanian. Taxon ranges for mid-Capitanian are from Wignall 571 572 et al. (2009) and for end-Capitanian from Jin et al. (2006) and Wang and Sugiyama (2001). 573

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